



DESIGN EXPLORATION AND TECHNOLOGY DEVELOPMENT OF THE STEP LI2O CERAMIC BREEDER BLANKET

**CHRIS HARRINGTON, ADITYA PIDAPARTHY,
& THE STEP BLANKETS TEAM**

**OUTBOARD FIRST WALL AND BREEDER BLANKET
SYSTEM LEAD**

*IAEA Technical Meeting on Tritium Breeding Blankets and
Associated Neutronics*

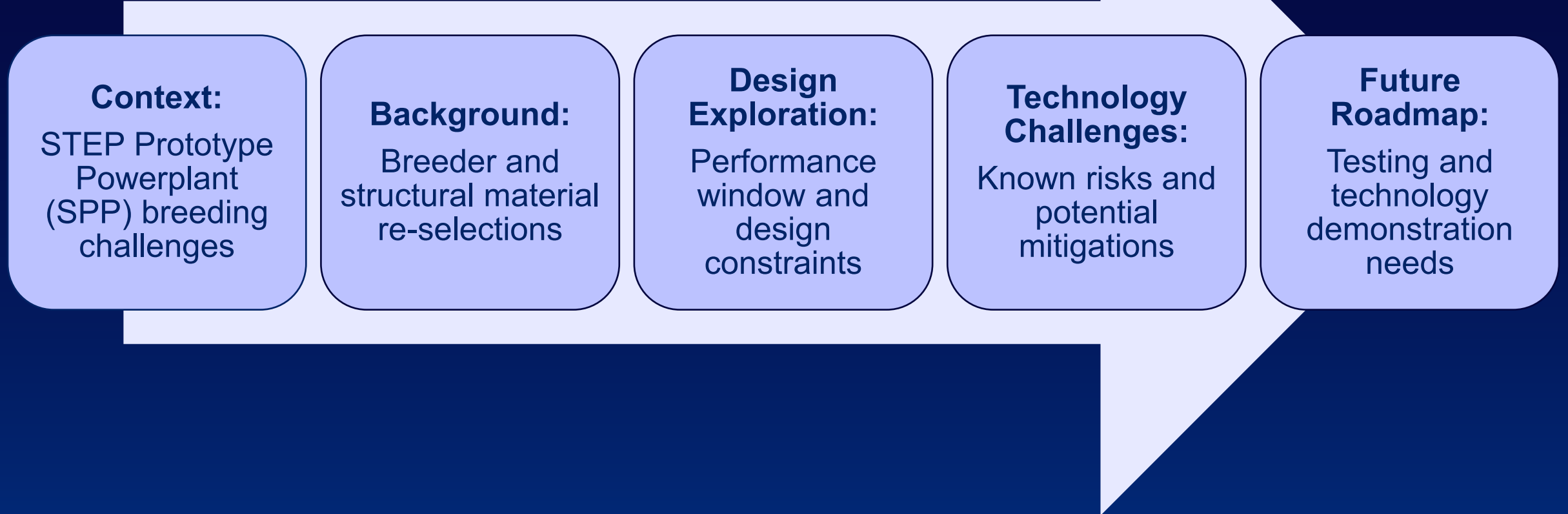
IAEA Headquarters, Vienna, Austria, 2nd – 5th September 2025



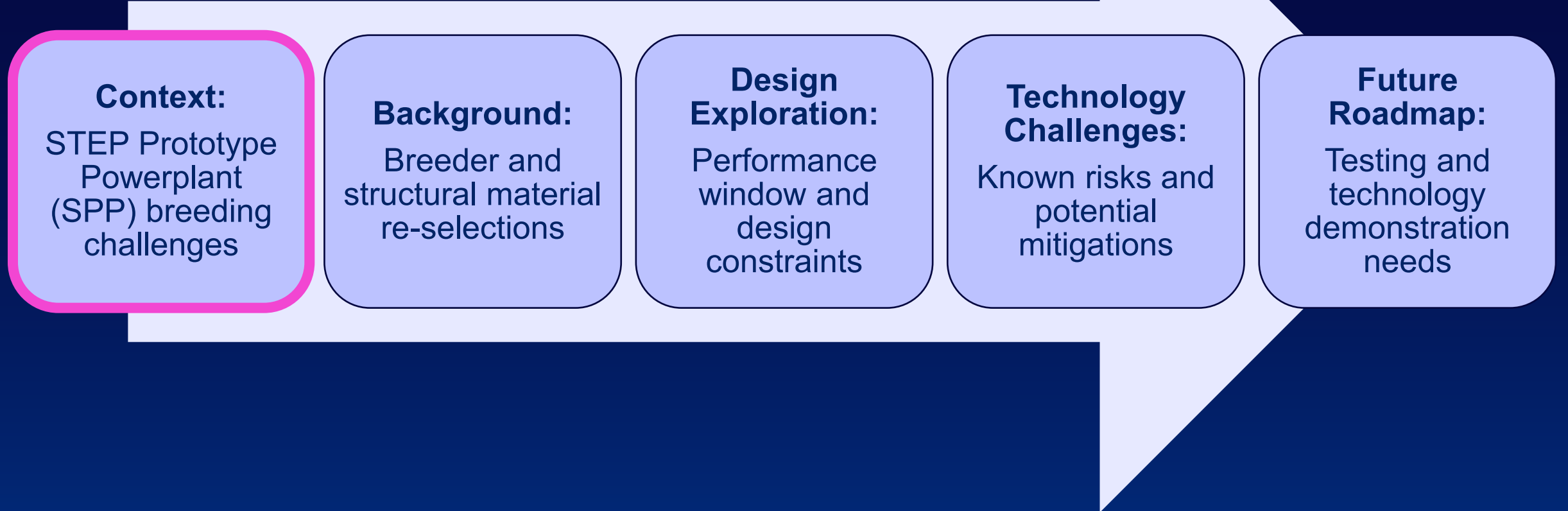
***"DELIVER A UK PROTOTYPE
FUSION ENERGY PLANT,
TARGETING 2040, AND A
PATH TO COMMERCIAL
VIABILITY OF FUSION"***

STEP MISSION

OUTLINE



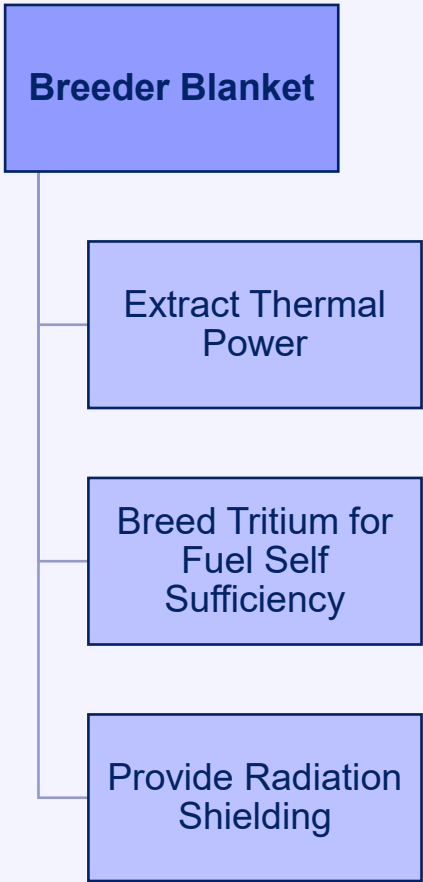
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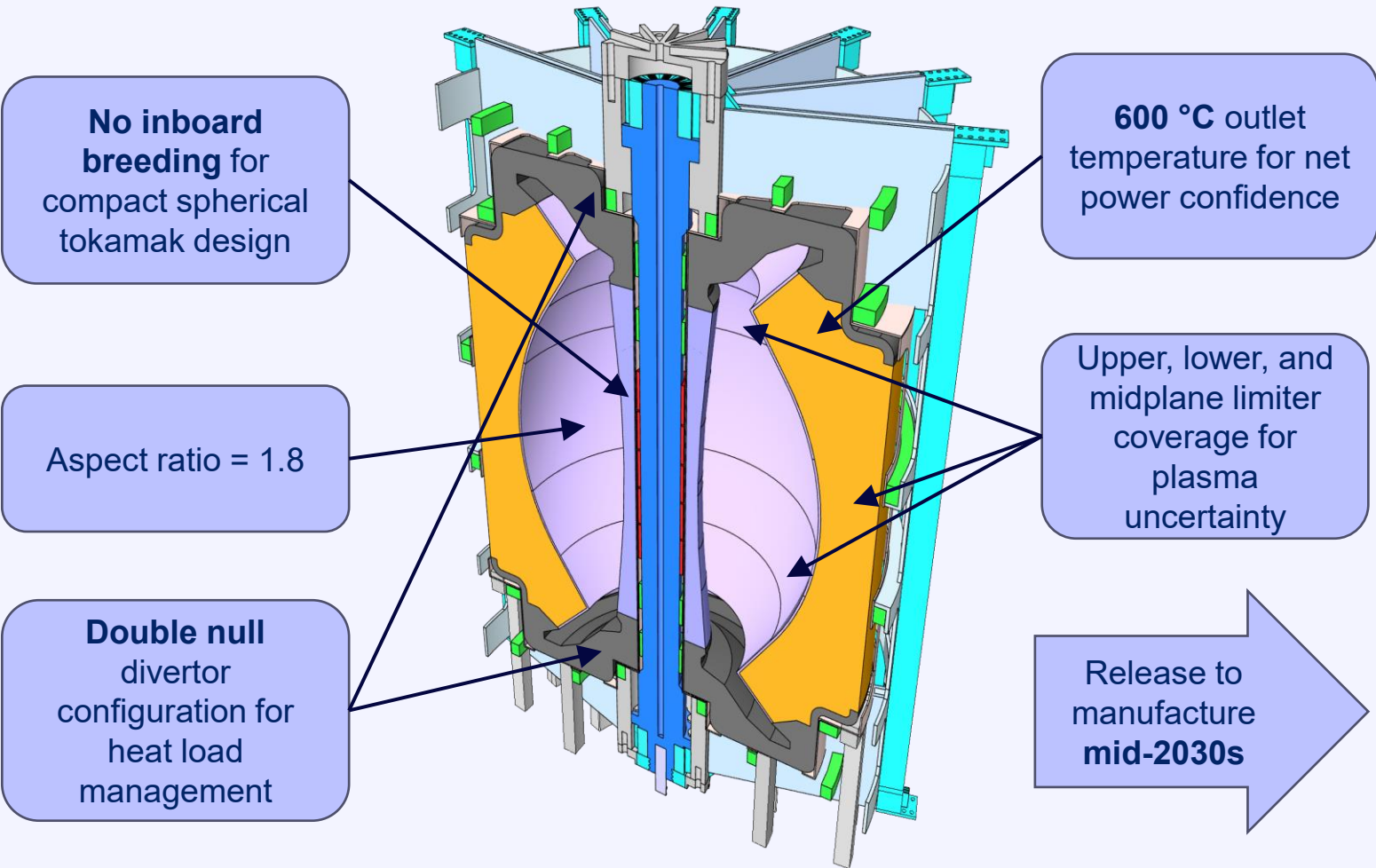
FUNCTIONAL REQUIREMENTS



