A combined XRF and XANES study on bottom ashes from municipal solid Waste-to-Energy plant

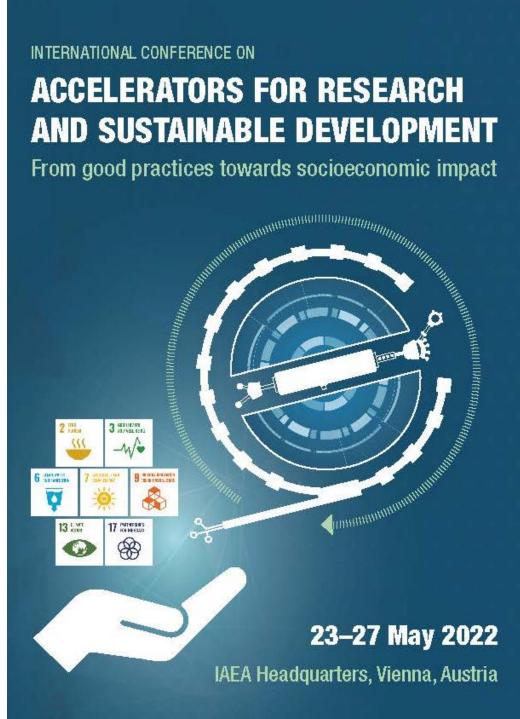
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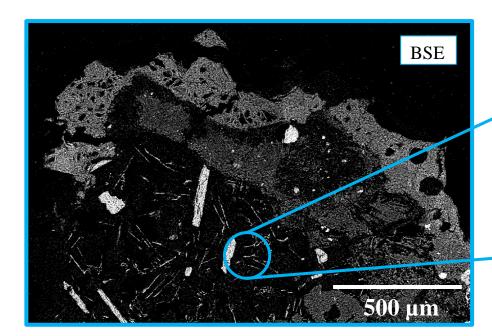
²Università di Parma, Dipartimento di Scienze Chimiche, della Vita e Sostenibilità Ambientale

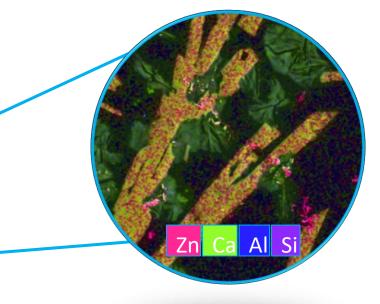
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Only in Italy, each year about 30 million tons of urban waste are produced, of which 5.3 million are disposed for combustion in incinerators and Waste to **Energy plants (WtE)**; the process then produces about one million tons of ashes. These ashes are made by bottom ashes (BA) and fly ashes (FA).



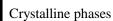






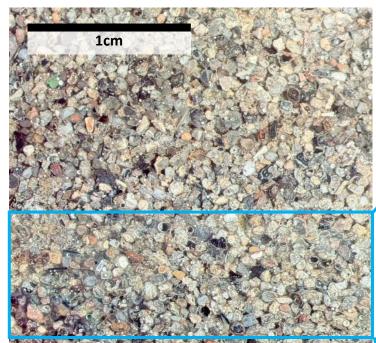
Amorphous phases





Whereas FA are classified as dangerous waste, BA can be recycled and are the main secondary raw material from combustion plants. Nevertheless, an assessment of the risk for environmental pollution and health is needed because of the content of potentially dangerous elements (PTE) such as Zn, Pb, Cu, Cr and Ni.

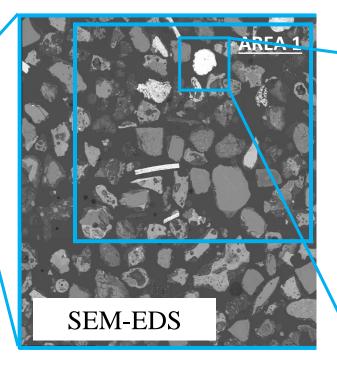
Optical microscope

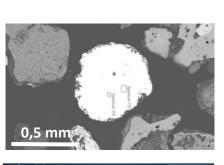


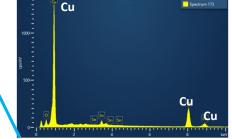
For this reason, few grains of BA (grain size 0.5 -1 mm) from a waste-to-energy plant, have been embed in epoxy resin and then polished on one side to perform SEM-EDS, XRF mapping and XANES measurements.

