

Corrosion behaviours of austenitic steel and ferritic steel in LBE with oxygen controlled conditions

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2019.10

Outline

- 1** Motivations
- 2** Experiment
- 3** Experiment Results
- 4** Analysis and Discussion
- 5** Summary

Research motivations

- For the development and application of Lead-bismuth cooled fast reactor and ADS system.
- To acquire corrosion data of structural materials in lead-bismuth alloy-for materials selection.
- To illustrate the corrosion behaviours of structural materials in lead-bismuth alloy, which shows the way to conduct materials protection in a LBE system.

Experiment

● Experiment sample and facility

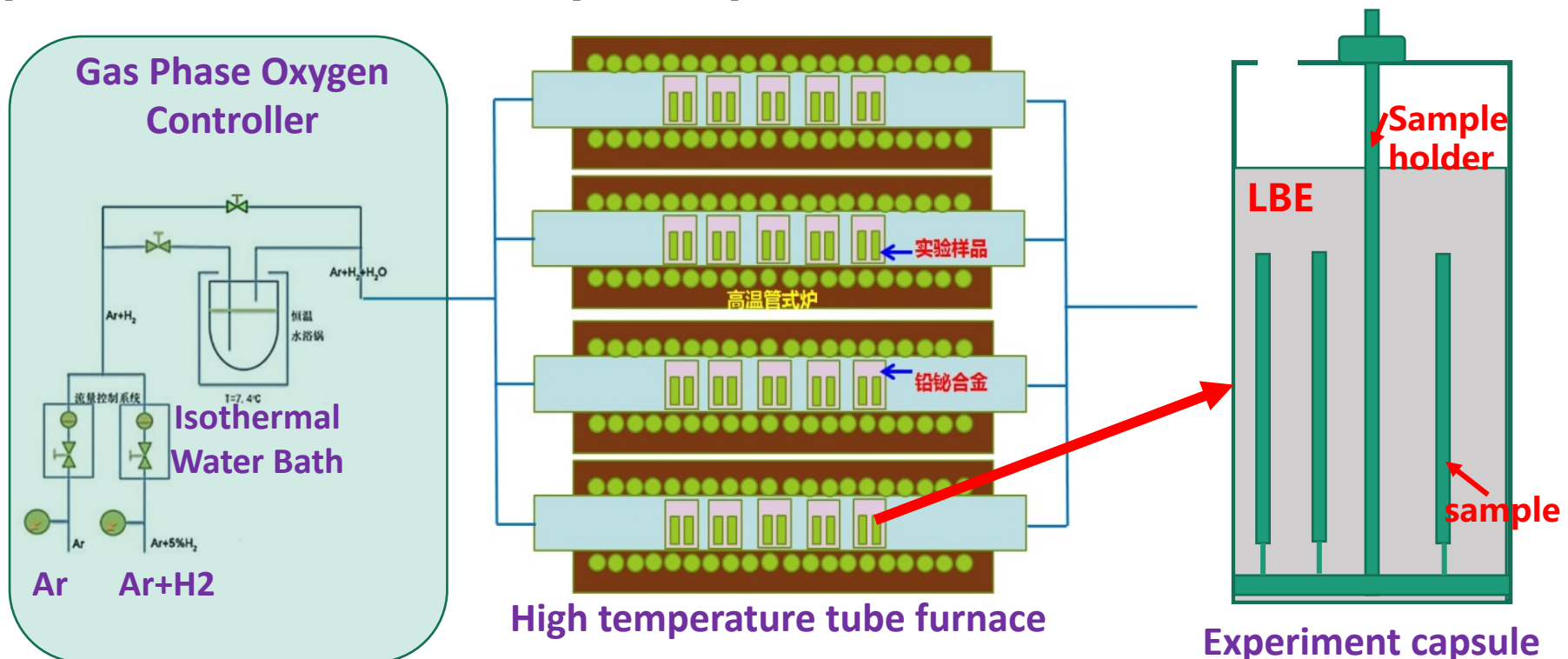
- Lead-bismuth alloy: Pb-Bi Eutectic (124°C melting point)
- Structural materials:
 - F-M steel: T91
 - Austenitic steels:
 - CN-1515: Cladding materials for CFR-600
 - 316Ti: 1st Generation cladding material for CEFR
- Sample size:
 - 20×10×2mm in three direction



Experiment

● Experiment samples and facility

- Each capsule contains three specimens and LBE about 240g, and punched holes on the top of capsules.



Experiment

- Experiment sample and facility



Experiment

● Experiment Condition

- Temperature: 450, 500, 550 and 600°C
- Time: 1000h and 3000h
- Oxygen Concentration: $10^{-6} \sim 10^{-7}$ wt.%

