

Numerical Simulation of Compositional Redistribution Driven by isotopologue Fractionation During Solidification of D-T Fuel in ICF Targets

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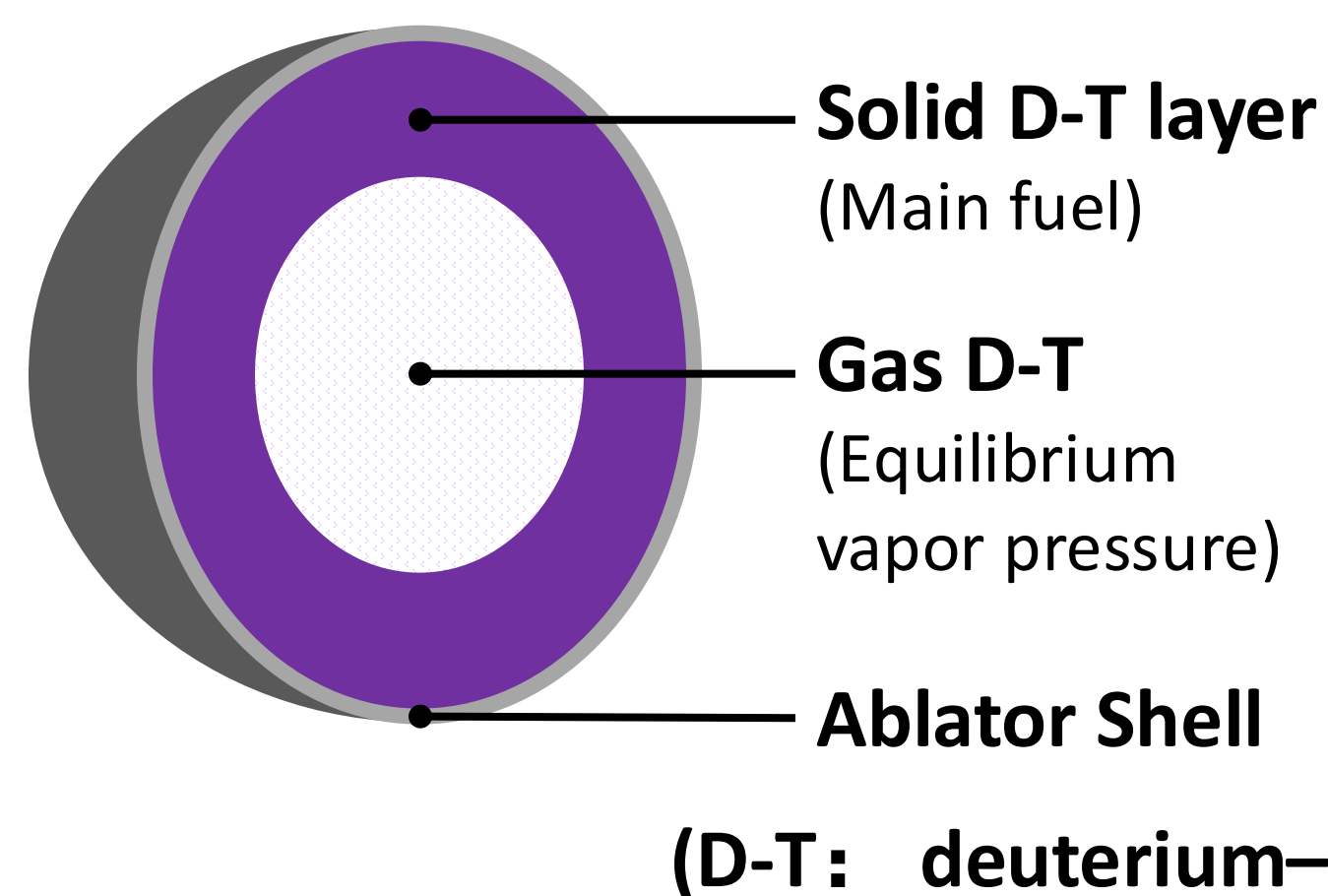
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ABSTRACT

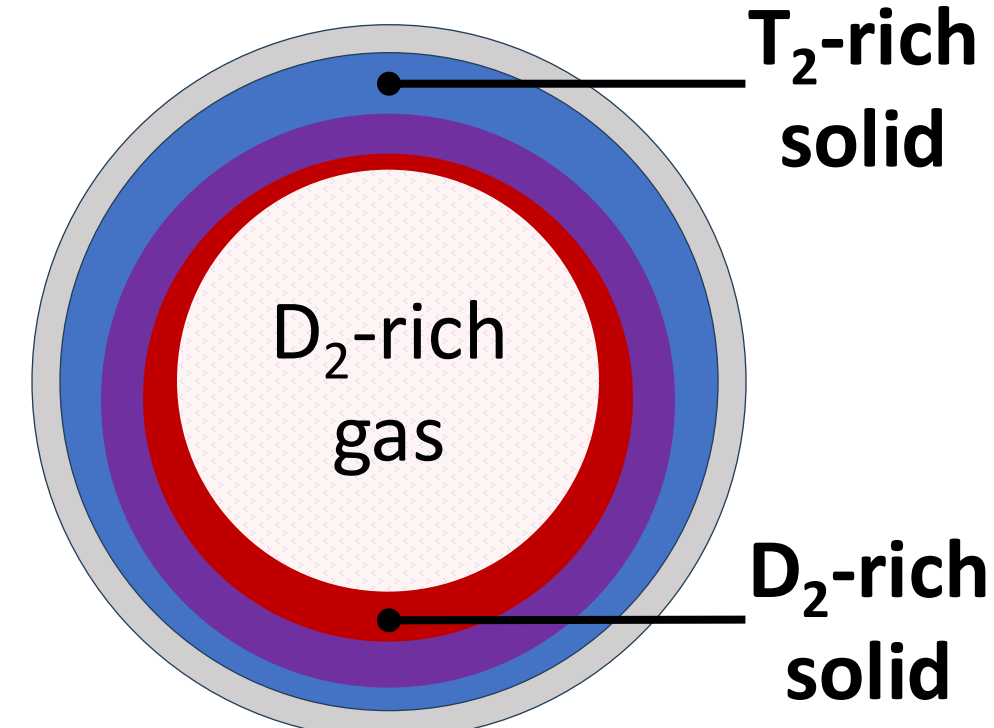
- In ICF, compositional homogeneity of cryogenic D–T fuel targets is critical for fusion performance.
- Isotopologue fractionation induces a radial concentration gradient within the solid D-T layer and alters the equilibrium composition of the central D-T gas.
- A comprehensive numerical framework was developed to simulate the compositional distribution in D-T mixture during solidification.
- The detrimental effects of composition inhomogeneity on fusion efficiency were examined.