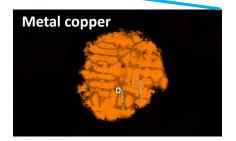
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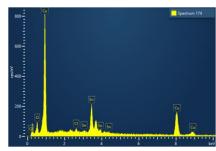
Recycling BA as a secondary raw material represent an interesting, environmentally friendly, alternative solution to landfill disposal. Nevertheless, any form of recycling requires an assessment of the potential pollution for environment and health risk.



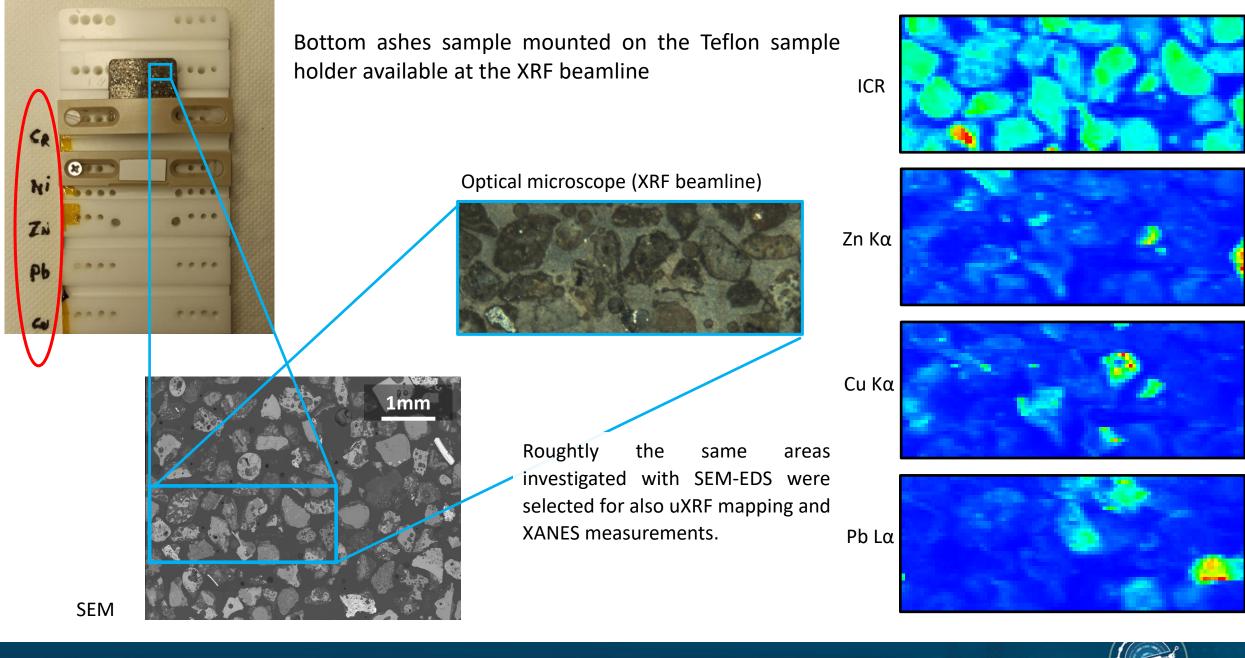


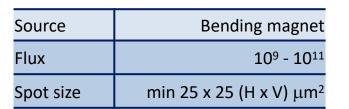






SEM-EDS ANALYSIS

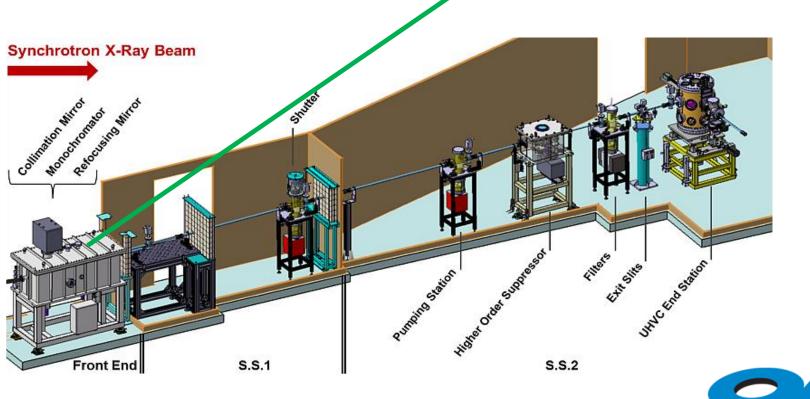




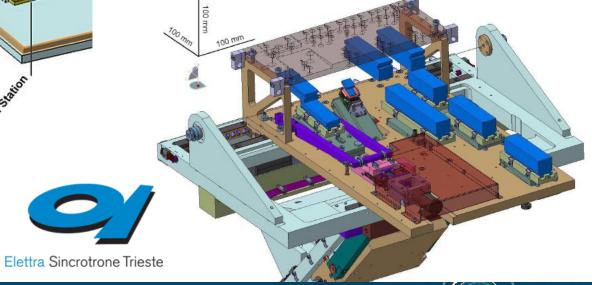
Elettra Spherical ring **Pseudo channel cut** collimation **MONOCHROMATOR** toroidal mirror refocusing mirror

XRF beamline overview





Optics type	E range (keV)	E resolution (∆E)	
Si(111)	3.6 - 14	~ 1 eV at 7 keV	
InSb(111)	2.0 – 3.8	~ 1eV at 2.2 keV	
ML: High E (RuB ₄ C)	4.0 – 14.0		
ML: Medium E (NiC)	1.5 – 8.0	~ 55 eV at 1 keV ~ 180 eV at 14 keV	
ML: Low E (RuB ₄ C)	0.7 – 1.8	100 CV Ut 14 NCV	



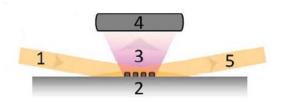
Jark et al., 2014

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Grazing angle geometries