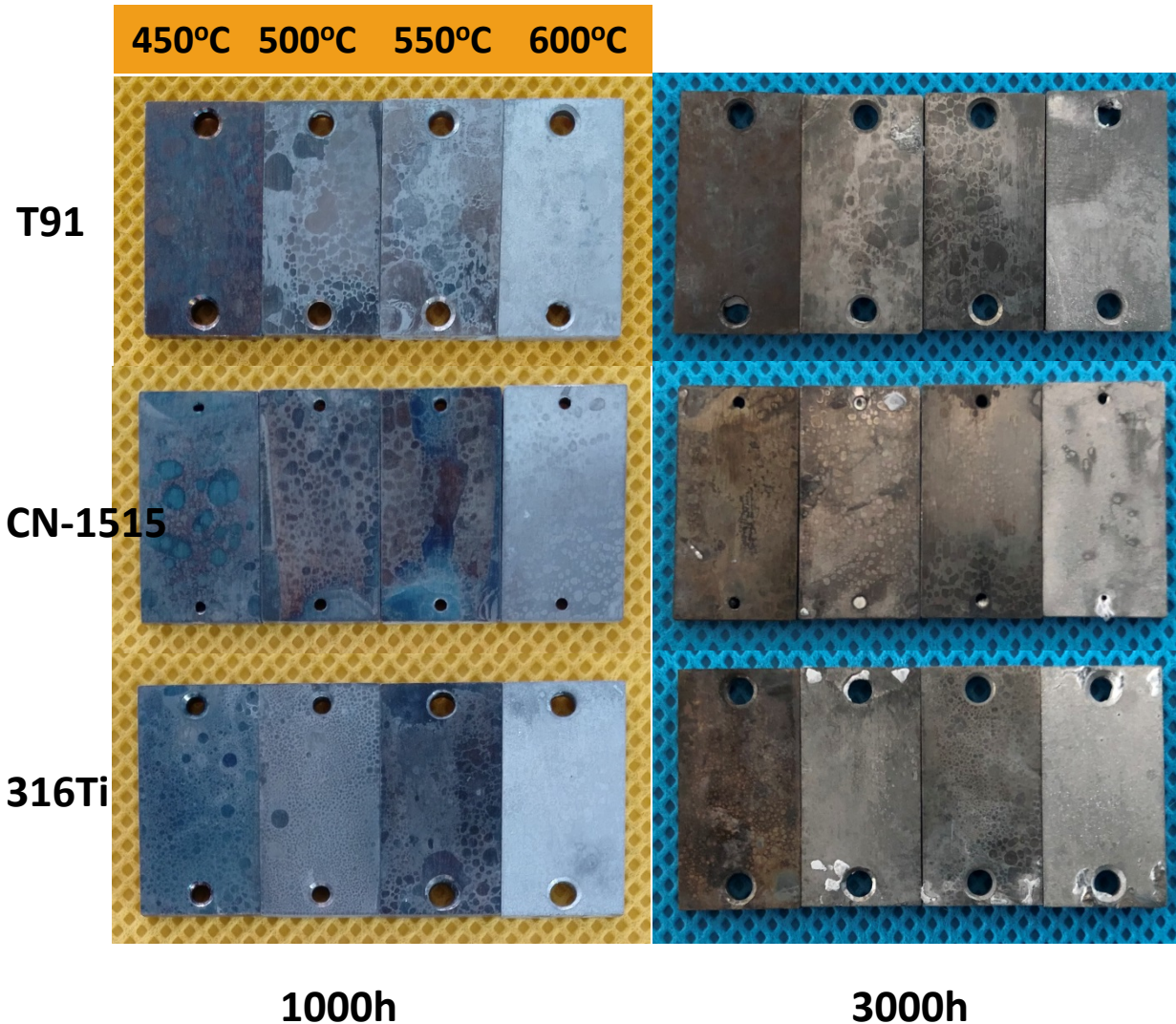
● Analysis and Characterization

- Specimen cleaning: $C_2H_5OH : H_2O_2 : CH_3COOH = 1 : 1 : 1$
- Scanning electron microscope, SEM: surface and section observation
- Energy disperse spectrometer, EDS: Surface and section composition analysis
- X-ray diffractor: corrosion phase detection

● Instruments information

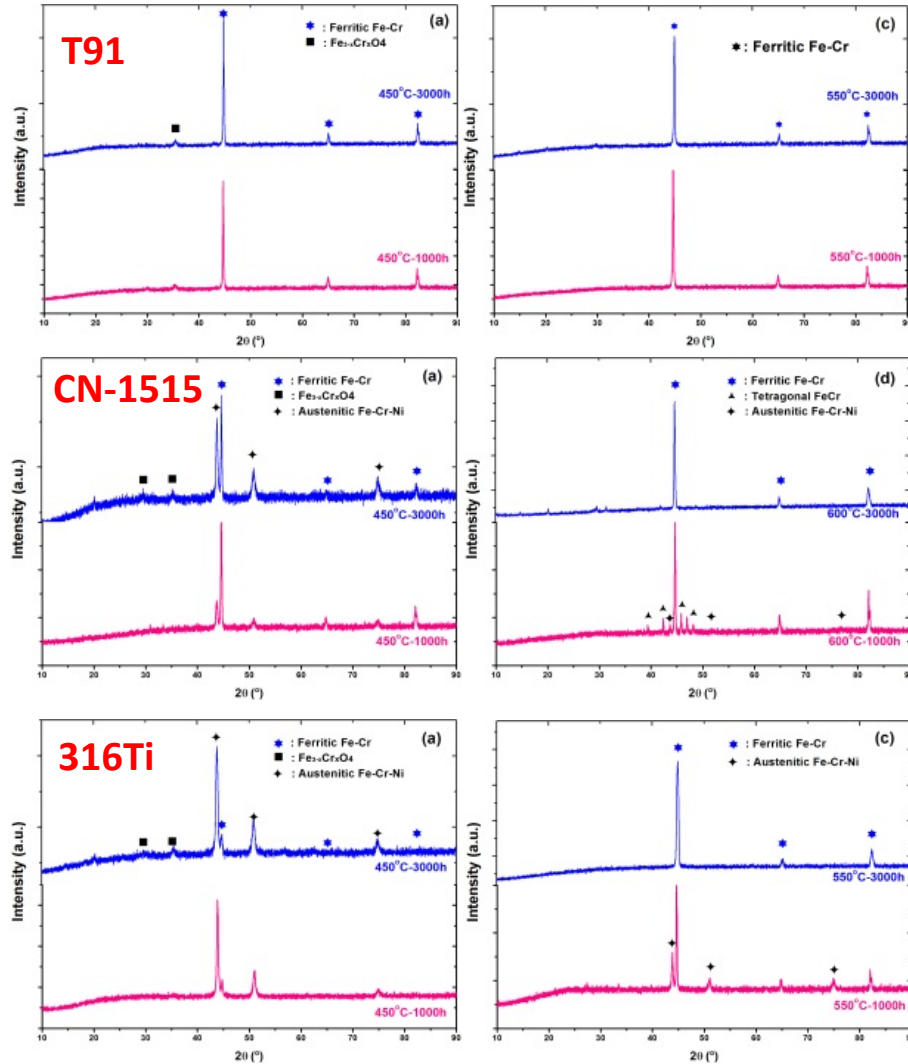
- SEM: ZEISS Supra 55 field emission
- EDS: Bruker E-flash 1510
- X-ray diffractor: Bruker D8 Advance

