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Accelerator Mass Spectrometry

AN ANALYTICAL TOOL WITH APPLICATIONS FOR A SUSTAINABLE SOCIETY

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ACCELERATORS FOR RESEARCH AND SUSTAINABLE DEVELOPMENT

From good practices towards socioeconomic impact





Accelerator Mass Spectrometry (AMS)



an Analytical Tool that Provides Applications to Enable Sustainability in our Society

Overview

- 1. Why do we need AMS? requirements and some general applications
- 2. How does AMS achieve its capability? *principles and challenges*
- 3. Sustainability in AMS technology is smaller better? specific design
- 4. Application Example 1: *Carbon cycling in the ocean*
- 5. Application Example 2: *Monitoring of Remediation of Hydrocarbon Spills*





AMS – Why was it developed?



- Emerging need for greater sensitivity than in existing techniques, initially in the Earth and Environmental Sciences and Archaeometry, later in the bio-medical and materials sciences.
- Requirement: to measure rare isotopes (stable or radioactive)
 - at very low concentrations (typically 1 part in $10^{12} 10^{15}$)
 - in small samples (0.01 to 1.0 mg) of natural materials
- From its beginnings in 1977, the number of AMS labs has spread rapidly so that ~160 are operating across the world today



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- Very low concentrations (typically 1 part in $10^{12} 10^{15}$)
- **Challenge: isobars.** At this level of sensitivity, other atoms or molecules which have the same mass can arrive at the detector.
- Solutions: work with negative ions (anions), use high energies to break up molecules (or remove isobars with selective low-energy ion-gas or ion-photon reactions)
 - Small samples (0.01 to 1.0 mg) of natural materials
- Challenge: ion beam current: rare isotopes need to be measured at the same time as more abundant ones – up to or greater than 10¹⁴ times as many.
- Solutions: use very powerful negative ion sources 10s to 100s of μA, Use molecular anions if atomic ones don't provide enough current.





Some Typical Applications of AMS



- Chronology: Measure radioactive isotopes match the half-life to the time range needed – for example:
 - o ¹⁰Be (1.5 x 10⁶ y), ¹⁴C (5730 y), ²⁶Al (7.4 x 10⁵ y),
 - ${}^{36}CI (3 \times 10^5 \text{ y}), {}^{129}I (1.6 \times 10^7 \text{ y}) \dots$
 - $\circ \ ^{236}U\,(\,2.3\,x\,\,10^7\,y),\,^{238}U\,(4.5\,x\,\,10^9\,y)$
- Tracing: following the paths of specific chemicals through complex systems: e.g. plant or human metabolism, ocean circulation, environmental carbon cycling



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Some Typical (and Some Less Usual) Applications of AMS



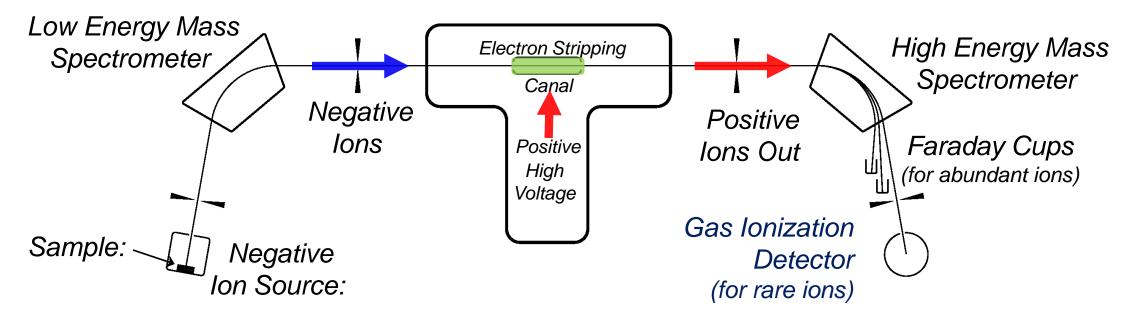
- Earth Surface Evolution: Landslide frequency and probability, earthquake frequency (¹⁴C, ¹⁰Be, ²⁶Al, ³⁶C), tectonic movement (¹²⁹I)
- Hydrocarbon Spill Remediation: tests of effectiveness of bio-remediation techniques (¹⁴C)
- Materials Characterization: analysis of radiopurity of materials for
 - o semi-conductors
 - ultra-sensitive detectors for AstroPhysical observations: e.g. Searches for Dark Matter





AMS Systems – Basic Parts

Except for Radiocarbon analyses, most AMS systems use a tandem electrostatic accelerator.