BACKGROUND

Cryogenic D-T target for ICF



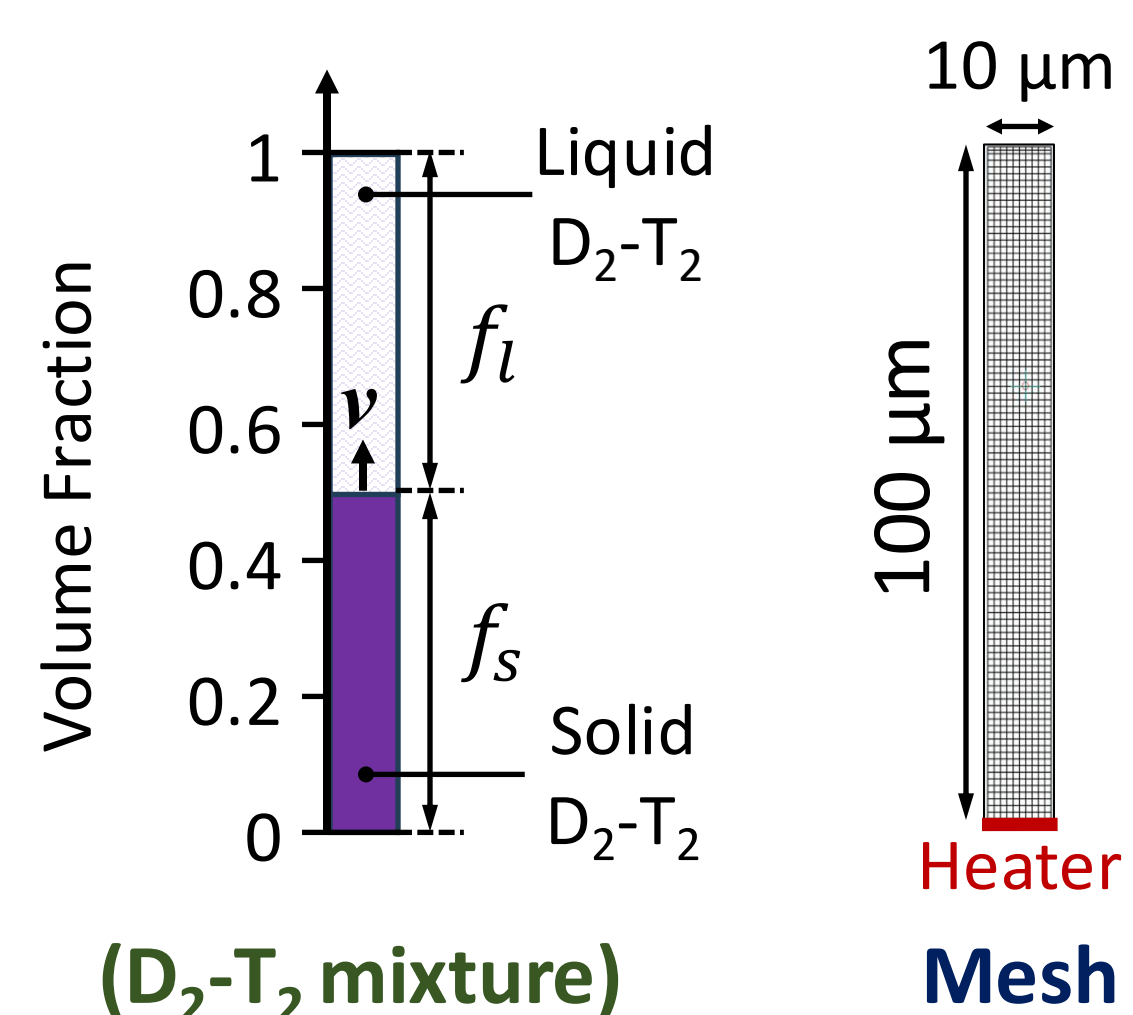
Fractionation



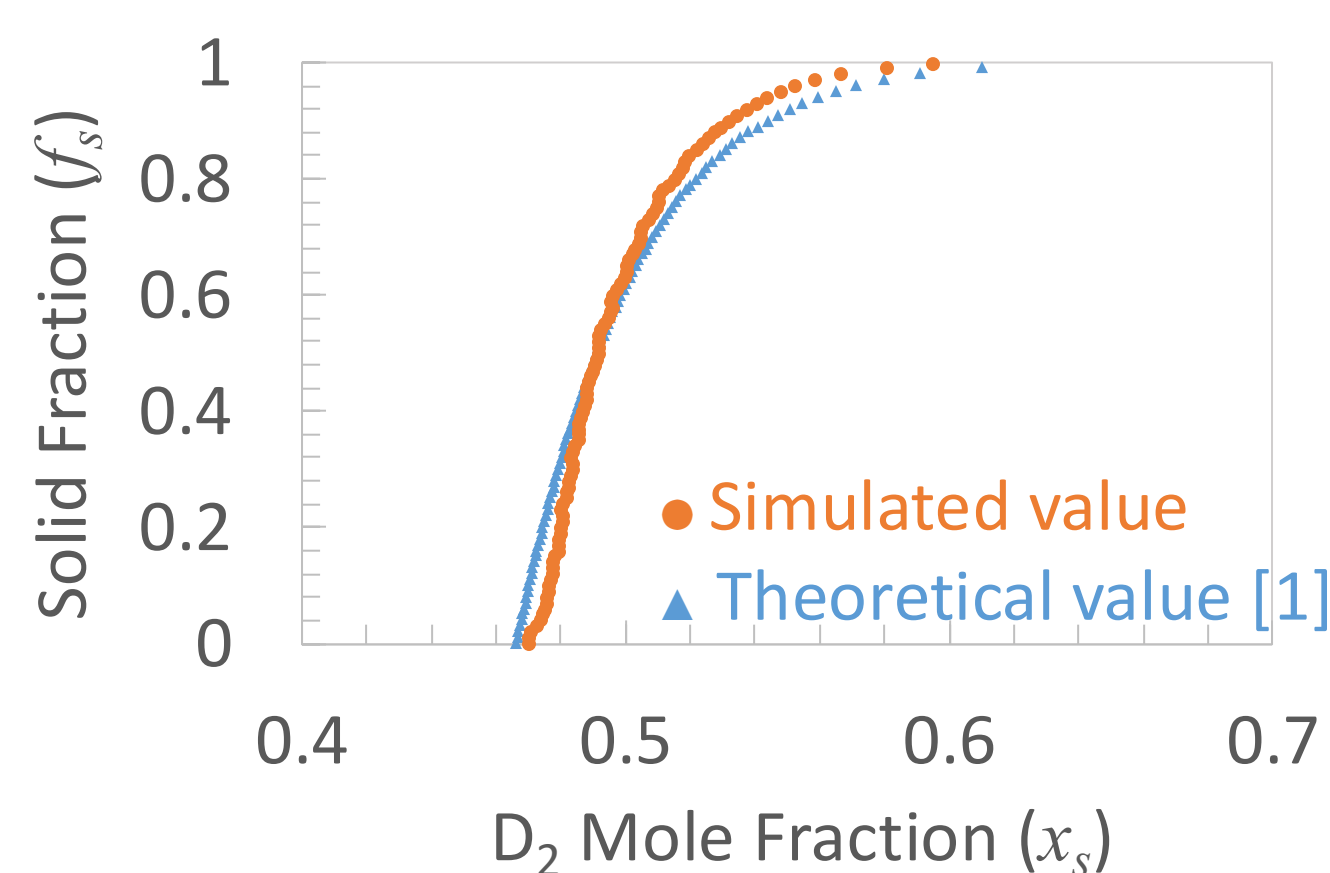
Numerical simulation method was developed to quantify the compositional distribution in D-T target during solidification