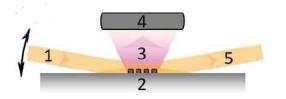
XRF beamline overview

Total reflection



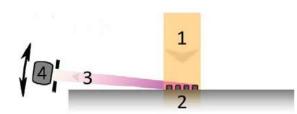
Trace element analysis
Surface contamination

Grazing incidence



Depth profiling measurements

Grazing exit



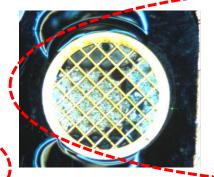
Surface mapping Depth and lateral inhomogeneity

Analytical capabilities:

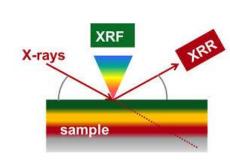
- GIXRF and GI-XANES (nano-layered structures or shallow dopants in semiconductors with nm depth resolution)
- TXRF and TXRF-XANES (elemental/chemical characterization of liquid samples residues or particulate matter samples collected directly on reflector surfaces the at the ultra-trace concentration level)
- 2D scanning XRF and XANES (study and quantification of the elemental composition or chemical speciation over different areas of the sample) standard 45/45 geometry.
- XRR (structural analysis thickness, density of thin films and multi-layered structures)

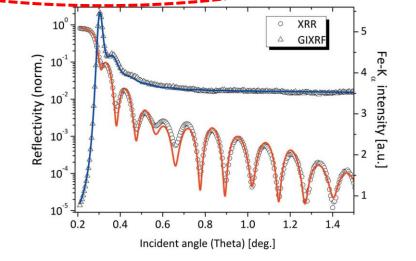
1.8 1.6 1.4 1.2 1 1 0.8 Appl. Phys. Lett. 106, 093108 (2015) 0.4 0.2 0.2 9550 9600 9650 9700 9750 9800 9850 9900 9950 1000