Key Functional Requirements



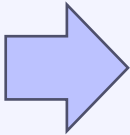
STEP Prototype Powerplant (SPP) Specific Challenges



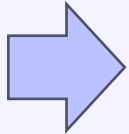
SEPTEMBER 2023: LIQUID LI AND ASSOCIATED CHALLENGES



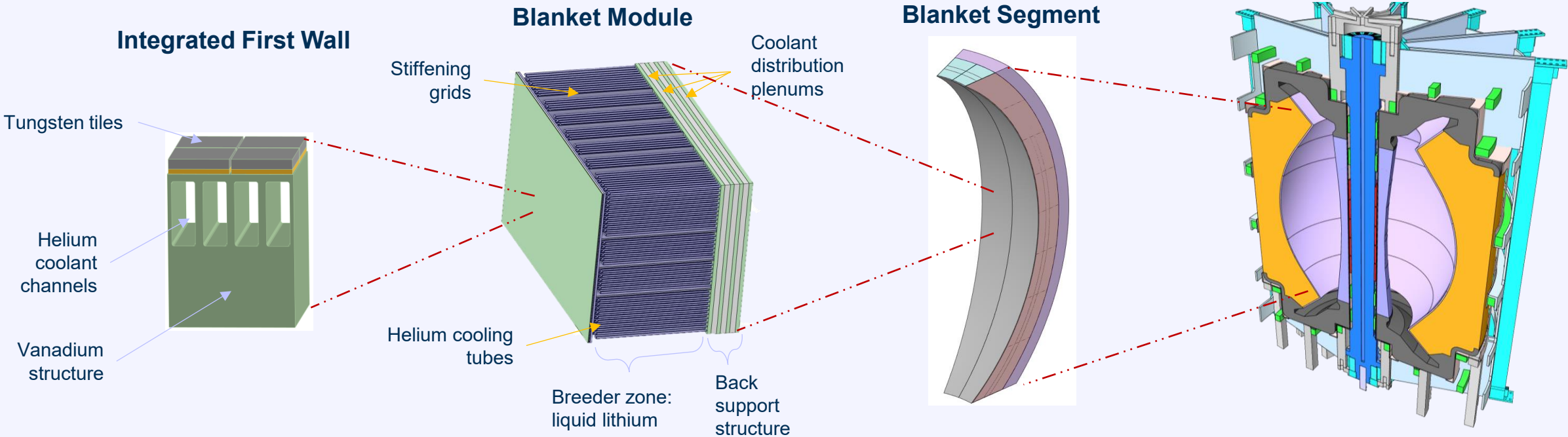
Design Rationale
<ul style="list-style-type: none">Liquid lithium blanket provides greatest breeding within spherical tokamak constraintsHelium cooling to reduce MHD issuesVanadium structure for Li compatibility



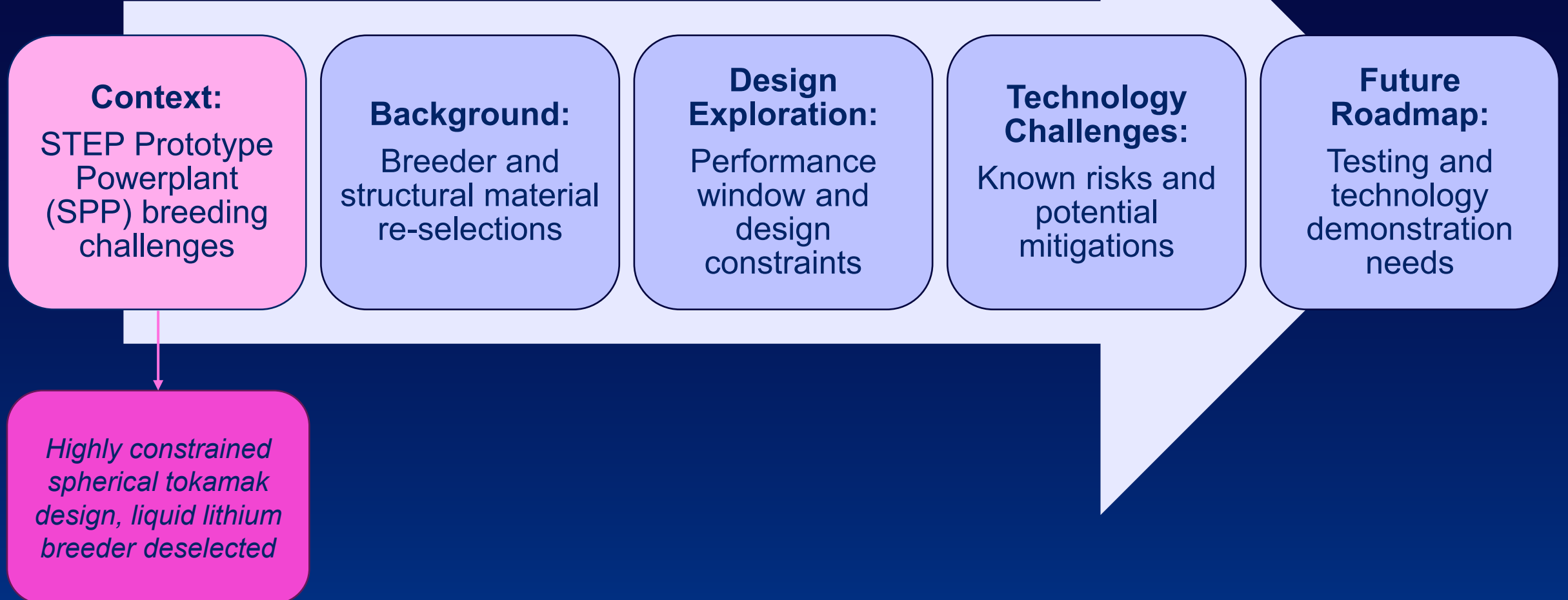
Blanket Specification
Type: Helium-Cooled Lithium
Coolant: Helium ~460 °C in, 600 °C out
Structural material: Vanadium
Breeder: Pure liquid Lithium (35–45%) Li6 enrichment
Multiplier: None



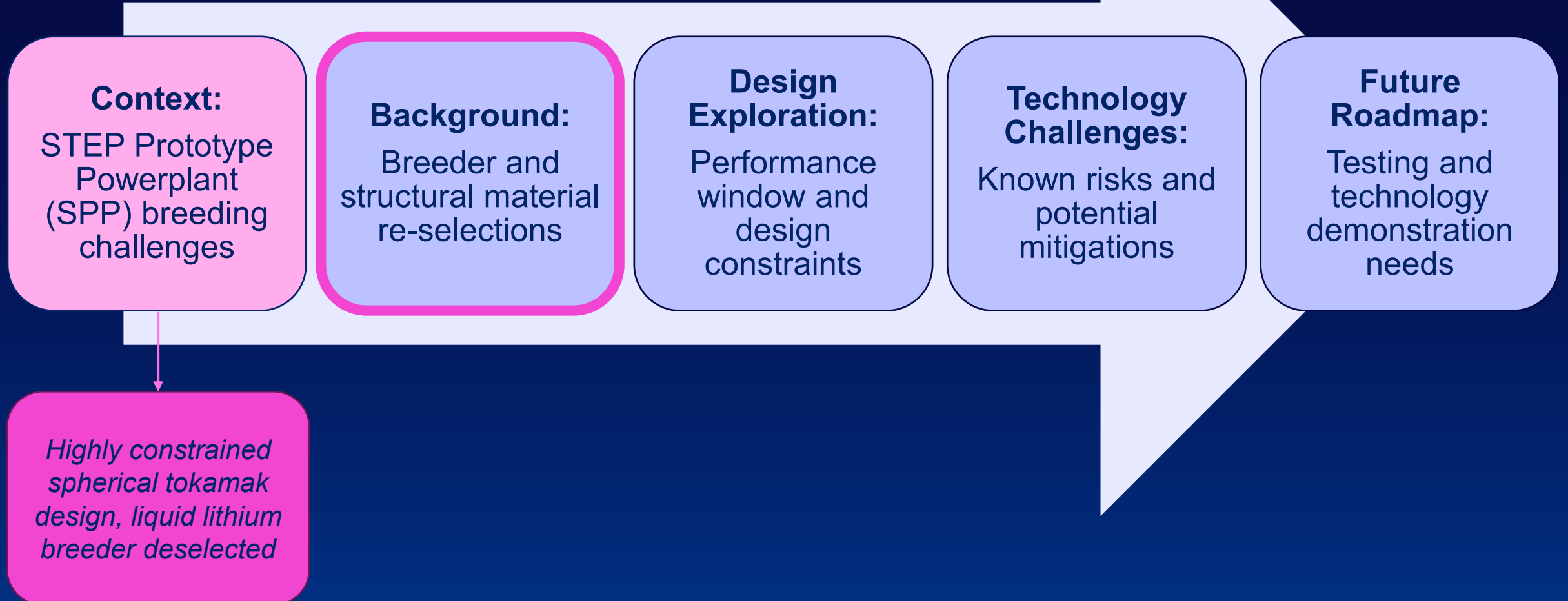
Development Challenges
Vanadium supply chain
Tritium extraction from lithium
Lithium safety and reliability
Helium leakage and pipe integration
Vanadium-lithium irradiation data



OUTLINE



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BREEDER OPTIONS: COMPARISON SUMMARY



Breeder	TBR	Tritium Transport/FC	Thermal Mgmt./Performance	Material Compatibility	Safety/Stability	Power	Composite Risk
Li	++	-	++	-	--	++	--
Li ₂ O	+	++	O	O	+	++	+
Li ₇ Pb ₂	++	--	--	O	--	++	O
Li ₄ SiO ₄	O	++	-	+	O	+	O
Li ₄ SiO ₄ -Li ₂ TiO ₃	-	++	-	+	O	+	O
Li ₈ PbO ₆	O	-	--	???	+	+	-
Li ₁₇ Pb ₈₃	--	-	++	+	O	--	--
Li ₅₀ Pb ₅₀	++	???	++	???	???	--	--
FLiBe	--	+	+	--	O	-	--