SIMULATION 1: Compositional distribution during unidirectional solidification

Unidirectional Solidification



Simulated versus theoretical

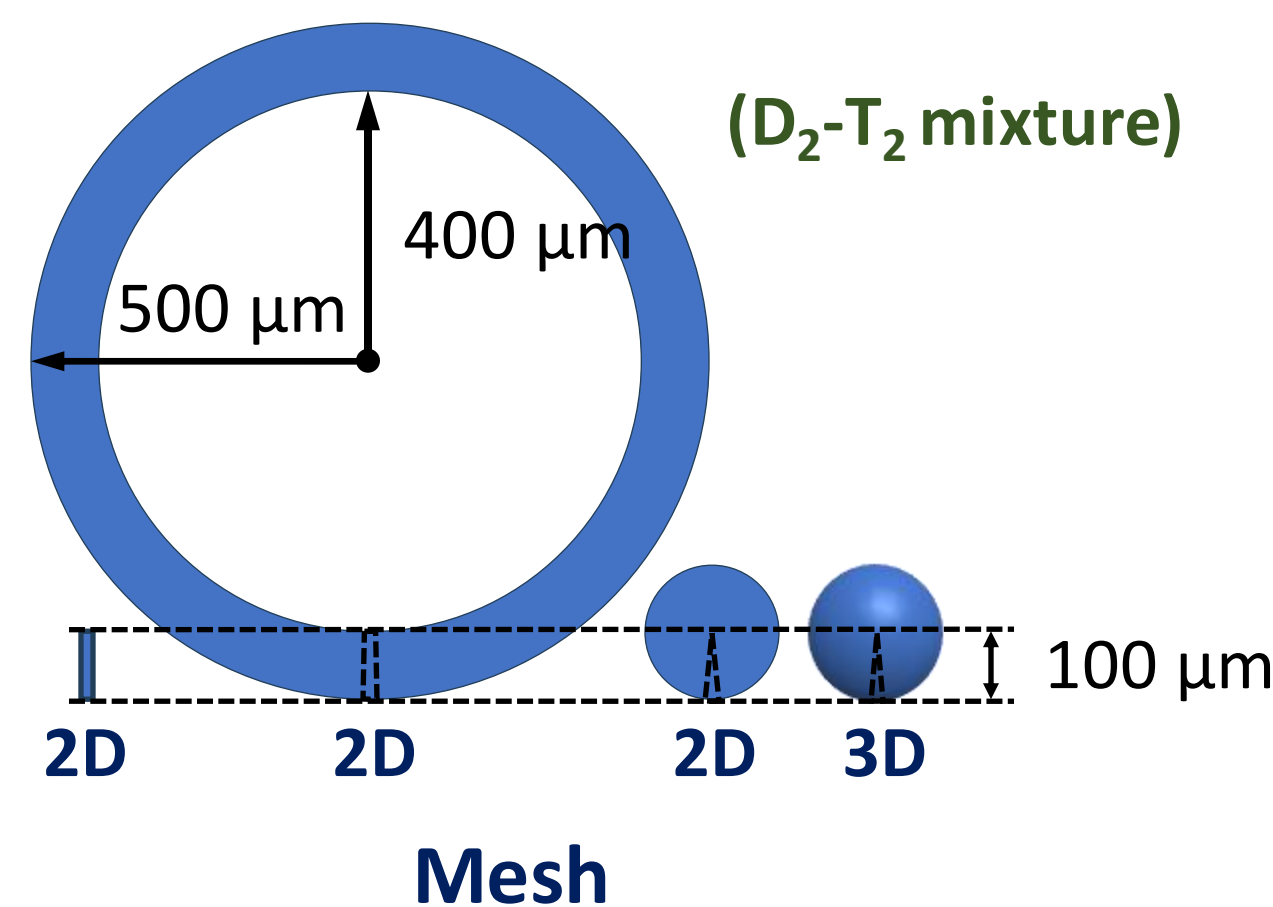


Maximum fractionation [2]