XRF beamline overview

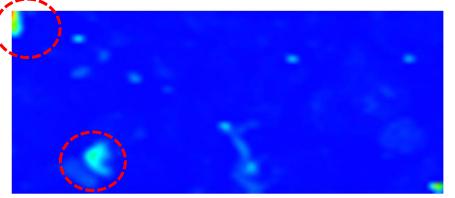


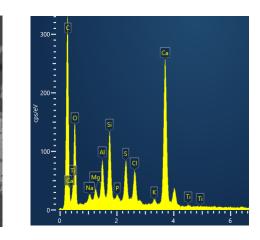


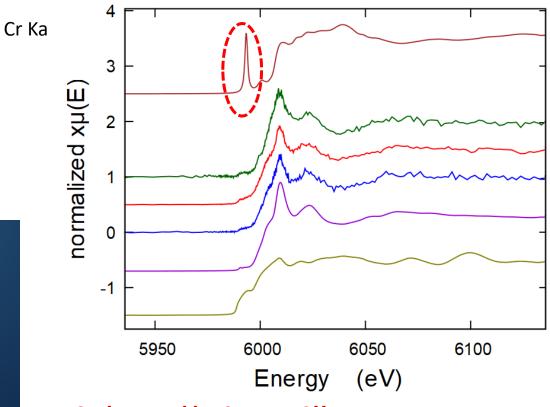


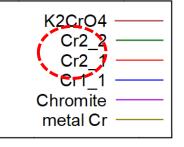


A preliminary look to the XANES data collected at the Cr K-edge permits to exclude the presence of Cr⁶⁺, as all the collected spectra have features resembling that of Chromite (Cr³⁺).

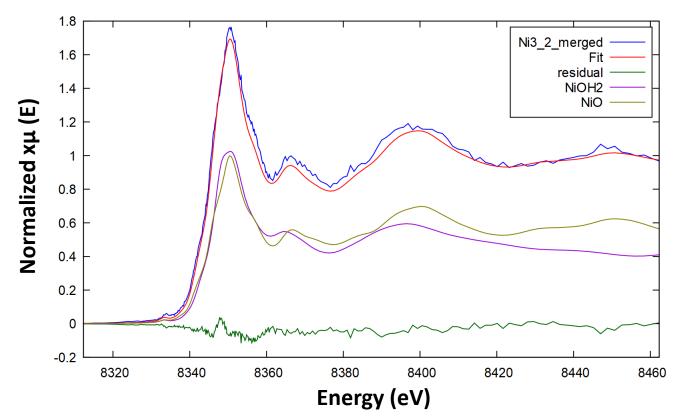








No Cr detected by SEM-EDS!!



Example of a fit obtained through LCF analysis on a Ni spectrum.

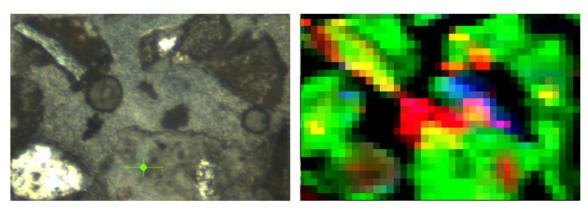
nickel

Concerning Ni, the oxidation state is considerably changing between different clasts, ranging from mainly metallic to divalent.

Results of the LCF analysis

Sample	Metal Ni Ni ⁰	NiO* Ni ²⁺	Ni(OH) ₂ * Ni ²⁺	Sum	
Ni1_1	-	17 (10)	83 (10)	100	
Ni1_2	-	40 (16)	60 (12)	100	
Ni1_3	35 (3)	13 (10)	52 (14)	100	
Ni2_1	66 (2)	34 (16)	-	100	
Ni2_2	-	-	100	100	
Ni2_3	28 (11)	49 (9)	23 (8)	100	
Ni3_1	84 (5)	16 (20)	-	100	
Ni3_2	-	67 (7)	33 (4)	100	

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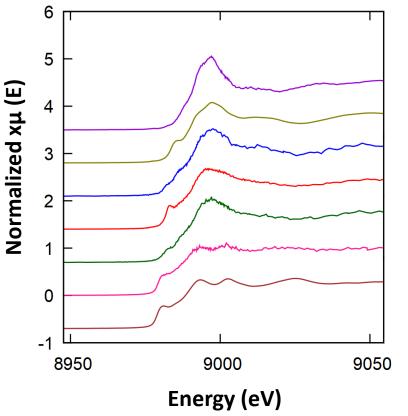
The clast named Cu3_4 from micro camera (on the left) and the corresponding XRF map of Cu (red), Zn (blue) and Ca (green) highlighting its heterogenicity (right).