Macromorphology of specimen after corrosion



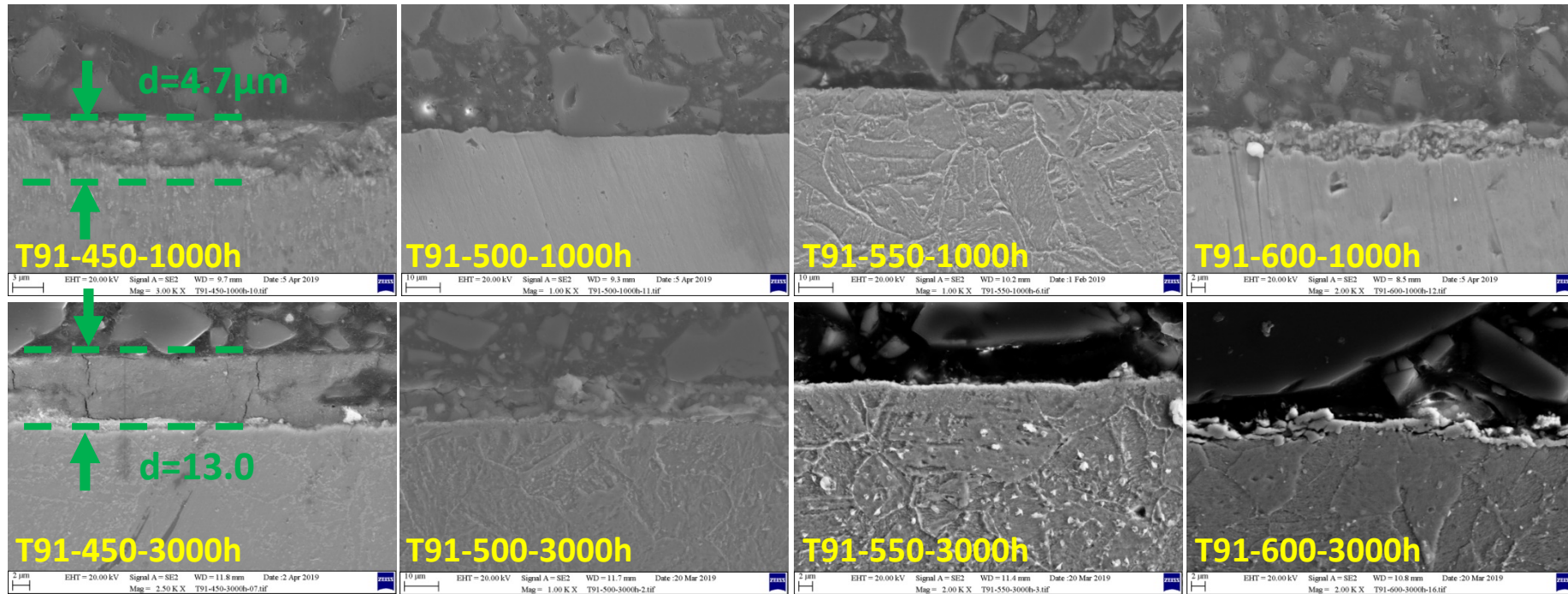
- After LBE exposure experiment, the surface colour of samples changes greatly
- The colour change means a corrosion of materials in LBE
- The difference colour of sample maybe indicates different type of corrosion.

Corrosion phase analysis (XRD)



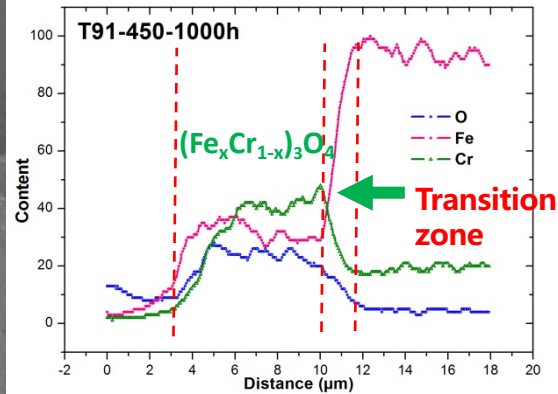
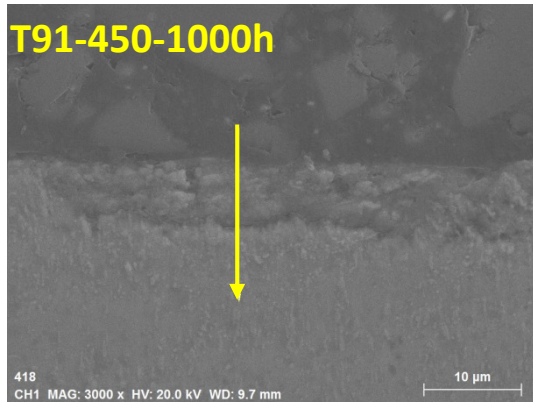
- At 450°C, XRD analysis shows the formation of oxide on the surface of three materials.
- For CN-1515, a ferritic layer is also formed.
- With the increment of temperature, no oxide formation is found for T91, and FCC structure is not stable in LBE, a FCC-BCC phase transformation happened