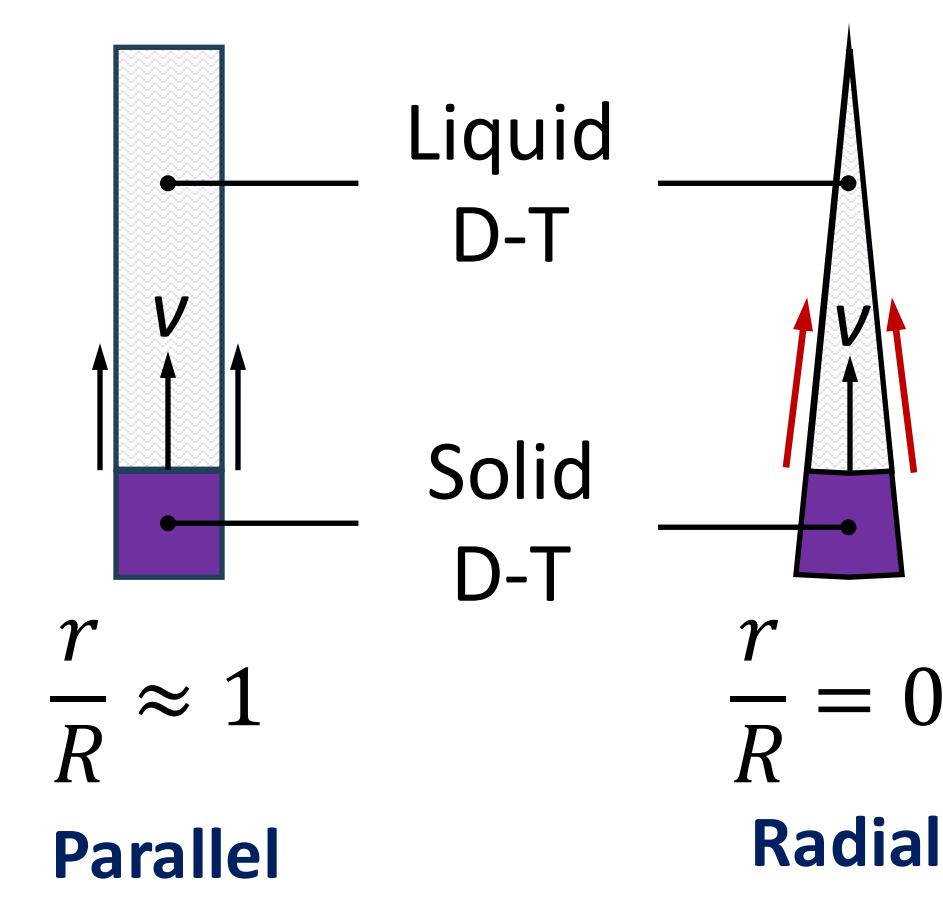
Simulation results show good agreement with the theoretical values confirming the validity of the numerical model