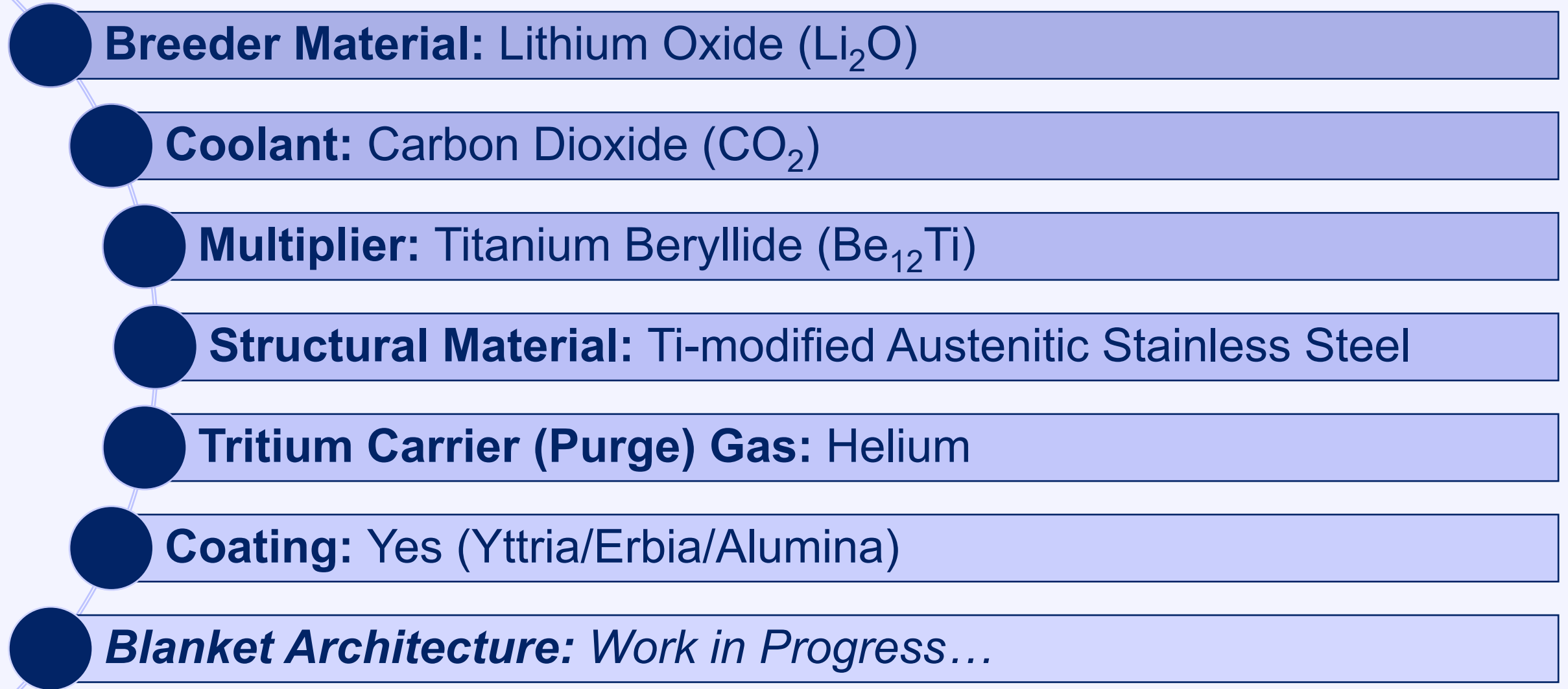
LEGEND	
++	Looks very promising, Almost no issues found or very minor concerns.
+	Looks promising. Some issues found. But generally, looks manageable.
O	Middling performance or Some important concerns, or marginal properties. Needs further study.
-	Poor performance. Some major concerns, would require massive effort.
--	Very Poor Performance. Mostly improbable for SPP timelines.
???	Significant unknowns/uncertainty constituting major risk for SPP timelines.

STRUCTURAL MATERIALS: COMPARISON SUMMARY

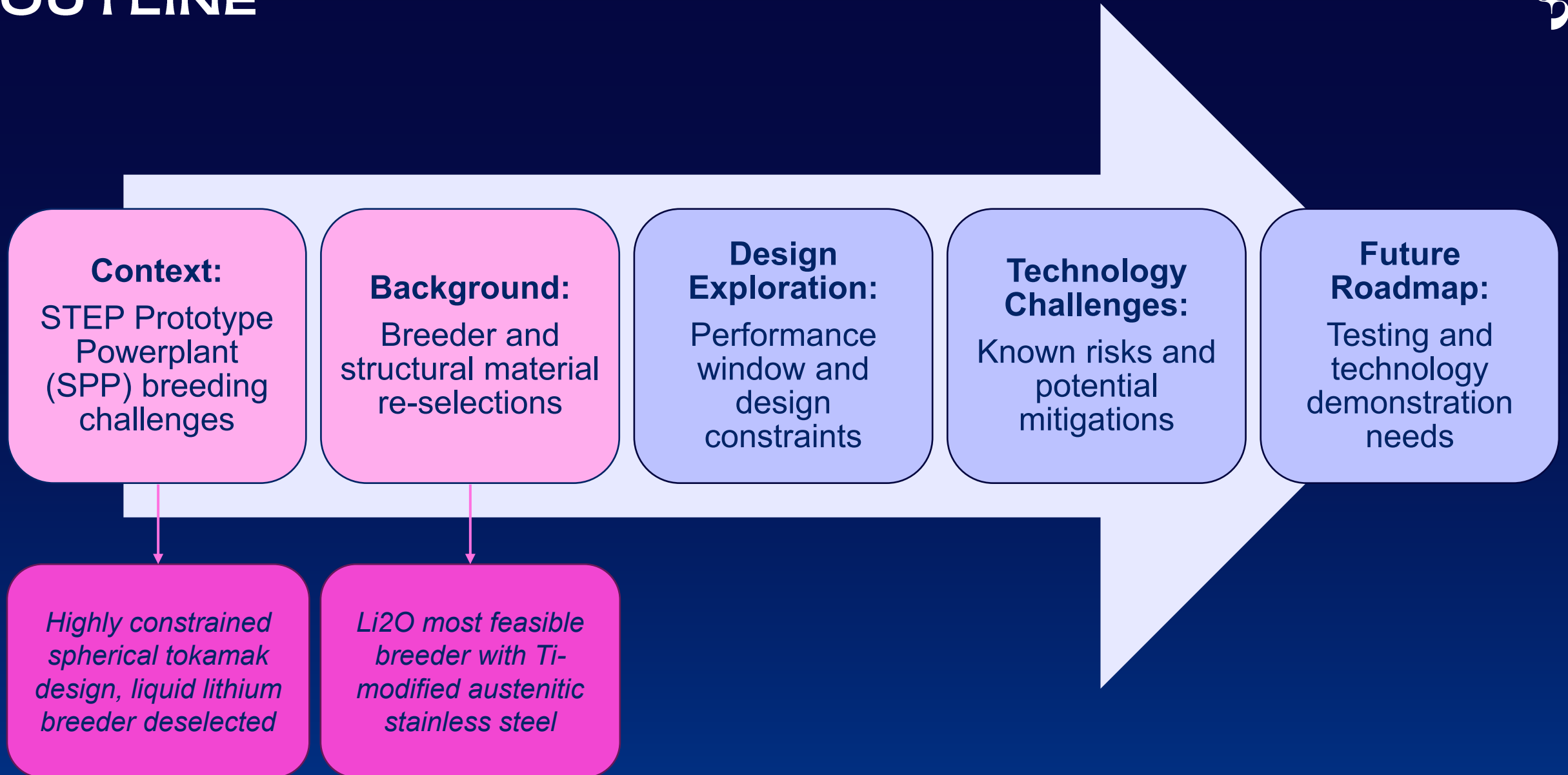


Requirements	Target performance	Grade 92	Ti-modified Austenitic Stainless Steel
Resistance to irradiation swelling	Swelling at 10 dpa < 1%	< 1% up to 100s dpa (ion irradiation)	< 1% up to 50 dpa
High temperature strength (irradiated & unirradiated)	Yield strength > 300 MPa	> 150 MPa at 650°C	YS of ~500MPa at 650°C
Creep rupture strength (unirradiated)	At 650°C for 5200h > 100MPa	~100MPa at 650°C (unirradiated)	~200 MPa at 650C
Resistance to irradiation induced embrittlement	DBTT < 350°C	249°C after n irradiation to 2.5dpa at 300°C	No DBTT
Low temperature hardening embrittlement after 10 dpa	Change in YS < 50%	Possible – YS increases 60% following irradiation to 6.5dpa at ~295°C.	YS increases 17% up to 51 dpa
Strain at rupture	(after 10 dpa) > 5%	Unlikely – total elongation 2.7% after at 6.5dpa	~9% up to 54dpa (at 510C) Creep rupture elongation ~2-3% after 5000hrs life (700C)
Fracture toughness at minimum operating temperature (10dpa)	> 70 MPa. \sqrt{m}	Likley - KJq ~145MPa.m ^{0.5} at 650°C	~60MPa.m ^{0.5} at 400°C, irradiated at 410 °C
Time to LLW (~3FPY)	< 200 years	Yes (~30years)	No (~400years)
Supply chain readiness	TRL > 4	Yes (5-6)	Yes (DIN1.4970)
Δ TBR	>= 0	0	-0.03
Coatings compatibility	Good compatibility	CTE mismatch 44%	CTE mismatch 33%
Compatibility with CO2	Good compatibility	Formation of dual-layered oxide	High-Cr alloy.
H/He embrittlement	No concern	Unknown at relevant fusion conditions	Unknown effect of He generation at high energy n