SEM analysis of corrosion section-T91

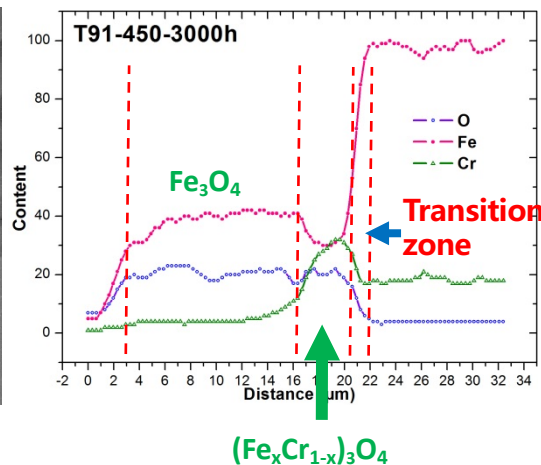
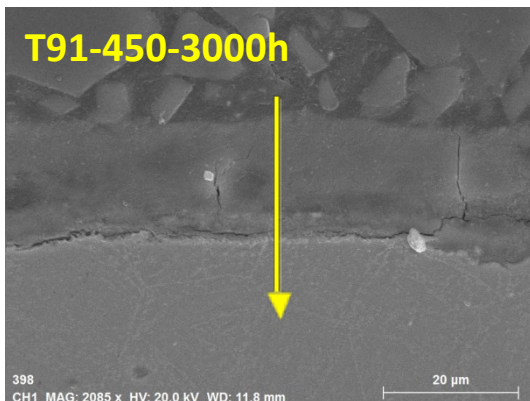


- At 450°C, oxidation corrosion takes place, which is in agreement with the XRD results ;
- At 500 and 550°C, only a slight corrosion is observed ;
- With the increase of temperature, at 600°C, a obvious dissolution corrosion is happened on the surface of T91.

T91: Corrosion product composition analysis

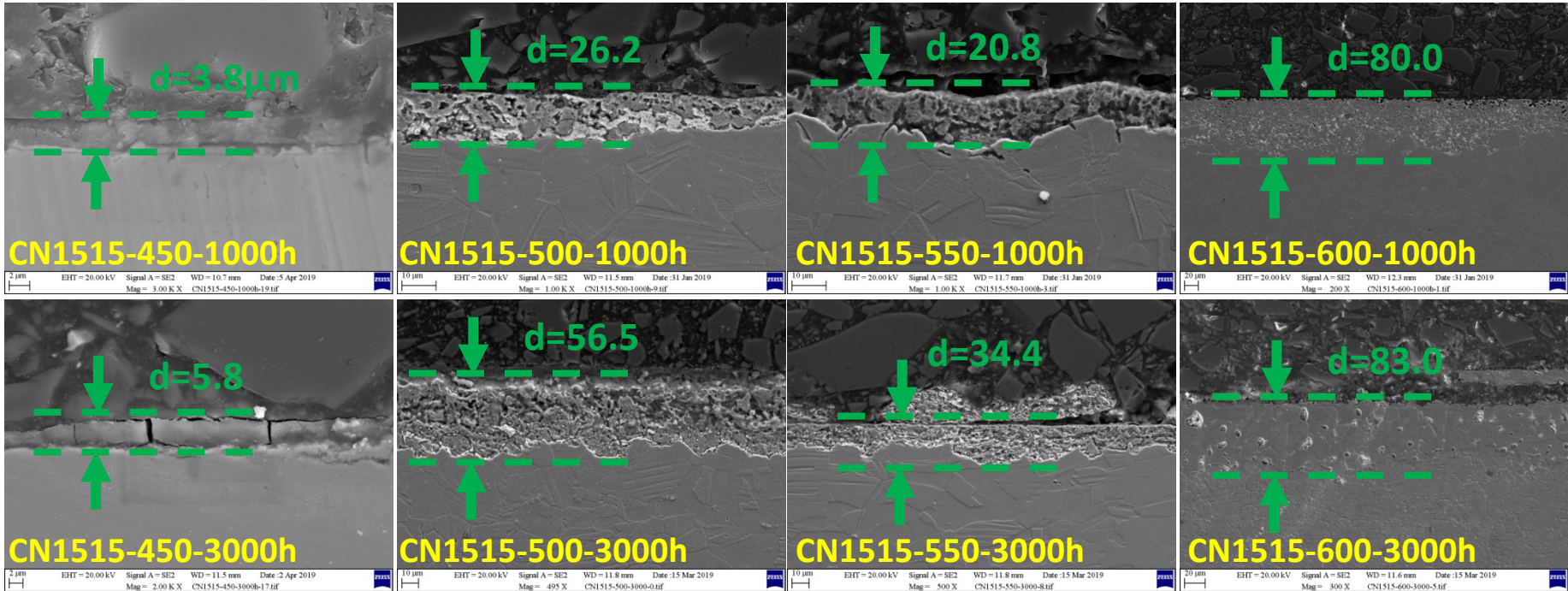


- At 450°C , EDS line scanning shows a single layer of $(\text{Fe}_x\text{Cr}_{1-x})_3\text{O}_4$ magnetite is formed after 1000h LBE corrosion.



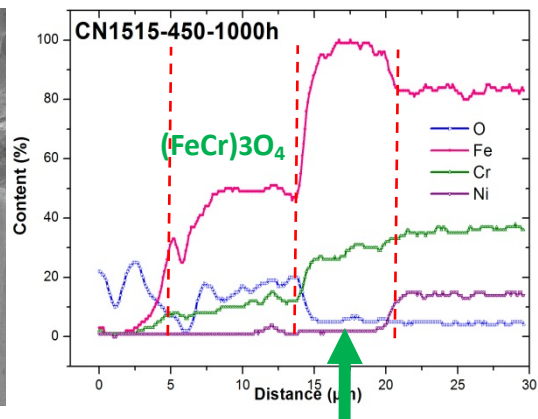
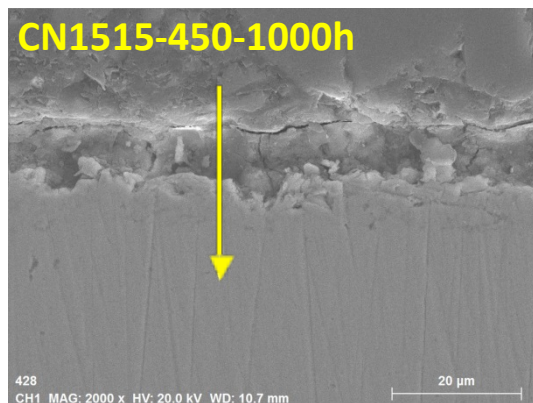
- At 450°C , after 3000h , a oxide film is composed of two layers.
- Out layer is Fe_3O_4 , and inner layer is $(\text{Fe}_x\text{Cr}_{1-x})_3\text{O}_4$, respectively.