SIMULATION 2: Compositional distribution during radial solidification

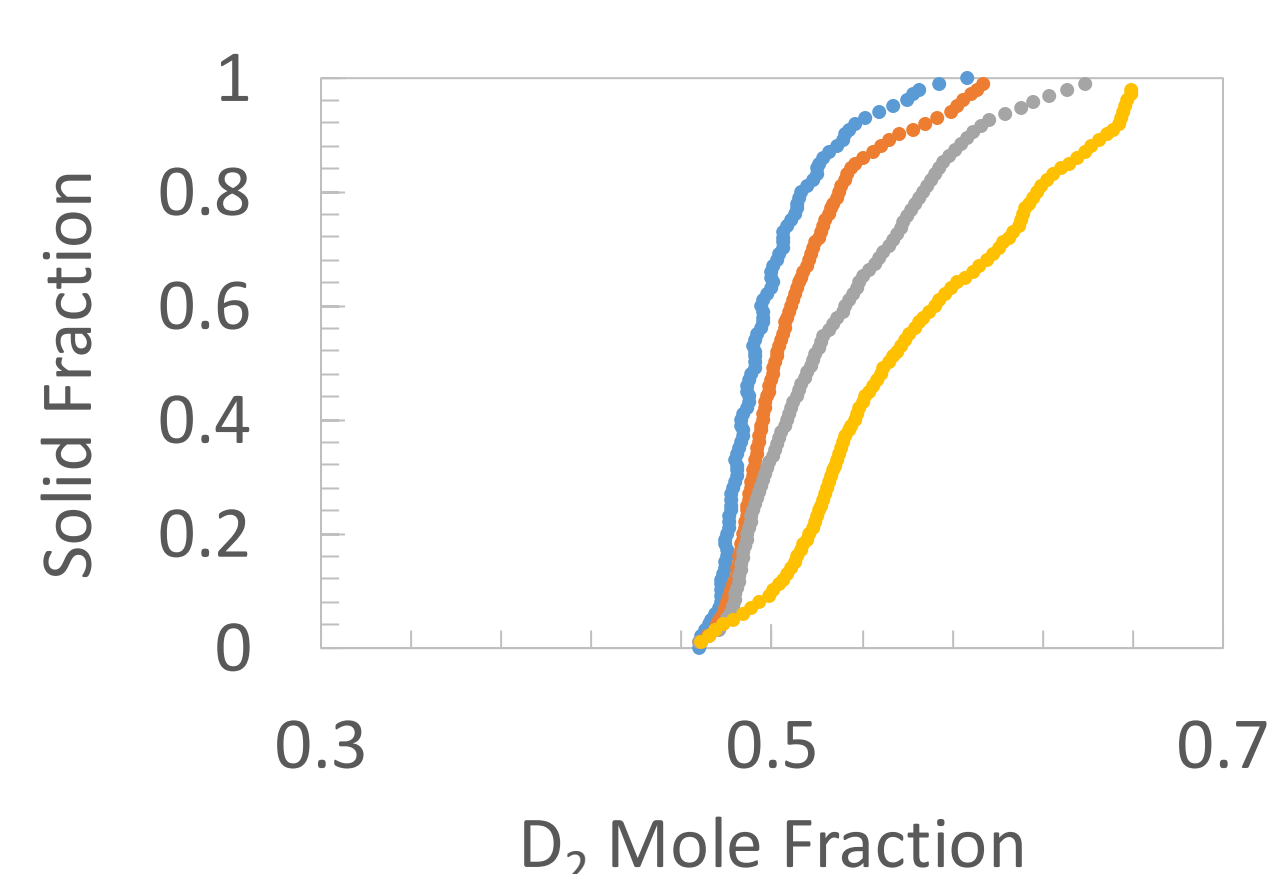
Models with different r/R



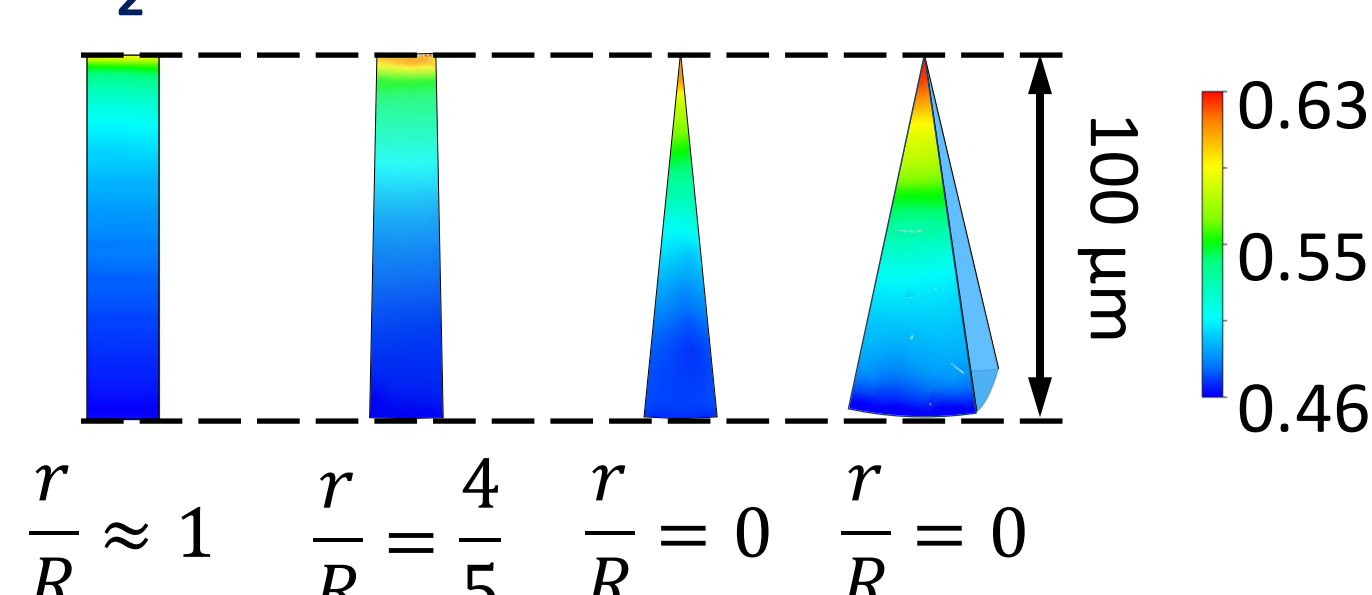
Solidification Direction