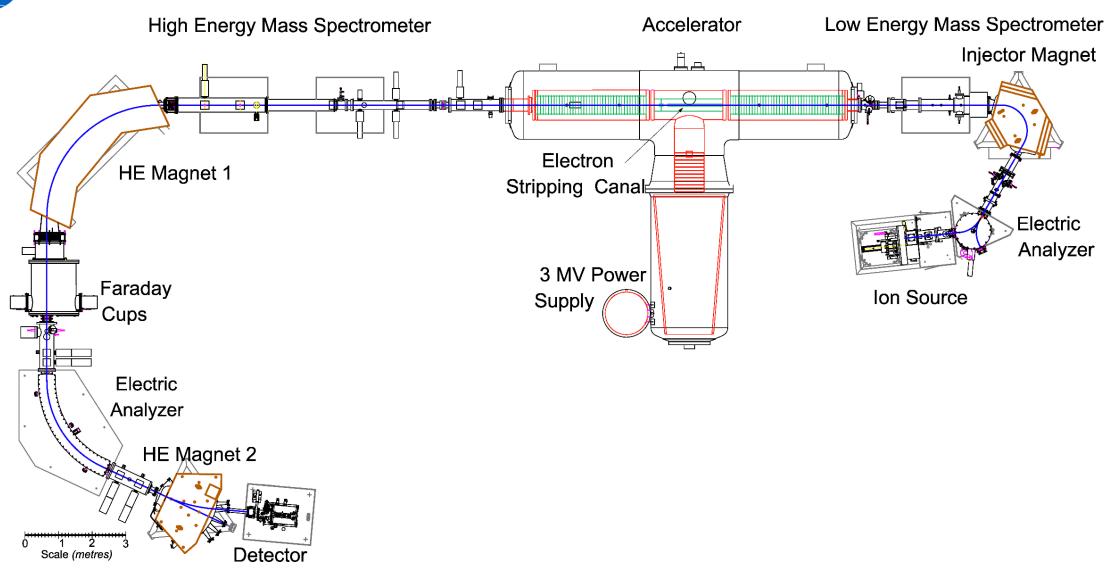
- *Electron Striping Canal* breaks up molecules by removing binding electrons.
- Gas Ionization Detector: Single atom counting ability with extremely low noise.





Of course, it's never all that simple





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Sustainability in AMS Technology



- Manufacturers are now building smaller AMS systems:
 - tandem accelerators with lower terminal voltages e.g. 200 or 300 kV
 - single ended systems (either the low energy or the high energy spectrometer on a high voltage platform)
 These have much smaller space and power requirements
- In larger systems, magnets typically consume the most power (more than the accelerator) Systems built for specific isotope analysis (e.g. ¹⁴C), can use permanent magnets with a small coil for minor adjustments
- . . . an example of a ¹⁴C system recently installed in our lab





A Single Isotope System for ¹⁴C





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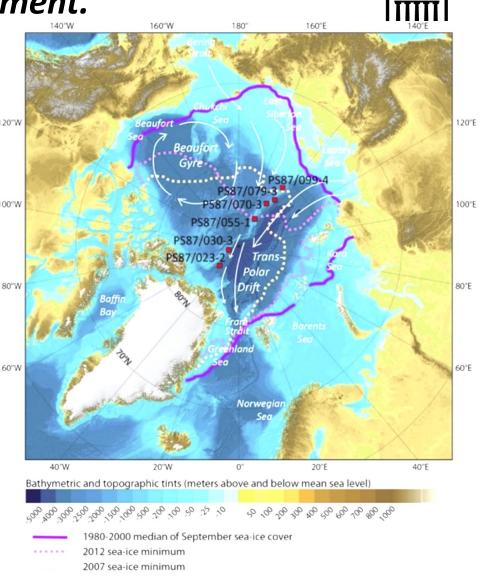
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Application 1: Carbon in the Arctic Environment.