SEM analysis of corrosion section-CN-1515

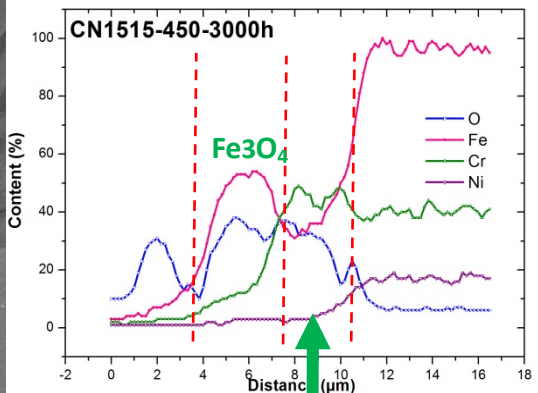
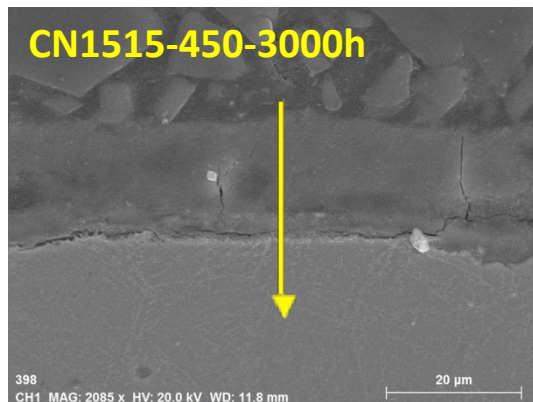


- At 450°C, oxidation corrosion takes place, which is in agreement with the XRD results ;
- At 500, 550 and 600°C, an obvious dissolution corrosion is observed ;