Radial distribution



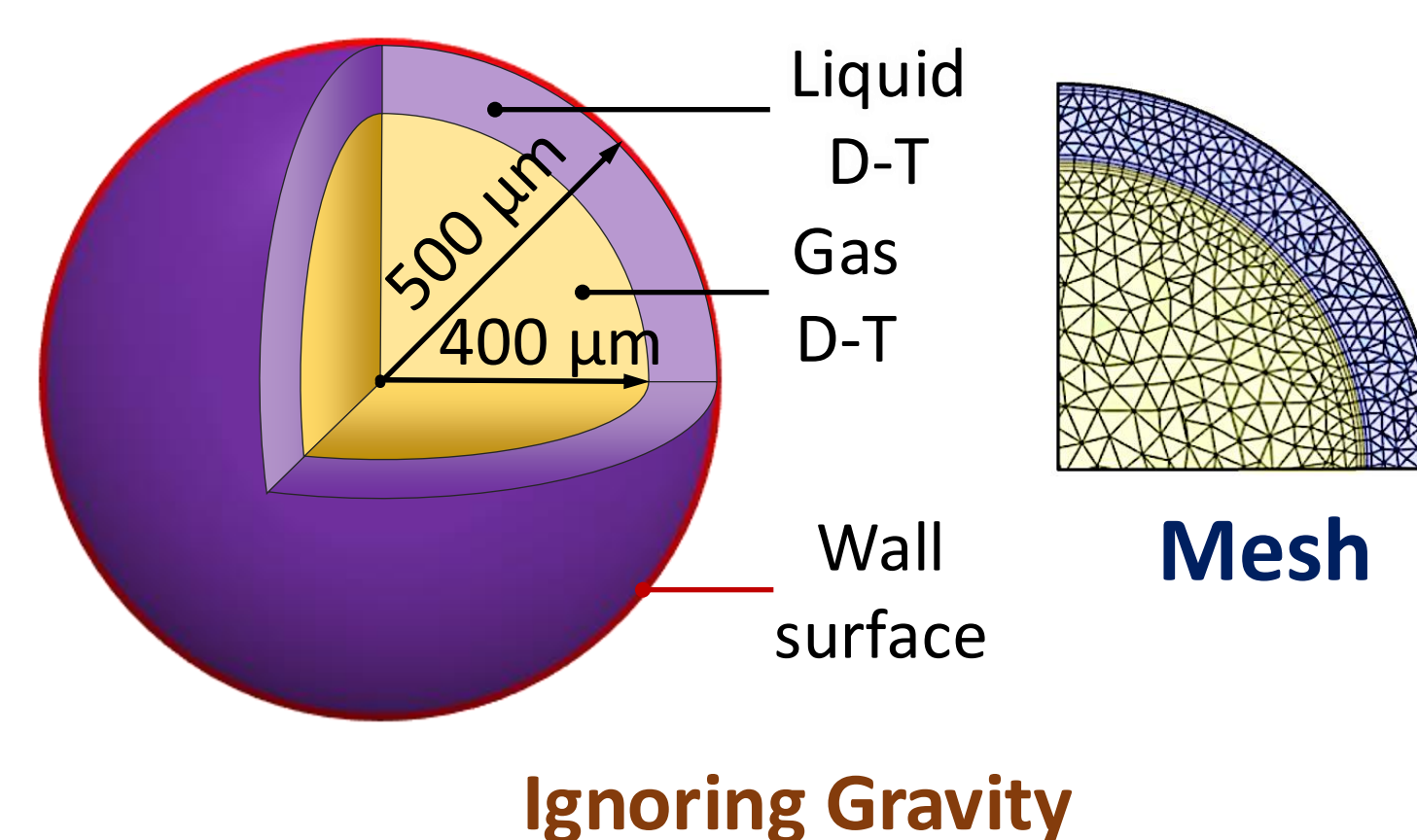
D2 mole fraction distribution



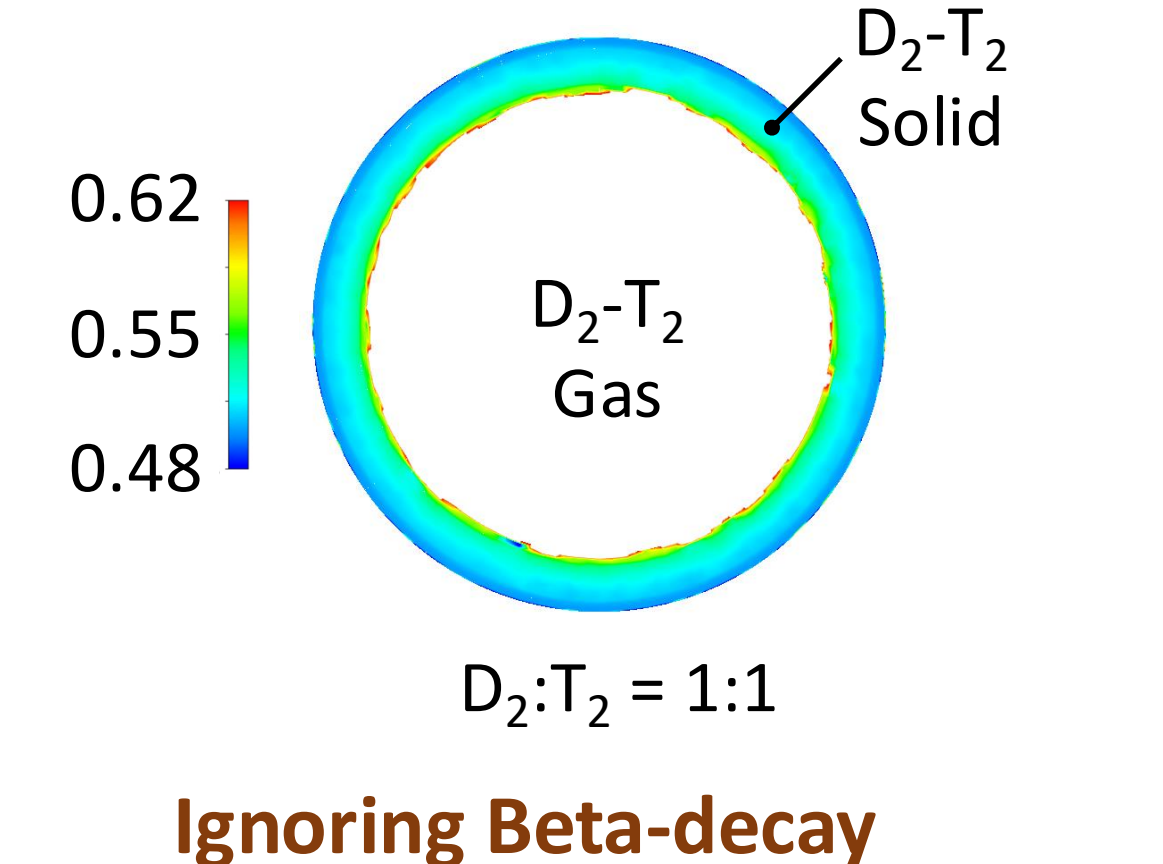
Relatively high r/R is recommended in target design