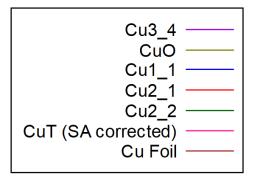
Results of the LCF analysis

Sample	Metal Cu	Cu ₂ O Cu ¹⁺	CuO Cu ²⁺	Sum
Cu1_1	-	6 (4)	94 (2)	100
Cu2_1	10 (4)	24 (5)	66 (3)	100
Cu2_2	-	13 (4)	87 (3)	100
CuT	80 (3)	16 (4)	4 (2)	100

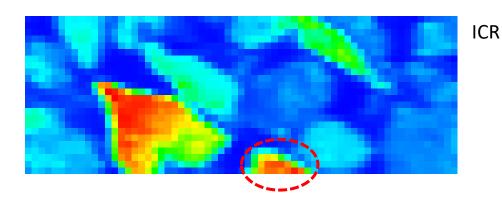
copper

Cu appears to be present both in metal and oxidized form. The precise oxidation state and coordination geometry of all the investigated chemical elements have been determined through linear combination fit (LCF) analyses.





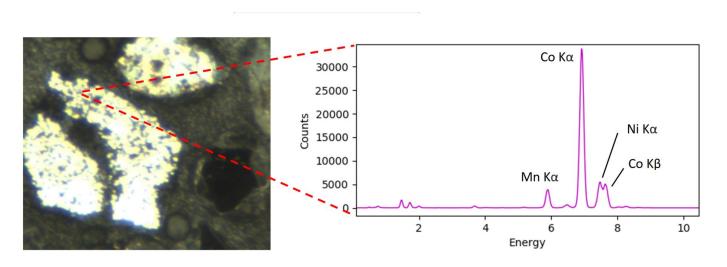
Comparison between normalized Cu K-edge XANES spectra collected on different clasts and Cu-bearing reference compounds.



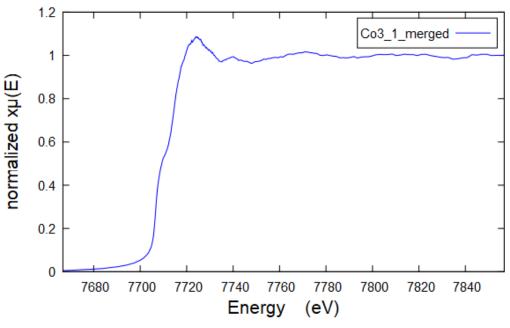
cobalt

Only one clast have been investigated by XANES at Co K-edge. XRF mapping highlight that is composed mainly of Mn, Co and Ni. LCF result give a composition of 68% (7) metal Co and 32% (4) CoO.

On the same clast, also Ni K-edge XANES spectrum was collected (labeled as Ni3 1) and LCF result is 84% (5) metal Ni and 16% (20) NiO. Hence, according to the chemical composition and XANES results, it might be a MnCoNi alloy.

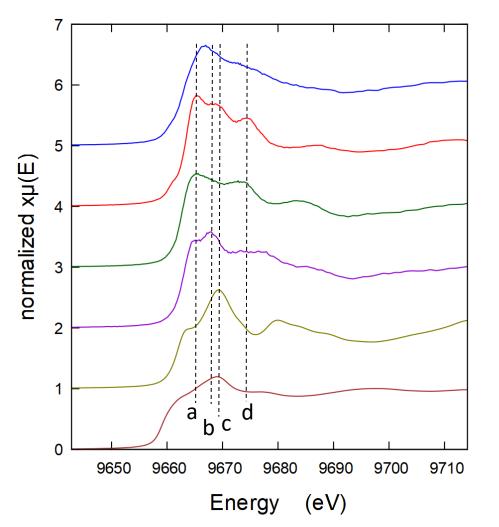


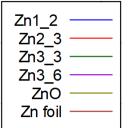
The photo on the left has been taken with the micro camera available at XRF beamline on the clast named Co3 1. On the right, the XRF spectrum of the clast.