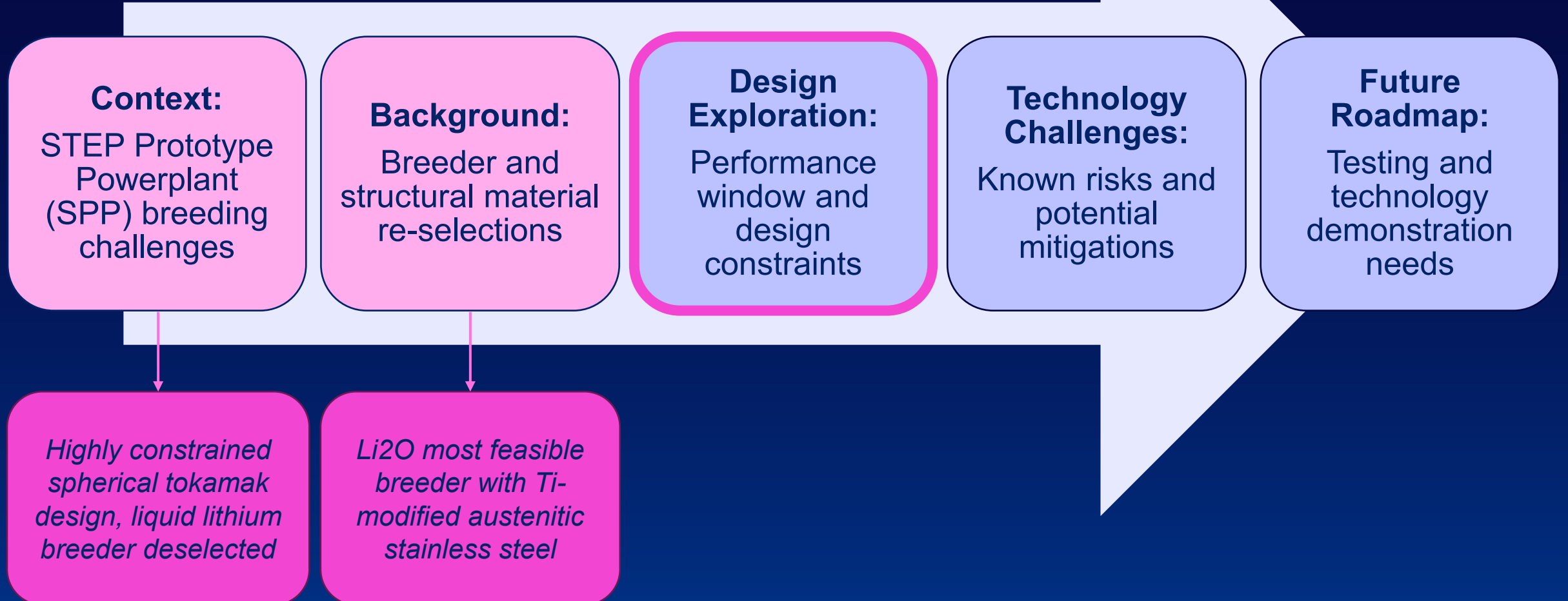
SOLID BREEDER DESIGN SPACE



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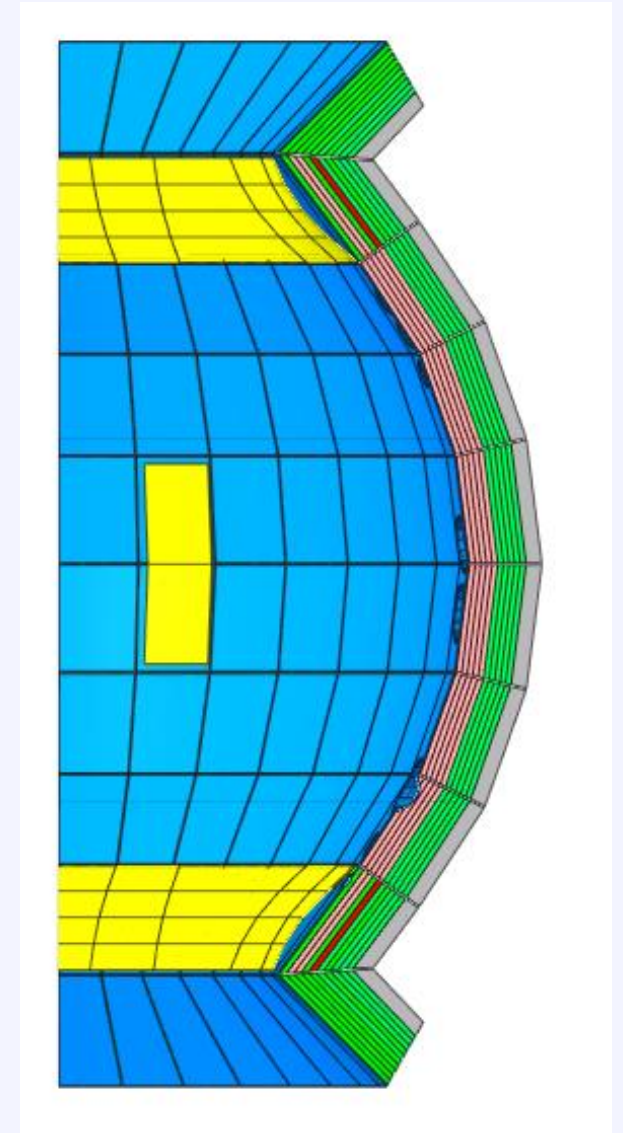
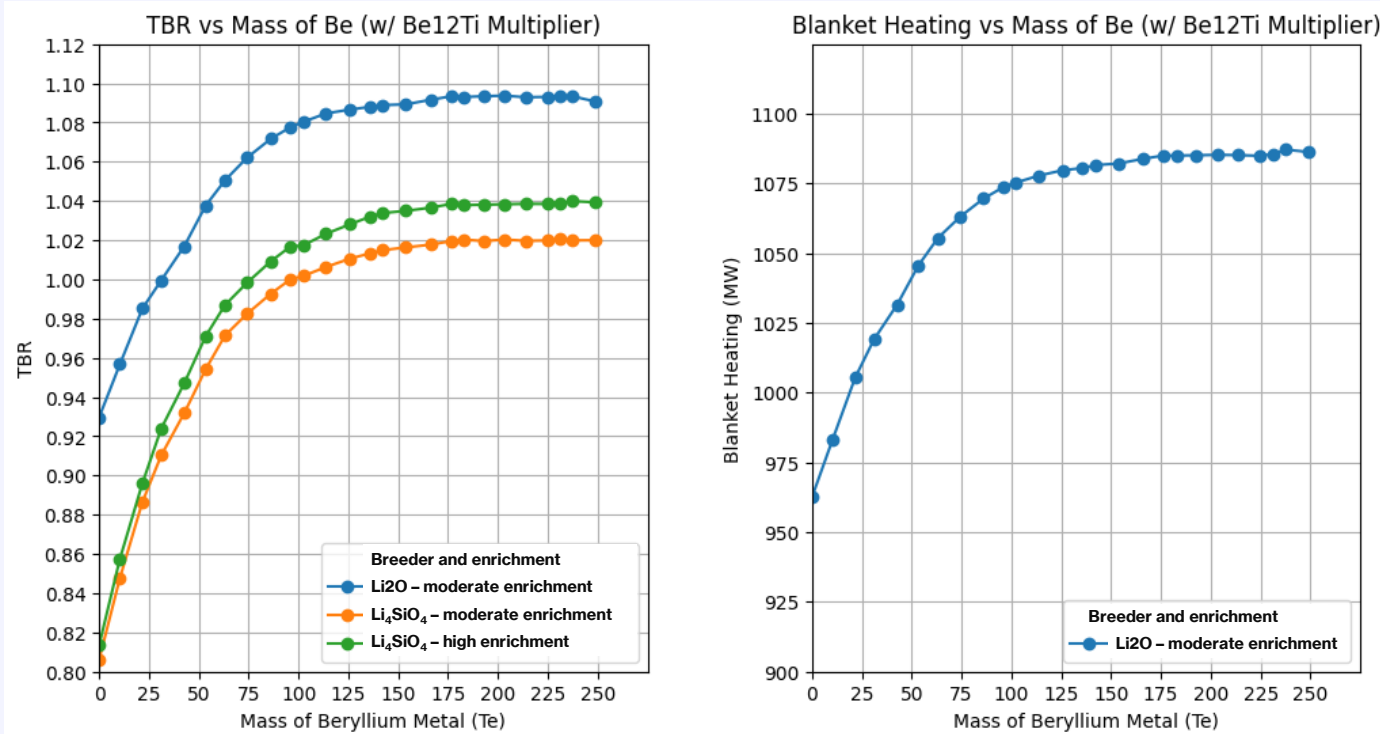
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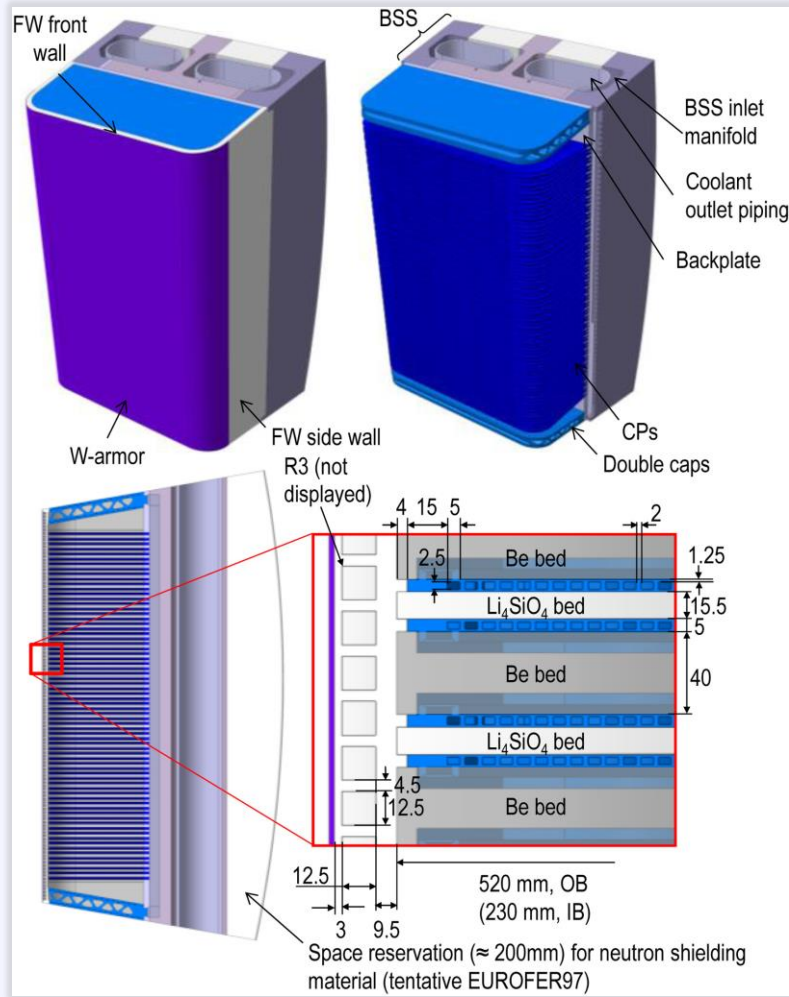
TRITIUM BREEDING PERFORMANCE



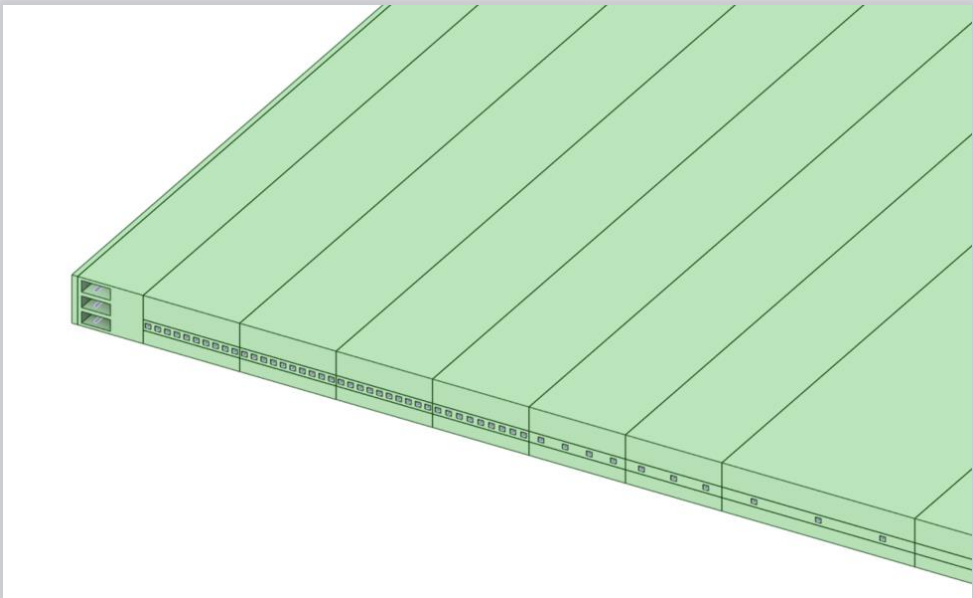
- OpenMC model with radially and poloidally varying volume fractions of breeder, multiplier, structure, and coolant
- Placement of Be12Ti placed strategically to give maximal TBR increase per additional mass (most efficient near first wall at the equatorial plane)
- TBR plateaus with around 125 Te of BeTi



