







The water cooled performance for fission and fusion reactors

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Important unique properties of water are as follows:

- 1. Molecular Structure and Properties
- 2. Isotopic Content
- 3. Latent Heat and Specific Heat
- 4. Density Relationships
- 5. Viscosity-Density Relationships
- 6. Surface Tension
- 7. Transparency
- 8. Pressure
- 9. Salinity
- 10. Water Current
- 11. As Universal Solvent

# **Thermodynamic Properties of Water and Steam**



#### **FIGURE A-9**

T-s diagram for water.

Copyright @ 1984. From NBS/NRC Steam Tables/1 by Lester Haar, John S. Gallagher, and George S. Kell. Reproduced by permission of Routledge/Taylor & Francis Books, Inc.

### TABLE

Temp. rature °C	Pressure Pa	Density kg/m <sup>3</sup>	Enthalpy kJ/kg	Heat capacity kJ/(kg·K)	Specific heat kJ/kg	Thermal conductivity 10 <sup>3</sup> W/(m·K)	Dynamic viscosity 10 <sup>6</sup> (Pa·s)	Kinematic viscosity 10 <sup>6</sup> (m <sup>2</sup> /s)	Thermal diffusivity $10^{6}(m^{2}/s)$	Prandtl number
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Basic Design of a PWR - Overview OF THE Three Main Circuits



# WWER-1000







PWR NPP thermodynamic and heat balance analysis.



# Typical Fuel Assembly in LWR Reactors





## Fuel assemblies vary with manufacturer and over time

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# PWR

# Boric Acid Corrosion of Carbon Steel Reactor Pressure Boundary Components in PWR Plants

## Corrosion and deposition on the secondary circuit of steam generators steam putlet





# general corrosion, stress corrosion cracking, pitting corrosion, stainless steels, nickel base alloys safety, security, reliability,

# PWR

pH control curves: (a) modified chemistry regime and (b) elevated pHT regime at 307 °C [ 5].





## **Zinc Water Chemistry**



# PWR

# Table pH control PWR plant is recommended for boron-free operation

**Ammonia-Based Water** 



Table pH control PWR plant is recommended for boron-free operation

Chemistry parameter	Normal range
pH at 286.5 ℃	6.9 - 7.4
pH at 25 °C	9.6 - 10.1
Potassium, ppm	1.7 – 5.0
Ammonia, ppm	≤2.0
Dissolved Hydrogen, cc/kg H2O	25 - 50
Dissolved Oxygen, ppm	≤0.005

### **Potassium-Based Water Chemistry**



### **Lithium Water Chemistry**

### PWR as change of primary coolant chemistry for long-term fuel cycle



Typically, 12-month fuel cycles begin with no more than 1200 ppm boron at a start of a cycle, so the maximum of 2.2 ppm lithium is required to satisfy the requirement of pH 6.9 (Fig.). Long-term fuel cyclesof 18 or 24 months have forced chemistry personnel to select an operating pH regimethat minimizes "negative effects" rather than maximizes "the benefits"

# **PWR** Neutron Activation of Coolant Water

Radiolysis is the dissociation of molecules by ionizing radiation. It is the cleavage of one or several chemical bonds resulting from exposure to high-energy flux.

Isotope	Natural Abundance[%]	Reaction	Activation product	t <sub>1/2</sub> [s]	Decay products	m
<sup>16</sup> O	99.76	(n , p)	<sup>16</sup> N	7.13	2.742 MeV gamma (1%) 6.129 MeV gamma (67%) 7.115 MeV gamma (5%)	H <sub>2</sub> O
<sup>17</sup> O	0.04	(n , p)	<sup>17</sup> N	4.14	0.383 MeV neutron (35%) 0.884 MeV neutron (1%) 1.171 MeV neutron (53%) 1.700MevV neutron (7%) 0.110 MeV neutron (3%)	H' + OH $H_2O_1$ $H_2O_2$ $H_2O_2$ $H_2O_2$ $H_2O_2$ $H_2O_1$ $H_2O_2$ $H_2O_1$ $H_2O_2$ $H_2O_1$ $H_2O_2$ $H_2O_1$ $H_2O_1$ $H_2O_2$ $H_2O_1$ $H_2O_2$ $H_2O_1$ $H_2O_2$ $H_2O_1$ $H_2O_2$ $H_2O_1$ $H_2O_2$ $H_2O_1$ $H_2O_2$ $H_2O_1$ $H_2O_2$ $H_2O_2$ $H_2O_1$ $H_2O_2$ $H_2O_1$ $H_2O_2$ $H_2O_1$ $H_2O_2$ $H_2O_$
<sup>18</sup> O	0.04	$(n,\gamma)$	<sup>19</sup> O	26.9	0.197 MeV gamma (63%) 1.357 MeV gamma (33%) 1.444 MeV gamma (3%)	Radiolytical decomposition of water

# Table - Summarized data of activated isotopes of cooling waterobtained from ENDF/B-VII.1 data library

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# 17N and gamma dose field due to decay of isotope 190

# Evolution of ACR Physics from CANDU 6

#### Major Differences between CANDU 6 and ACR

- Coolant
  - CANDU 6 (D<sub>2</sub>O)
  - ACR (H<sub>2</sub>O)
- Fuel
  - CANDU 6 (NU in 37-element bundle)
  - ACR (2.0 % SEU in 42 pins, Central Pin Dy/NU, CANFLEX bundle)



### **Advanced Power Reactor 1400 MWe**

# Korea Electric Power Corporation (KEPCO) and Korea Hydro & Nuclear Power Co., Ltd. (KHNP)



# **Advanced Passive**



AP1000 MWe



# Passive Safety Systems

### Japan's DEMO

# **Cooling Water System design for fusion Tokamak Reactor**



Component	Temperature	Pressure	Thermal energy
Blanket	290–325 °C	15.5 MPa	1574 MW
Divertor (RAFM)	290–325 °C	15.5 MPa	291 MW
Divertor (Cu-alloy)	200–230 °C	5 MPa	172 MW
Back plate	200–210 °C	3 MPa	16 MW
Vacuum vessel	100–105 °C	1.1 MPa	0.043 MW

# Fig. In-vessel components image of Japan's DEMO

## ITER MCNP

# Water Radiolysis in Fusion Neutron Environments for ITER



(a)Illustration of the three buildings constituting the ITER complex.



(c)Biological dose rate Activated pipe chases and cryostat <u>uSy/hr</u>, at 106s after shutdown – example results.



(e)Total photon dose rate (µSy/h).



(b)Biological dose rate <u>uSy/hr</u>, neutrons during plasma operation – example results, level L2.



(d)Total photon dose rate (Sy/h), during plasma operation.



(f)Integral dose to silicon from 54 cask transfers (from three port cell locations), in units of Gy.

Fig. MCNP Calculation results for detailed radiation maps of the ITER complex have been produced during plasma operation and after shutdown to the required dose rate cut-off criteria



### SUMMARY

Optimization of Water Coolant Chemistry to Ensure Reliable Water Reactor Fuel Performance at High Burnup and in Ageing Plant, it was necessary to control the chemical water by added suitable chemical elements, to avoid corrosion problems and also to properly adjust the normal operation and shutdown the nuclear power plant, and this has already been implemented for decades and it is a proven technology (PWR, VVER, BWR, CANDU, etc.),. But in the case of fusion reactors, Still under research ,for example Japan's DEMO, it is clear in this research that Japan's DEMO use of water as a coolant is used in the same range of temperatures and pressures used In PWR.

# THANK YOU For your Attention