Normalized Co K-edge XANES spectrum collected on clast 3 1.

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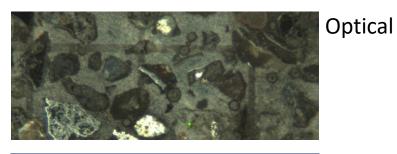


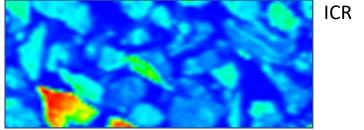


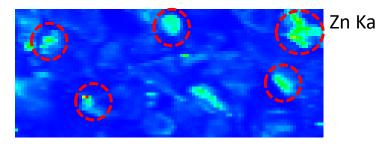
Zinc

Zinc appeared to be the more abundant among PTE. In none of the collected data, neither a small amount of metal Zn have been detected. Indeed, the edge position of all the spectra is similar and shifted with respect to reference Zn foil.

Preliminary LCF results point to the presence of only Zn^{2+} .

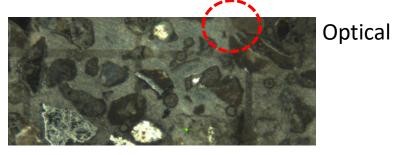


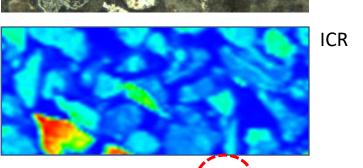


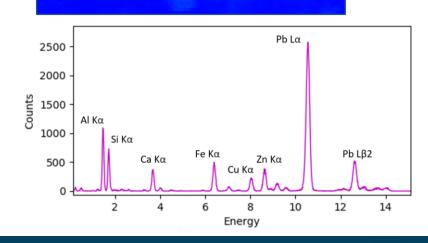


Comparison between normalized Zn K-edge XANES spectra of clasts (selected spectra are representative of the clast variability) with ZnO and Zn foil references. Reference lines labeled a, b and c correspond to the following energy positions: a) 9665,3 eV b) 9667,9 eV; c) 9669,5 eV; d) 9674,1 eV.

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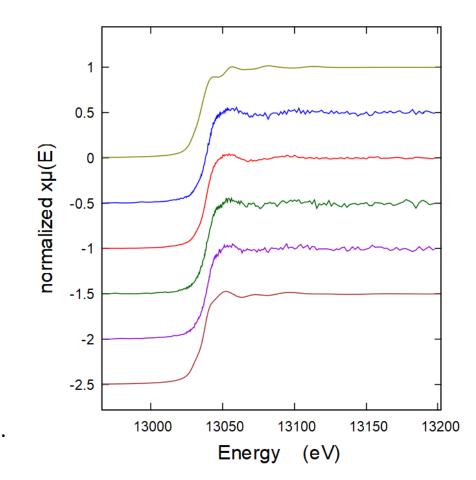


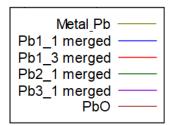




<u>lead</u>

All the XANES data collected at Pb L_3 -edge on various clasts are almost identical. A fitting approach through LCF analyses was used as for previous data but the results were not satisfactory. PbS and PbSO₄ reference spectra were also available but not used as S fluorescence lines were not detected.





full XRF spectrum of the clast Pb3_1.

Pb La

Final remarks

vsis currently in progress! ✓ XRF mapping demonstrated to be an excellent tool for such kind of samples, allowing elements not visible with standard SEM - EDS analyses.

✓ XANES LCF analyses permits to determine the chemical envirop be present in different oxidation states and structures

✓ Cr: Absence of Cr⁶⁺; all clasts contain only Cr³

✓ Pb: Oxidized with the same che of Pb-rich areas could point

ng different clasts. The chemical composition

✓ Cu, Ni and C

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among PTE. Seems present only as Zn²⁺ but in various structures.

Thank you

Acknowledgements

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