THERMAL PERFORMANCE ASSESSMENT



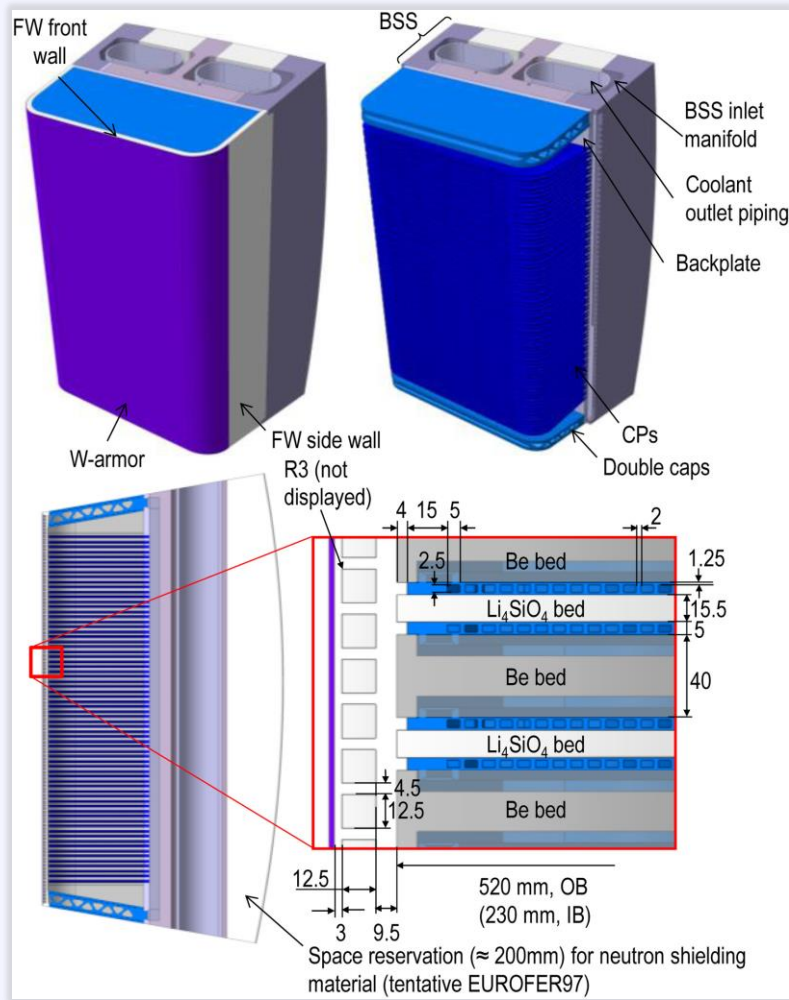
- Assume a layout similar to DEMO HCPB design 2016
- Apply volumetric heating from OpenMC calculation
- Run Ansys steady state thermal analysis (simple thermal conduction problem)
- Coolant flow rates and HTC's calculated from energy balance and number of coolant channels



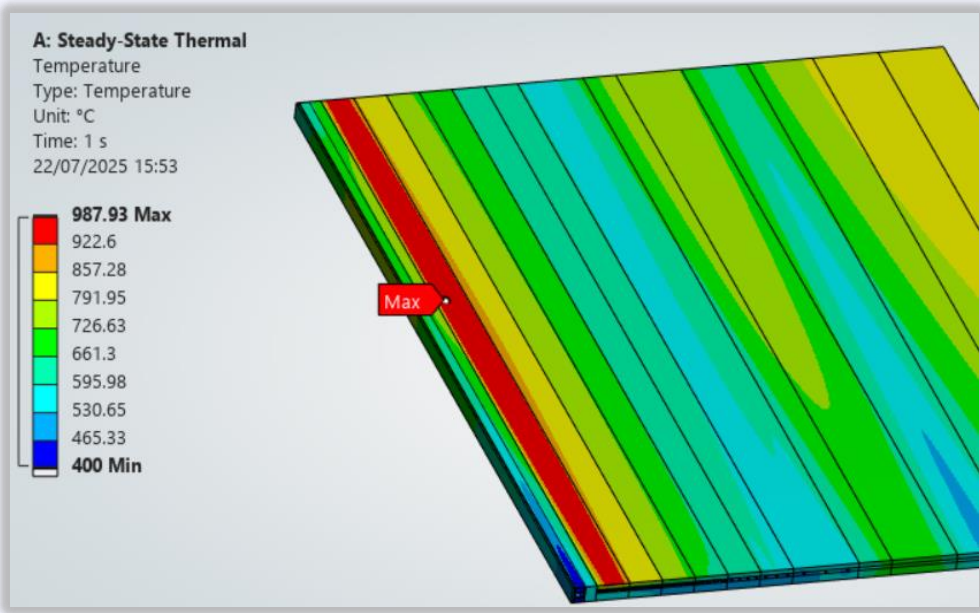
$V_{coolant}$	18 m/s
T_{in}	440 °C
Peak q'''	27.1 MW/m ³
Li ₂ O layer height	14 mm
Be ₁₂ Ti layer height	25 mm
Plate height	5.5 mm

F. Hernandez et al, Fus. Eng. Des., 124 (2017) 882-886

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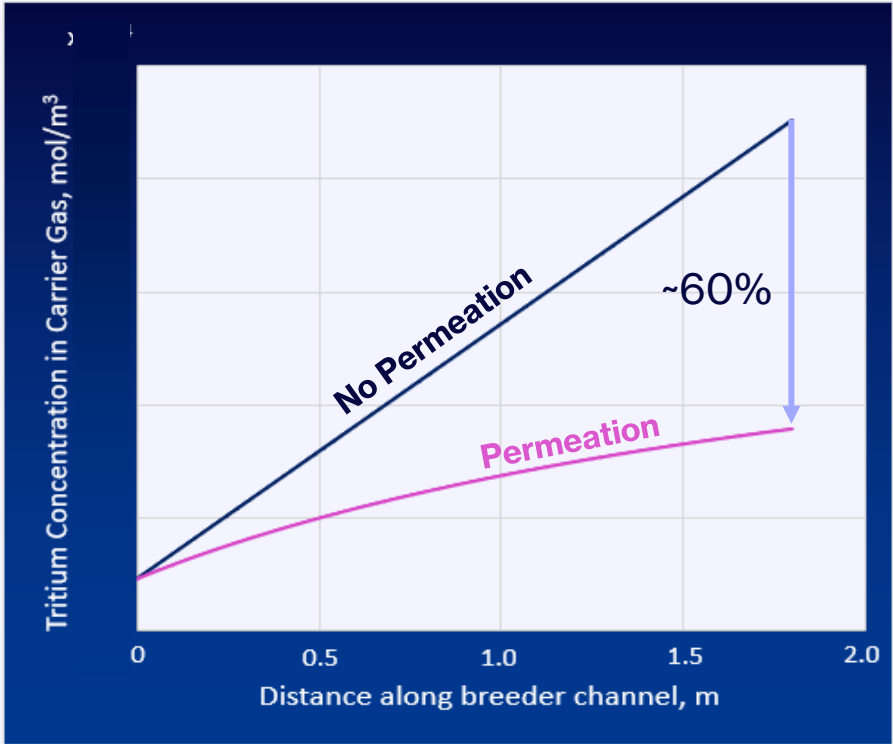
F. Hernandez et al, Fus. Eng. Des., 124 (2017) 882-886

TRITIUM PERMEATION ASSESSMENT



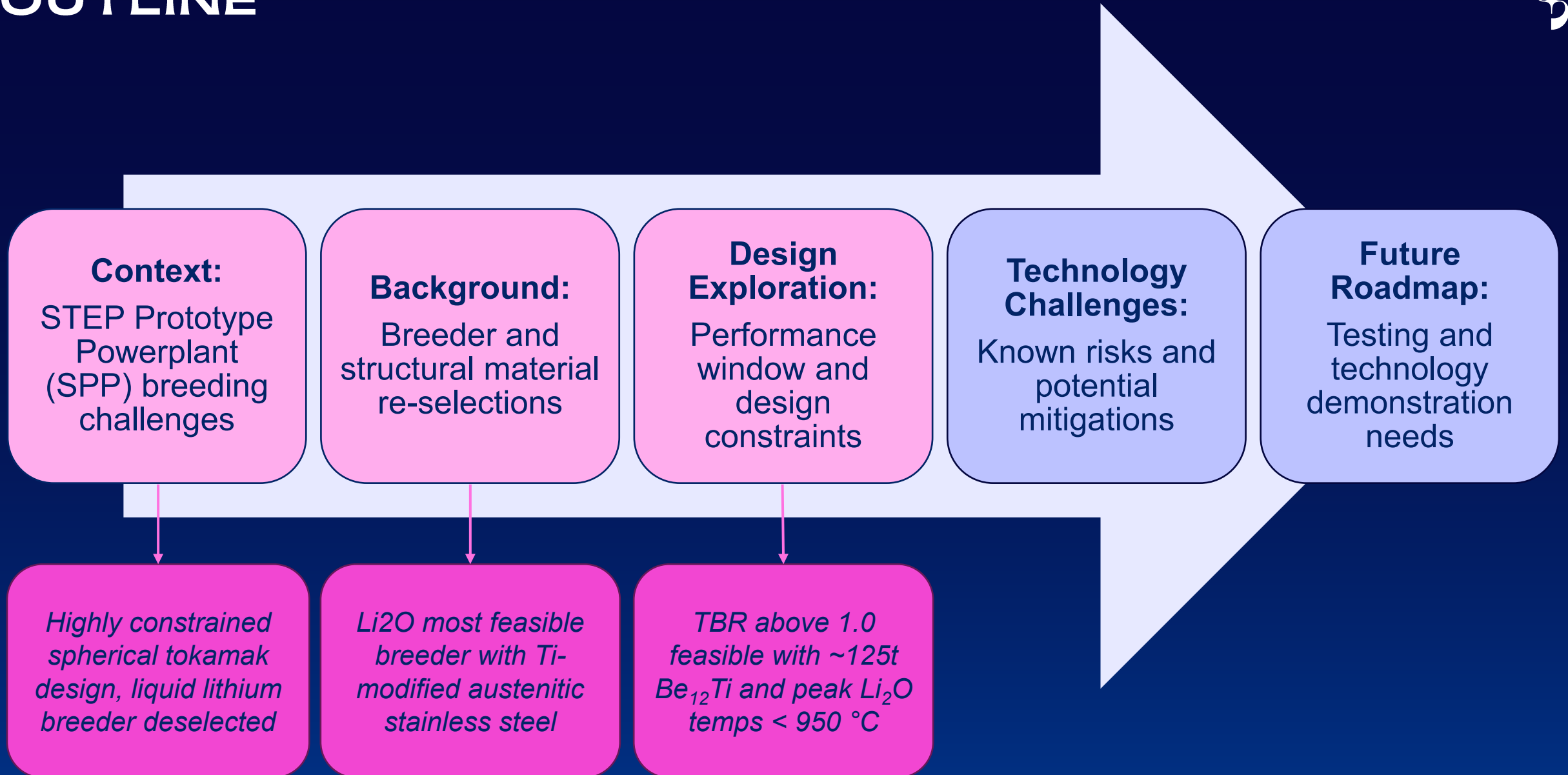
- COMSOL model of unit cell to model tritium permeation into coolant
- Worst-Case Scenario:
 - Significant amount of bred tritium permeates into the coolant
 - Tritium Extraction System (TES) only recovers smaller amount of bred tritium
- Sensitive to temperature, but reduction not possible without significant decrease from 600 °C outlet temp
- Higher purge gas flow helps but quickly leads to high pumping powers