SIMULATION 3: Compositional distribution in D-T target

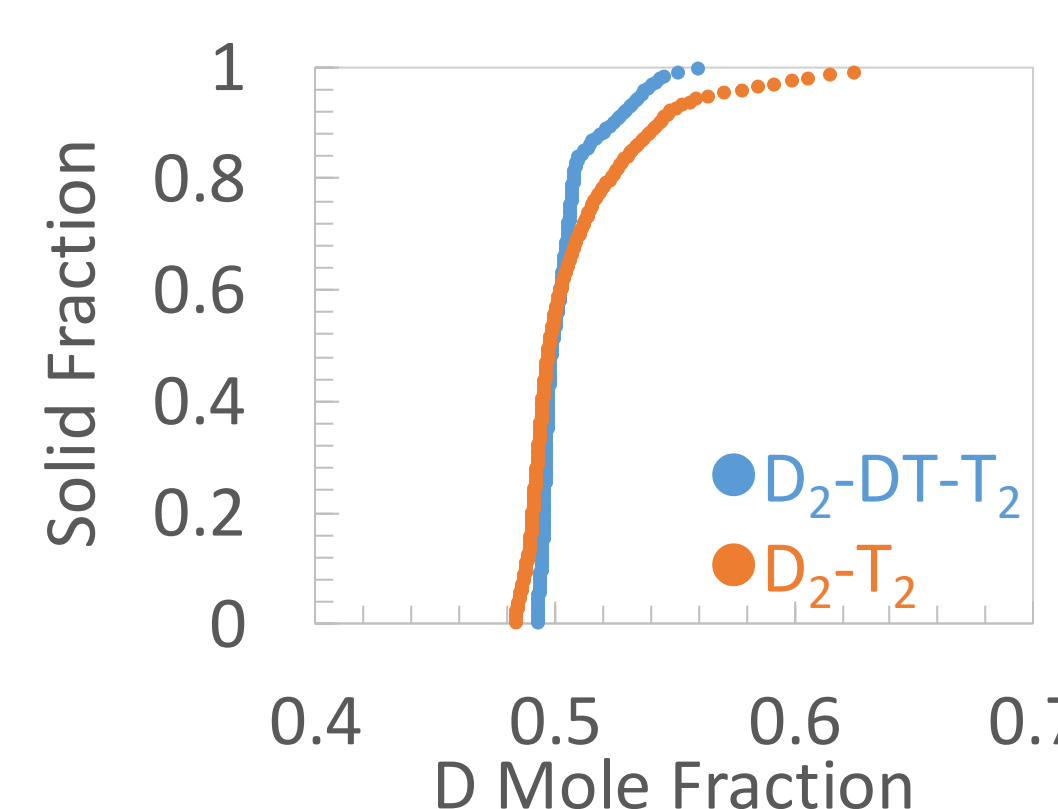
3D model of the D-T target



D2 mole fraction distribution



Radial distribution



Empirical Formulas for P_i^S (Pa) vs. T (K) [2]

Raoult's law: $P_i^G = P_i^S X_i^S$

$X_i^G = \frac{P_i^G}{\sum_i P_i^G}$

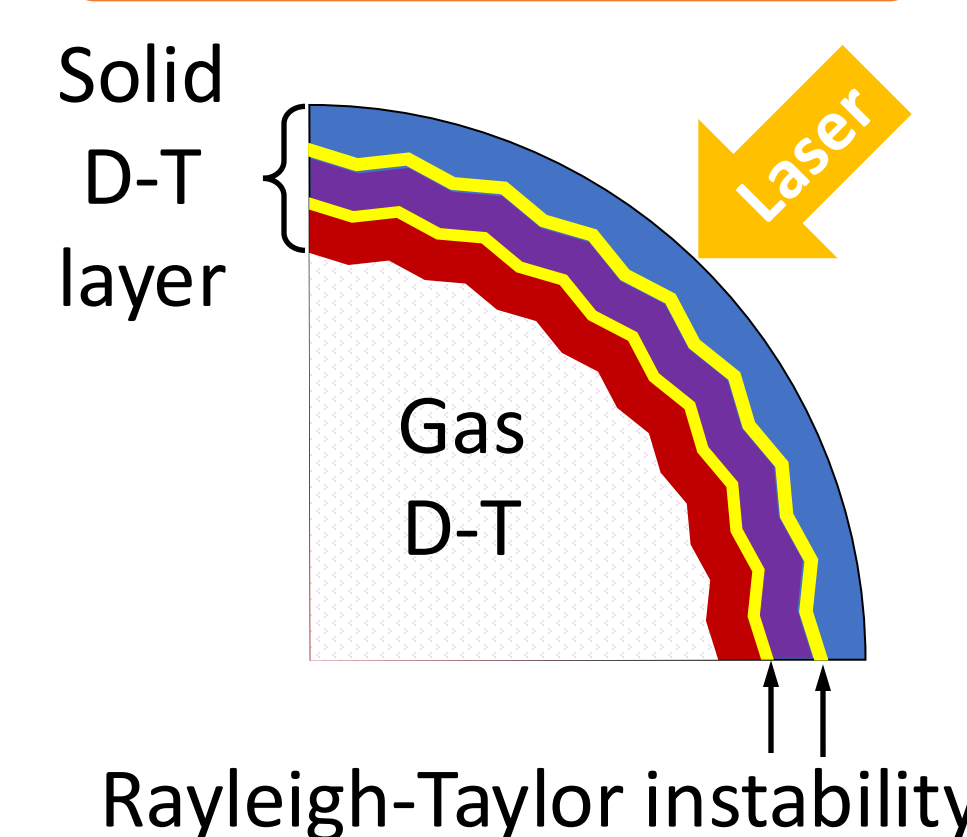
P_i^S : Equilibrium vapor pressure of the pure component i
 P_i^G : Partial pressure of the component i
 X_i^G : Mole fraction of the component i

	D ₂ :T ₂ = 1:1	D ₂ :DT:T ₂ = 3:4:3 [2]
Radial distribution of D atom content in solid	0.480 ~ 0.600	0.493 ~ 0.571
D atom content in gas	75.6%	67.9%