- Arctic regions have experienced 2x the global increase in average temperatures – resulting in a "two-edged sword" for carbon storage / emission
- On-going studies of carbon in the water column (Walker, www uOttawa) and sediments (de Vernal, UQAM) already providing information about the region's response to previous climate changes.
- The small sample AMS capability will enable Compound Specific Radiocarbon Analysis (CSRA) – provides information about the origin of the carbon in the sampling locations.
- CSRA also helps to understand and model carbon released from permafrost melting in historical and current events.







Recent Arctic Sampling Cruise *





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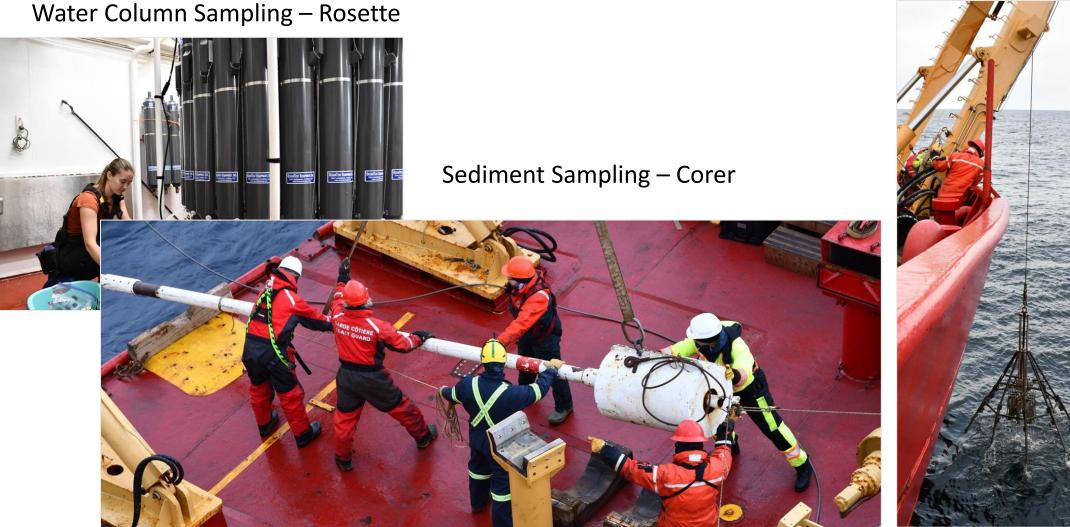
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Oceanographic Sampling Equipment (used on recen cruise)







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Application 2 – Bio-remediation of hydrocarbon spills *



- Microbes are "designed" and used to break down toxic hydrocarbons into more benign components – and release some CO₂ in the process. But are they "eating" the hydrocarbons of interest or a layer of forest fire debris?
- Soil gases can be sampled through a filter bed of Ba(OH)₂.
 CO₂ reacts with the Ba(OH)₂ and produces BaCO₃.
 The quantity of ¹⁴C in the CO₃ indicated the source of the "food" for the bacteria.
- 3. Filter cartridges are loaded with Ba(OH)₂ in the lab and are use by commercial operators to test remediation sites.
- 4. Exposed filter cartridges are returned to the AMS lab and are analyzed using an automated Carbonate Handling System connected directly to the AMS ion source

* Project by uOttawa M.Sc student Lindsay Shaw





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- For almost 45 years, AMS has provided highly sensitive isotope analyses for Earth, environmental, archaeometric, bio-medical and materials sciences.
- New AMS systems and techniques are being developed which require less energy, space and sample preparation time.
- Applications continue to be developed which are making an impact on research that is important for the sustainability of our society.



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