➡ Importance of permeation reduction coatings and/or coolant detritiation

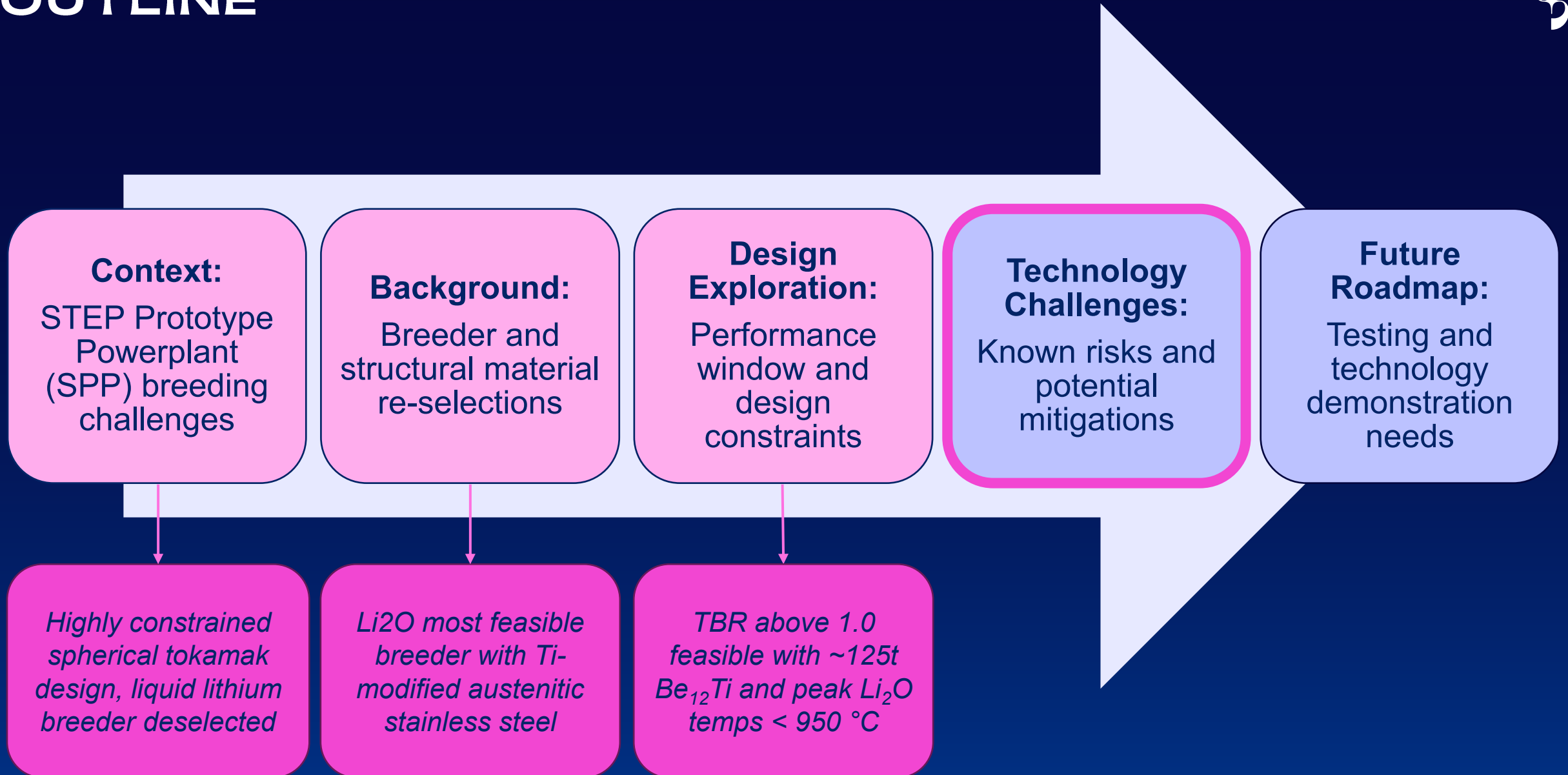


Breeding rate	2.5 x10 ⁻⁶ mol/s/m3
Gas velocity	2 cm/s
TES efficiency	90%
Present species	HT (not HTO)
Trapping effects	None

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HISTORICAL ISSUES REPORTED WITH LI2O



Stability, reactivity, and chemical degradation

- LiOH generation and equilibrium concentration
- Free oxygen equilibrium concentration
- Reactivity with CO₂

Corrosion and influence on structural material

- Contact corrosion rates
- LiOH corrosion rates
- Free oxygen corrosion rates
- (Dependency of structural material choice)

Mechanical degradation

- Sintering – rates and temperature/irradiation dependence
- Swelling rates under irradiation
- Creep deformation rates
- Mechanical strength
- Fracture / pulverisation under irradiation

Tritium diffusivity

- Dependency on oxide layer formation
- Dependency on grain growth, bubble or porosity formation
- Impact of manufacturing process
- Impact of redeposited corrosion products

HISTORICAL DATA: AN 800 °C TEMPERATURE LIMIT?



As we have seen (Section 2.1.2), a major concern with solid breeding materials is preventing them from experiencing temperatures outside their operating range. For Li₂O (see Table 5-3), this range is 410 to 800°C.

Jackson et al., A Review of Fusion Breeder Blanket Technology, CFFTP-G-84033, 1985

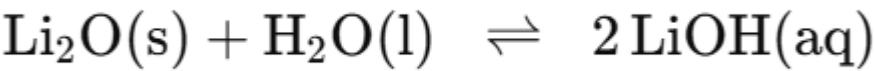
has a T₂O pressure of ~ 10⁻⁵ atm. Substitution of these pressures into eq. (3) indicates that the Li₂O temperature should not exceed 800°C. The effect of 10 vppm

Tetenbaum and Johnson, Journal of Nucl. Mat. 120 (1984) 213-216

PROPERTIES	RECOMMENDED TEMPERATURE LIMITS	
	MP, °C	T _{min} , °C
Li ₂ O	1433	410 ^d
γ-LiAlO ₂	1610	300 ^d
Li ₅ AlO ₄	1047	350 ^d
Li ₂ SiO ₃	1200	410 ^d
Li ₄ SiO ₄	1250	320 ^d
Li ₂ ZrO ₃	1616	400 ^e
Li ₈ ZrO ₆	1295	350 ^e
Li ₂ TiO ₃	1550	400 ^e

Blanket comparison and Selection Study, ANL/FPP-83-1, Volume II, 1983

LiOH–H2O EQUILIBRIUM HISTORICAL ASSESSMENT



- The peak temperature limit allowable for the Li2O is based on equilibrium concentrations of H2O and LiOH expected to be in the system
- The equation used is from *Tetenbaum, J. Nucl. Mater. 120 (1984) 213–216*:

$$\log P_{(\text{LiOH,g})/\text{atm}} = - 8635/T + 1/2 \log P_{(\text{H}_2\text{O,g})} + 4.57 \tag{19}$$

- This gives equilibrium partial pressures of LiOH and H2O at a given temperature

Temps in °C		Achievable H2O concentration (ppm)					
		100	<u>10</u>	1	0.1	0.01	0.001
Allowable LiOH concentration (ppm)	100	1041	1150	1277	1430	1616	1849
	10	868	948	1041	1150	1277	1430
	<u>1</u>	735	797	868	948	1041	1150
	0.1	629	679	735	797	868	948
	0.01	544	584	629	679	735	797
	0.001	473	507	544	584	629	679

- The **800 °C** limit is chosen based on assumptions for:
 - The *allowable* LiOH concentration to limit loss of lithium from the breeder volume = 1 ppm
 - The *achievable* H2O/HTO concentration in the purge gas = 10 ppm
- Different assumptions in ppm will lead to different interpretations of the peak temperature limit

LiOH-H₂O EQUILIBRIUM HISTORICAL ASSESSMENT

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- The equation used is from *Tetenbaum, J. Nucl. Mater. 120 (1984) 213-216*:

$$\log p_{\text{LiOH}} = 4.5013 - 9297.9/T + 0.5 \log p_{\text{H}_2\text{O}} \quad (22)$$

C.E. Johnson et al., International Workshop on Ceramic Breeder Blanket Interactions, University of Tokyo, 1992

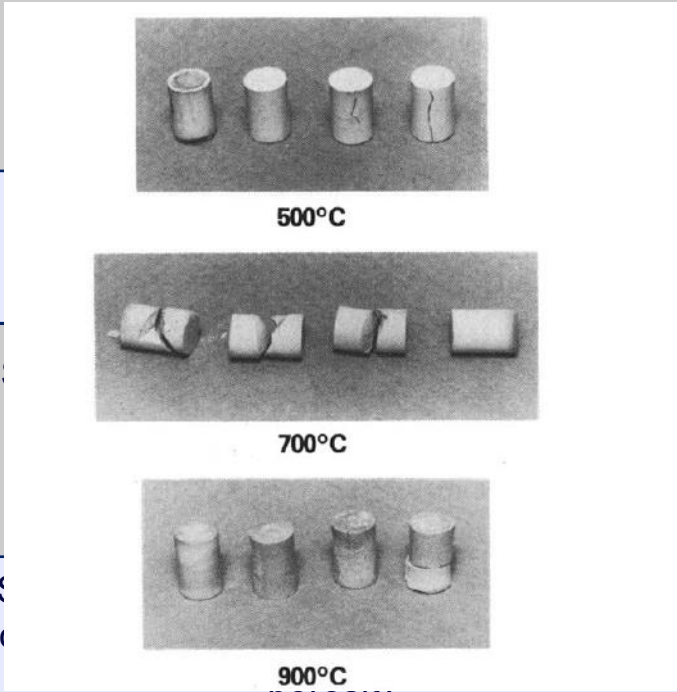
- This gives equilibrium partial pressures of LiOH and H₂O at a given temperature