CN-1515: Corrosion product composition analysis

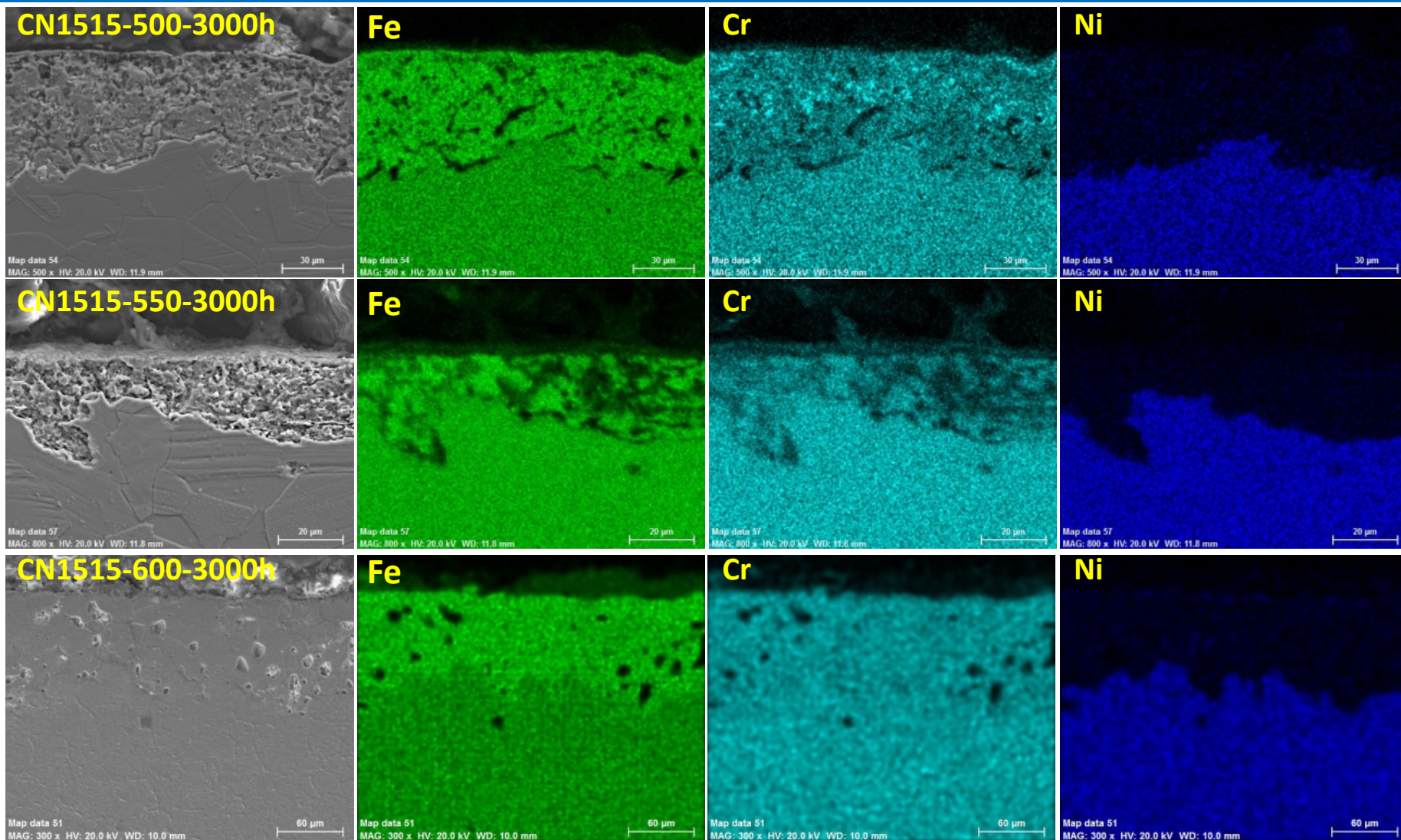


Ni dissolution aeron

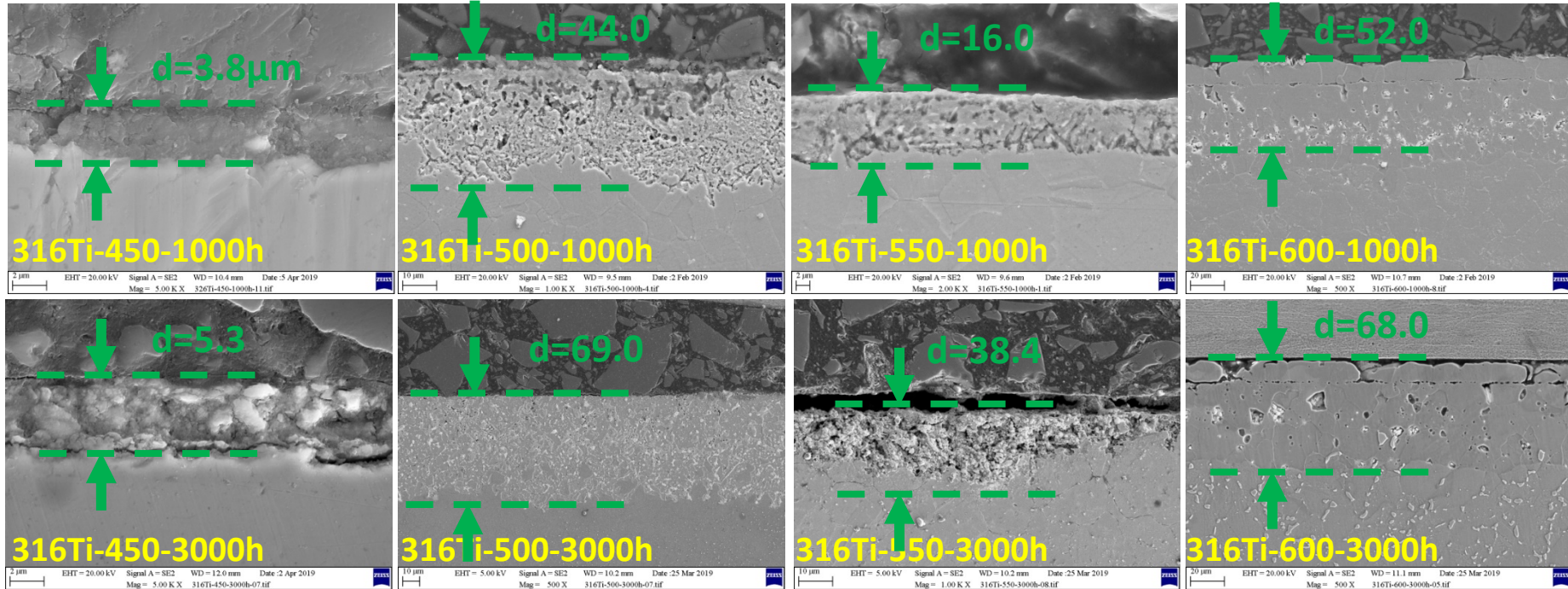


(FeCr) $3O_4$

CN-1515: Corrosion section EDS mapping analysis

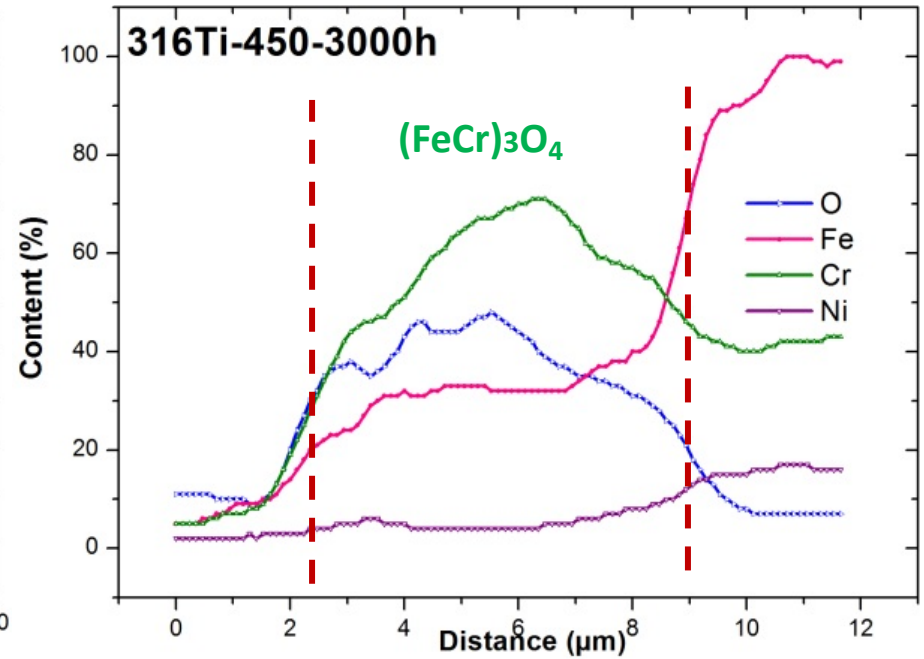
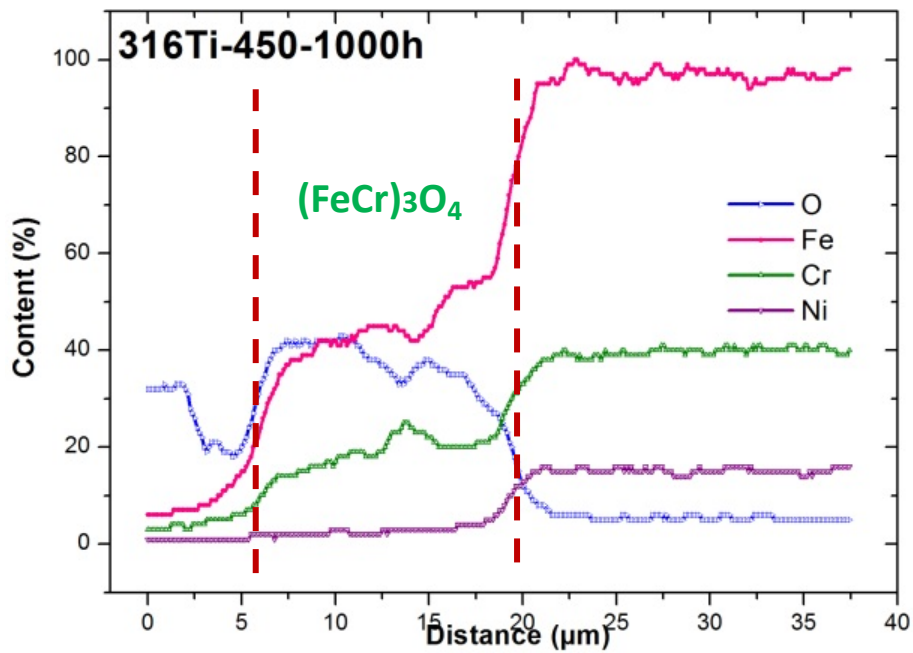