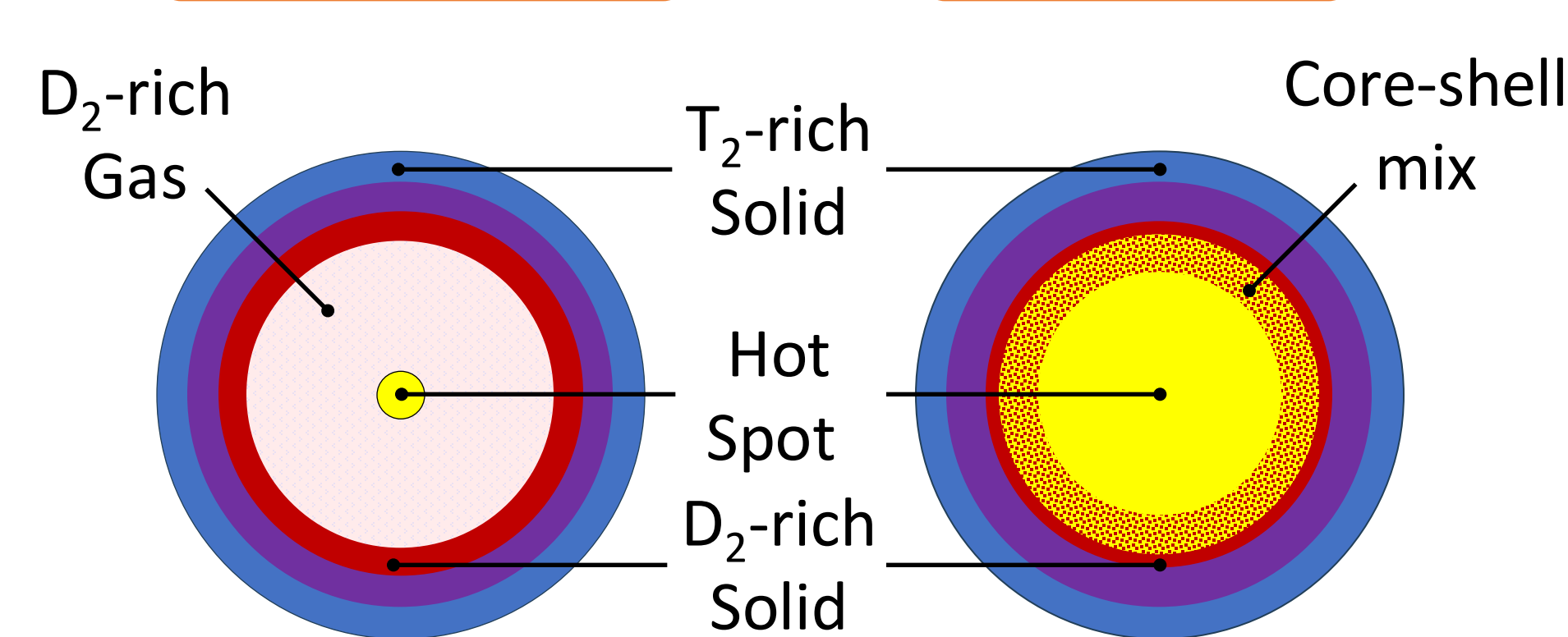
**D₂ distribution in the solid layer exhibits a radial mole fraction gradient
D atom content in the gas phase is significantly in excess**

DISCUSSION

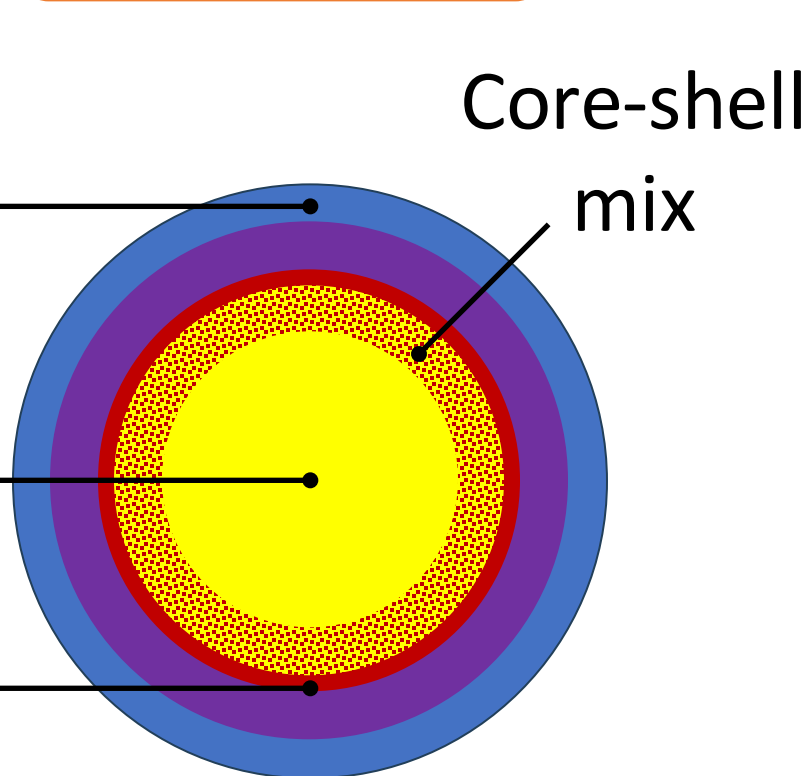
Compression Phase



Ignition Phase



Burn Phase



Rayleigh-Taylor instability:

- 1) Radial density gradient
- 2) Irregularities along isodensity surfaces

Rayleigh-Taylor instability can threat to implosion uniformity

D-T reaction: $D + T \rightarrow {}^4\text{He} + n$

Optimal Ratio: D:T = 1:1
T₂-rich areas: D:T < 1:1
D₂-rich areas: D:T > 1:1

D- or T- enriched areas may reduce fusion reaction efficiency

CONCLUSION

- Simulation results agree well with the theoretical values, confirming the validity of the numerical model
- Relatively high r/R is recommended in target design
- D atom distribution in the solid exhibits a radial mole fraction gradient; D atom content in the gas phase is significantly in excess
- Inhomogeneity in target may affect the ICF performance in multiple stages

These findings offers guidance for predicting and improving compositional homogeneity in target design

ACKNOWLEDGEMENTS

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REFERENCES

- [1] J. Zhang et al. Thermofluid simulation of hydrogen isotopologue mixtures during the solidification process. Fusion Eng. Des. 212, 114827 (2025).
- [2] P. C. Souers, Cryogenic hydrogen data pertinent to magnetic fusion energy. (California Univ., Livermore (USA). Lawrence Livermore Lab. 1979).