		Achievable H ₂ O concentration (ppm)					
Temps in °C	Allowable LiOH concentration (ppm)	Achievable H ₂ O concentration (ppm)					
		100	10	1	0.1	0.01	0.001
100 200 300 400 500 600 700	100	1157	1276	1417	1586	1793	2051
	10	967	1055	1157	1276	1417	1586
	1	821	889	967	1055	1157	1276
	0.1	706	760	821	889	967	1055
	0.01	612	657	706	760	821	889
	0.001	535	572	612	657	706	760
	0.0001	478	508	535	572	612	657

- The **800 °C** limit is chosen based on assumptions for:
 - The *allowable* LiOH concentration to limit loss of lithium from the breeder volume = 1 ppm
 - The *achievable* H₂O/HTO concentration in the purge gas = 10 ppm
- Different assumptions in ppm will lead to different interpretations of the peak temperature limit
- Later data suggests an 889 °C limit for the same assumptions!*

Li₂O IRRADIATION SWELLING

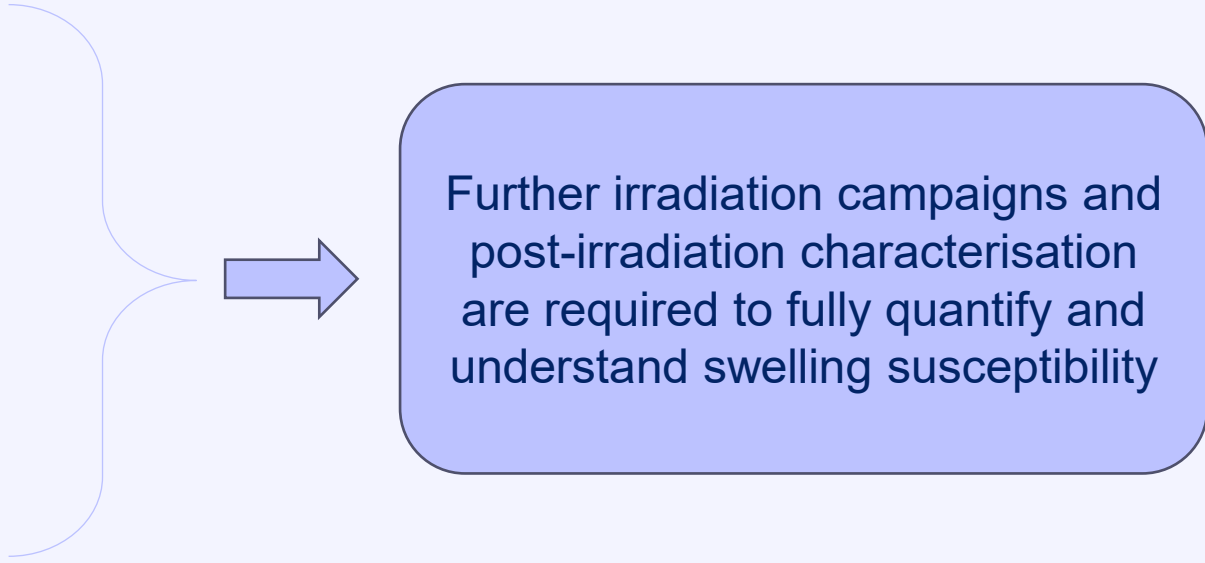


Programme	Reactor	Start	Li Burnup	Effective full power days	Reported Volumetric Swelling	Comments
FUBR-1A	EBR-II	1982	~0.5-3%		6 – 12%	Shrinkage @ 500 °C, attributed to poor moisture control and LiOH generation.
FUBR-1B	EBR-II	1985	2.5-7%		-8.5 – 4.4%	 <p>500°C</p> <p>700°C</p> <p>900°C</p>
BEATRIX-I	EBR-II	1986	7-11%*			
BEATRIX-II: Phase I	FFTF	1990	4.8%	300	~4 – 7%	
BEATRIX-II: Phase II	FFTF	1991	4.6%	203	*~20%	
CRITIC-1	NRU	1986	1%		Not reported	

Li₂O IRRADIATION SWELLING CONCLUSIONS

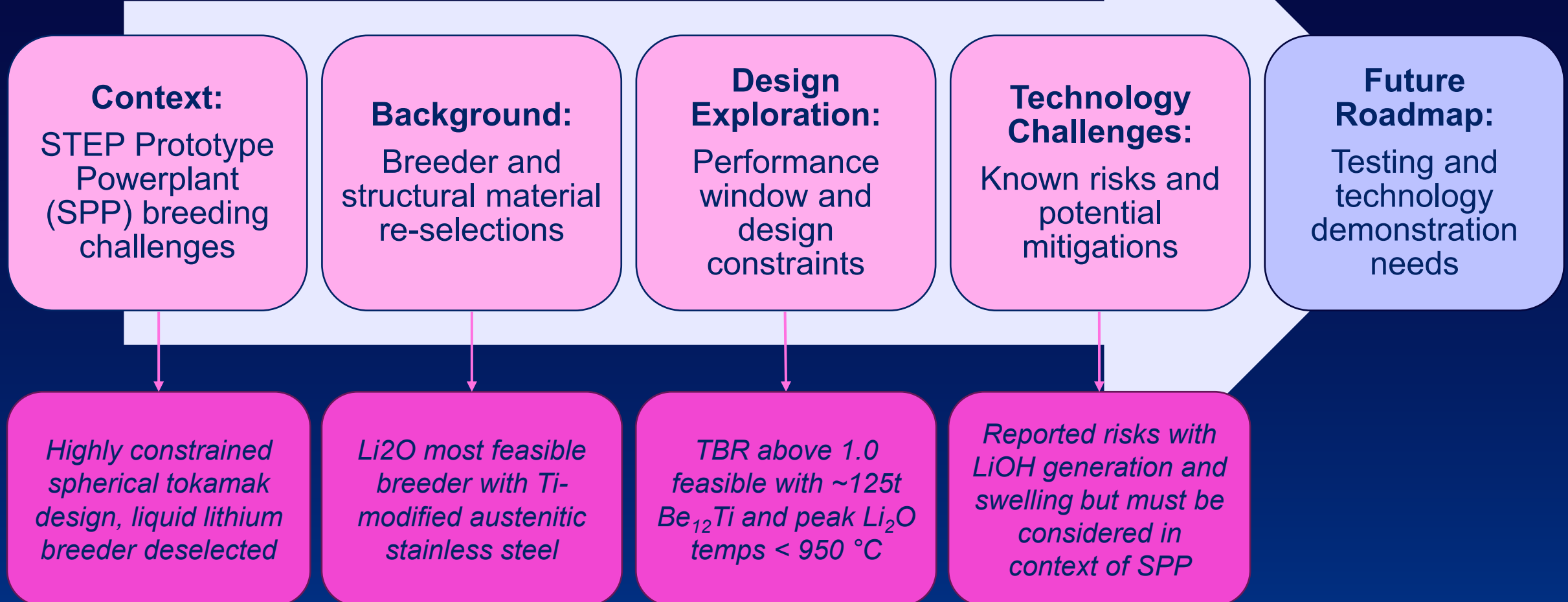


- Despite multiple irradiations, the data, reporting, and understanding of swelling of Li₂O is patchy. Swelling appears to be in the range of 5 – 8% but may be as high as 20%
- Influencing factors include:
 - Bulk temp / temp gradient
 - H₂O/T₂O concentration
 - Geometry/Particle dimensions
 - ⁶Li enrichment / burnup rate
 - Stress loading and creep rate
 - Initial density / porosity
 - Initial grain size
 - He retention & defect concentration
- **A mitigating factor for the SPP *may* be lower burnup, burnup rate and availability. However, limitations for commercial plant would still be present.**

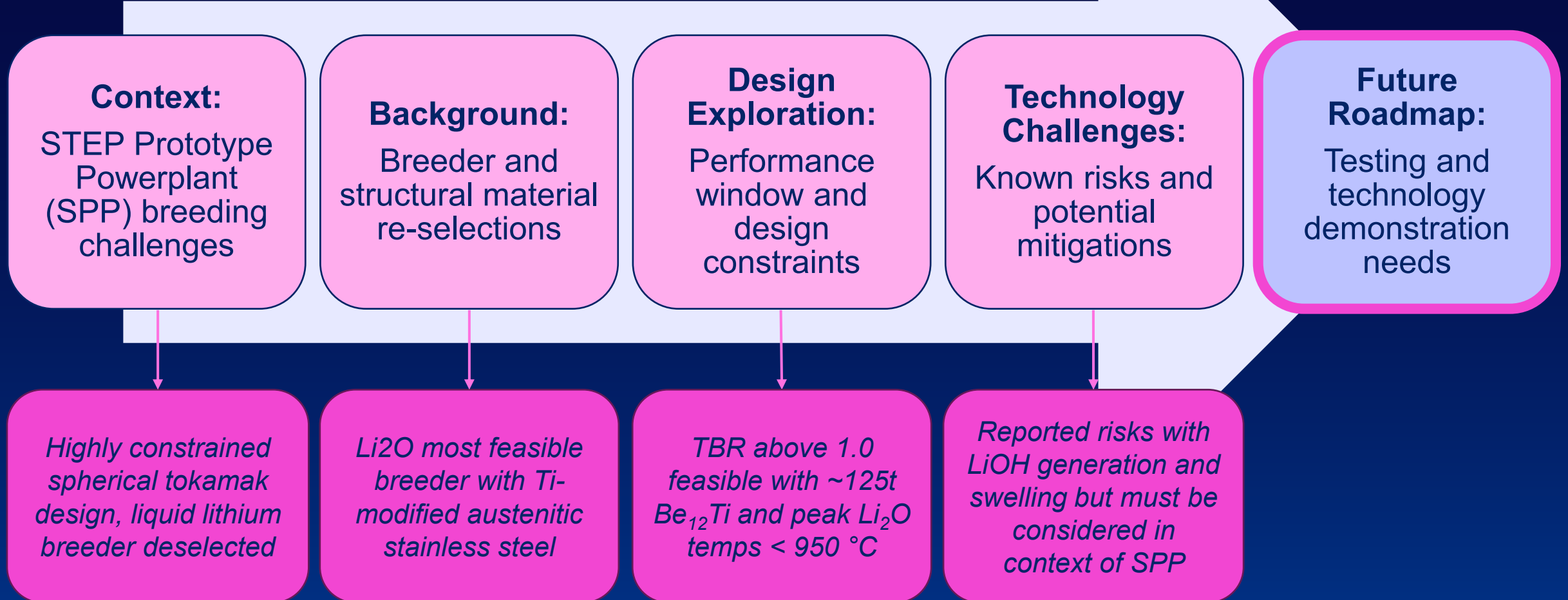


Further irradiation campaigns and post-irradiation characterisation are required to fully quantify and understand swelling susceptibility