SEM analysis of corrosion section-316Ti



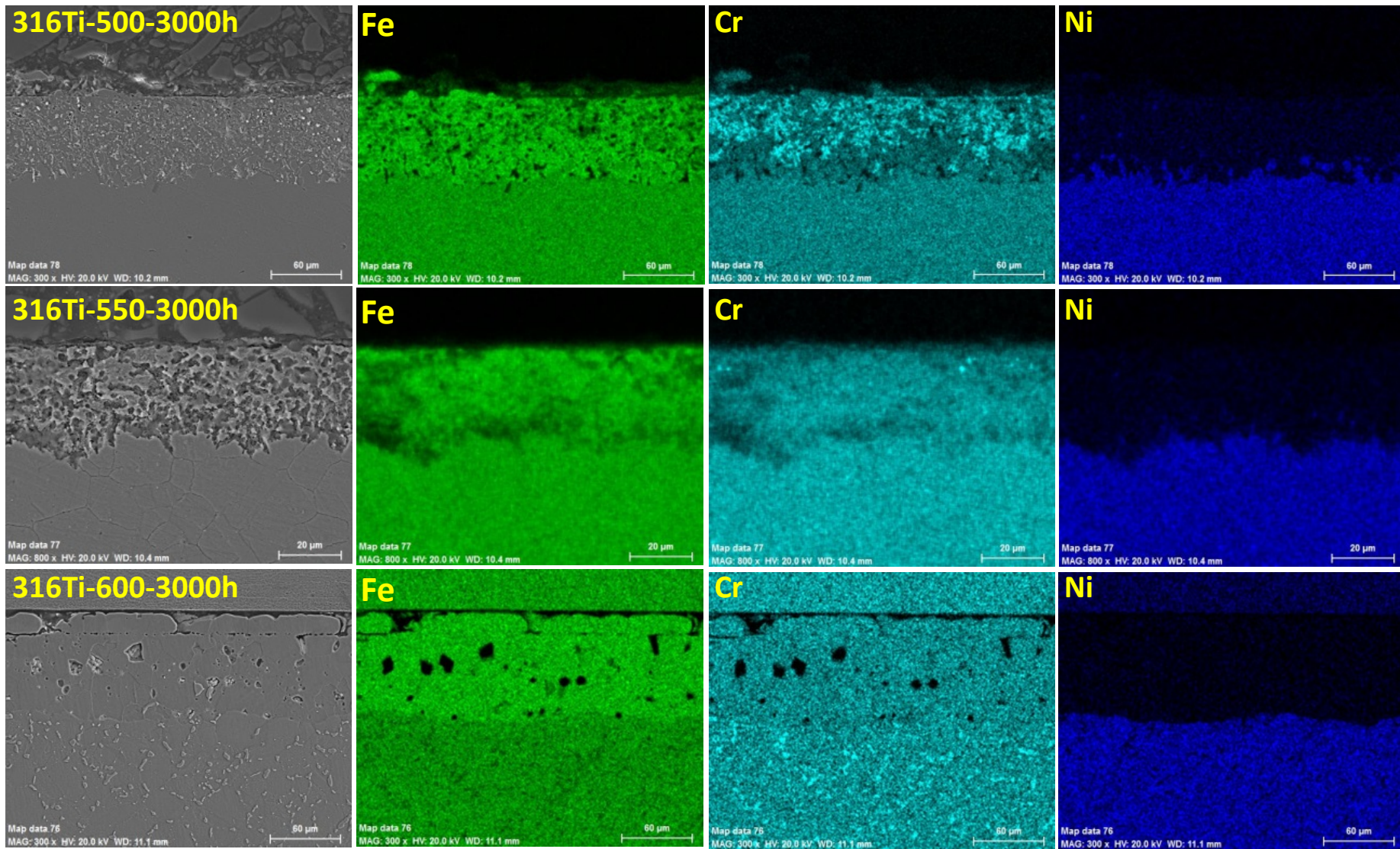
- At 450°C, oxidation corrosion takes place, which is in agreement with the XRD results ;
- At 500, 550 and 600°C, an obvious dissolution corrosion is observed ;