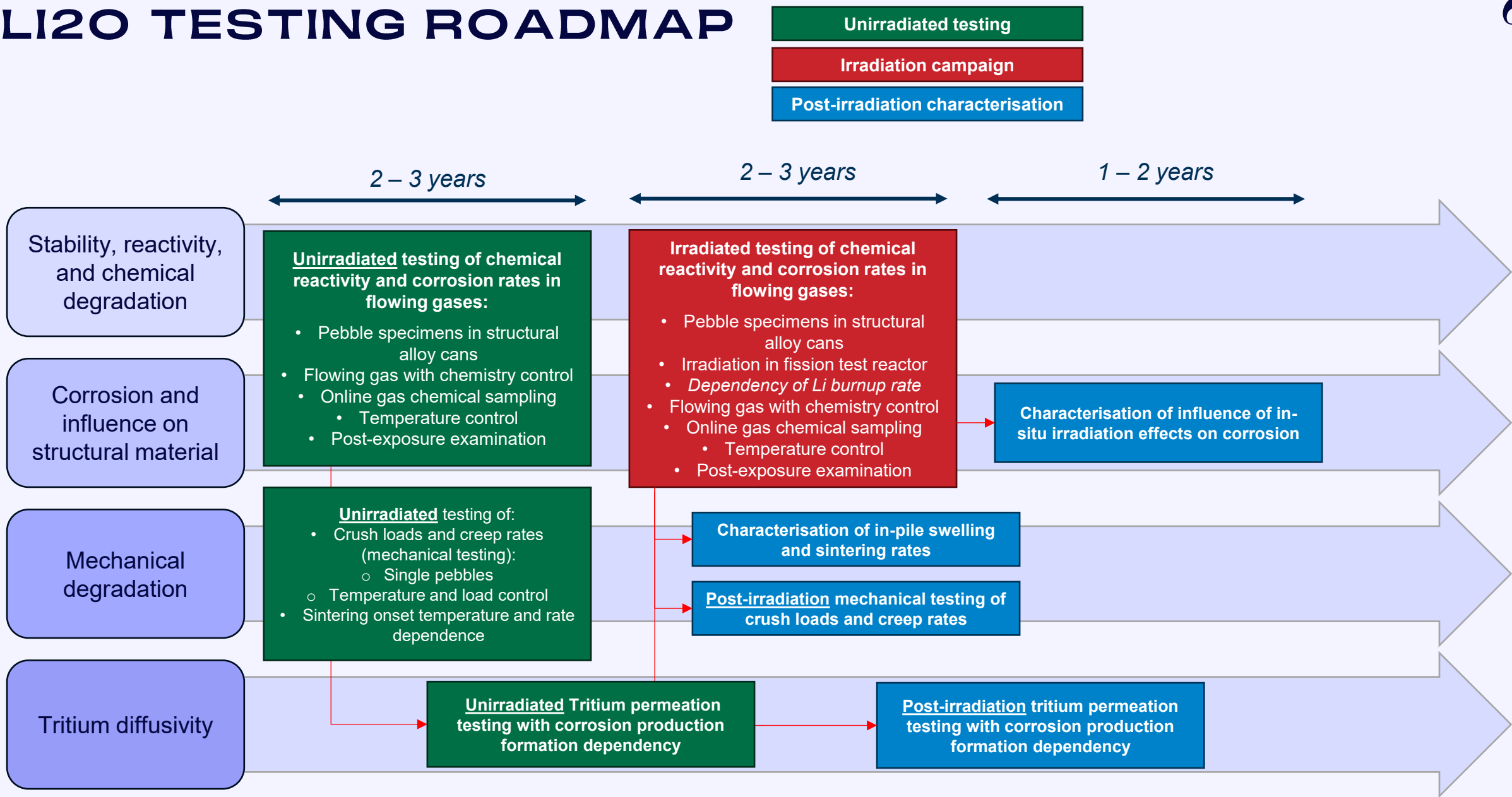
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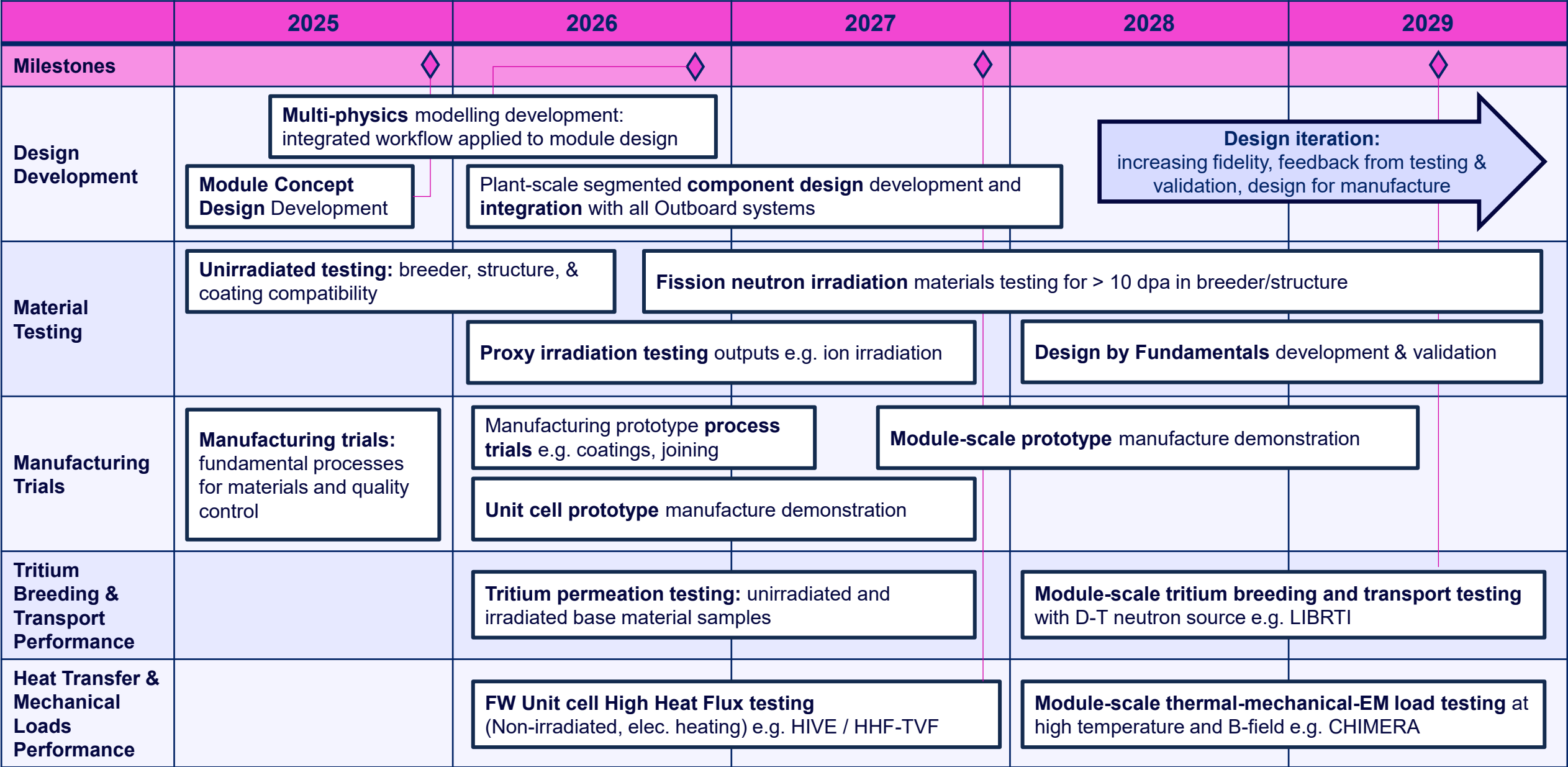
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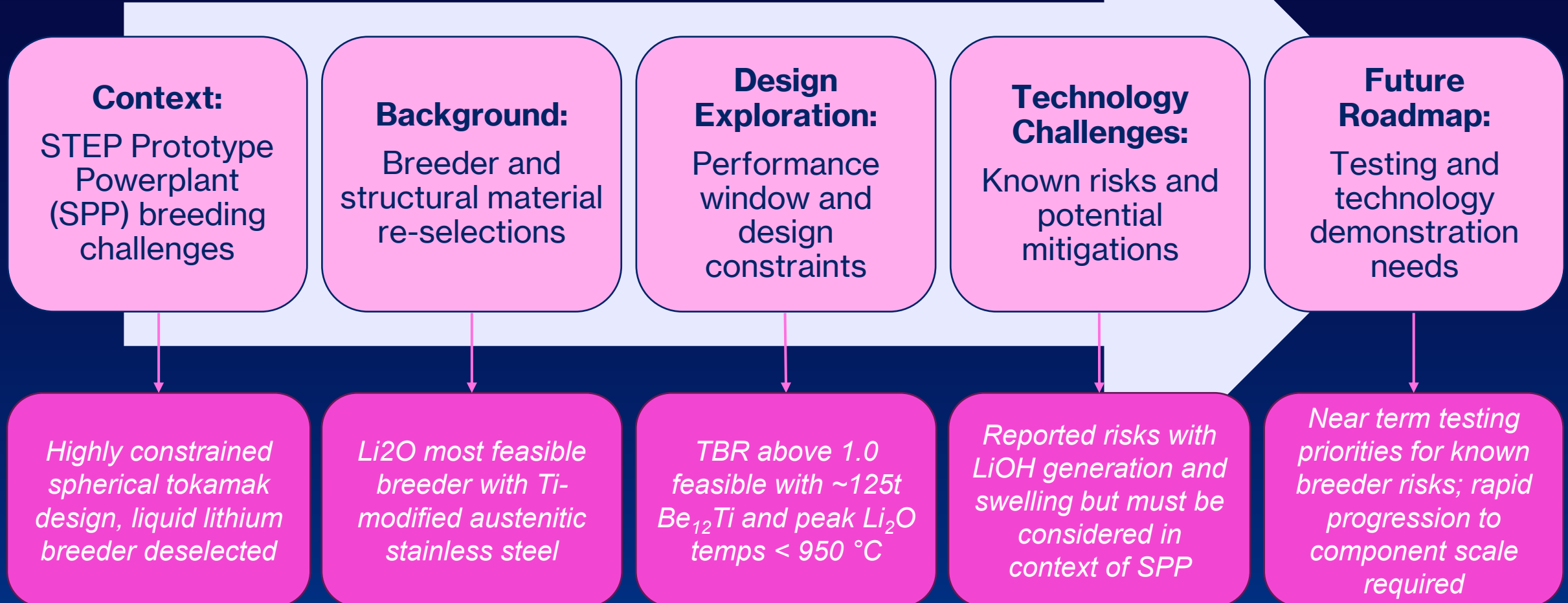
Li2O TESTING ROADMAP



5-YEAR BLANKET DEVELOPMENT ROADMAP



SUMMARY



STEP 

**THANK YOU
ANY QUESTIONS?**