316Ti: Corrosion product composition analysis

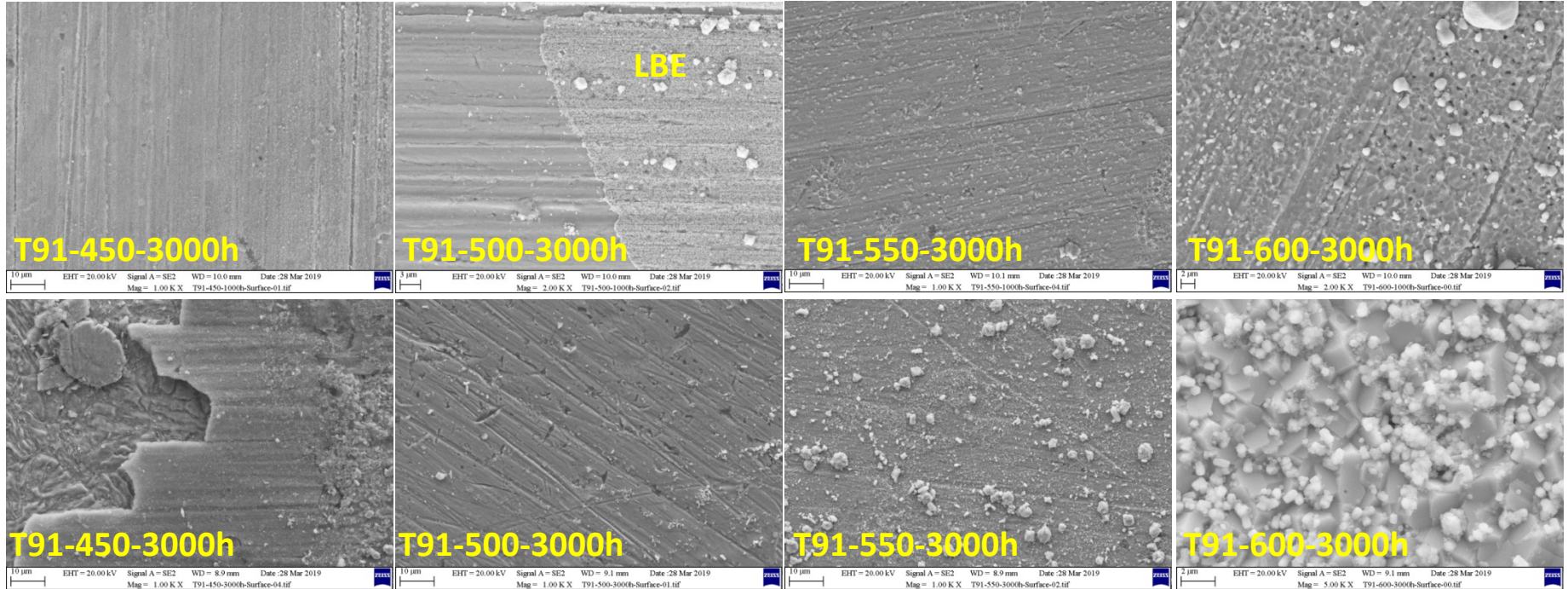


- EDS analysis shows the observed oxidation film is single layer

316Ti: Corrosion section EDS mapping analysis

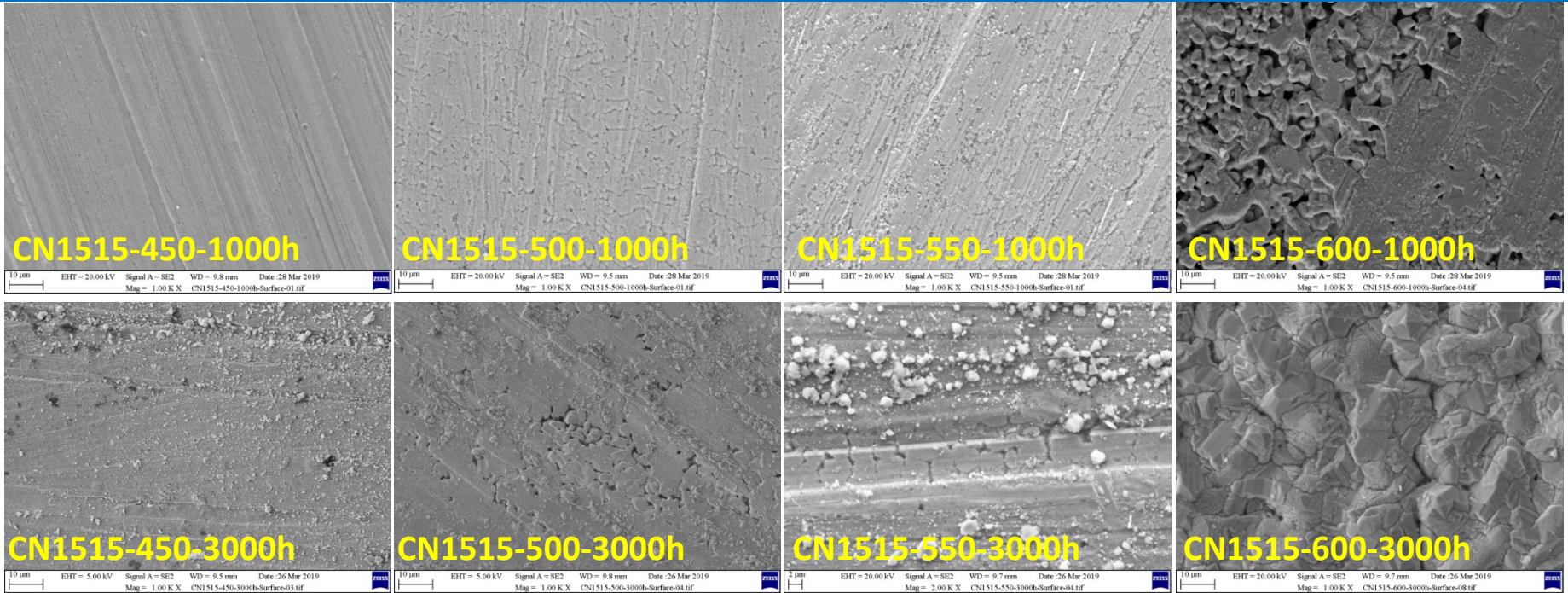


T91-Surface Observation



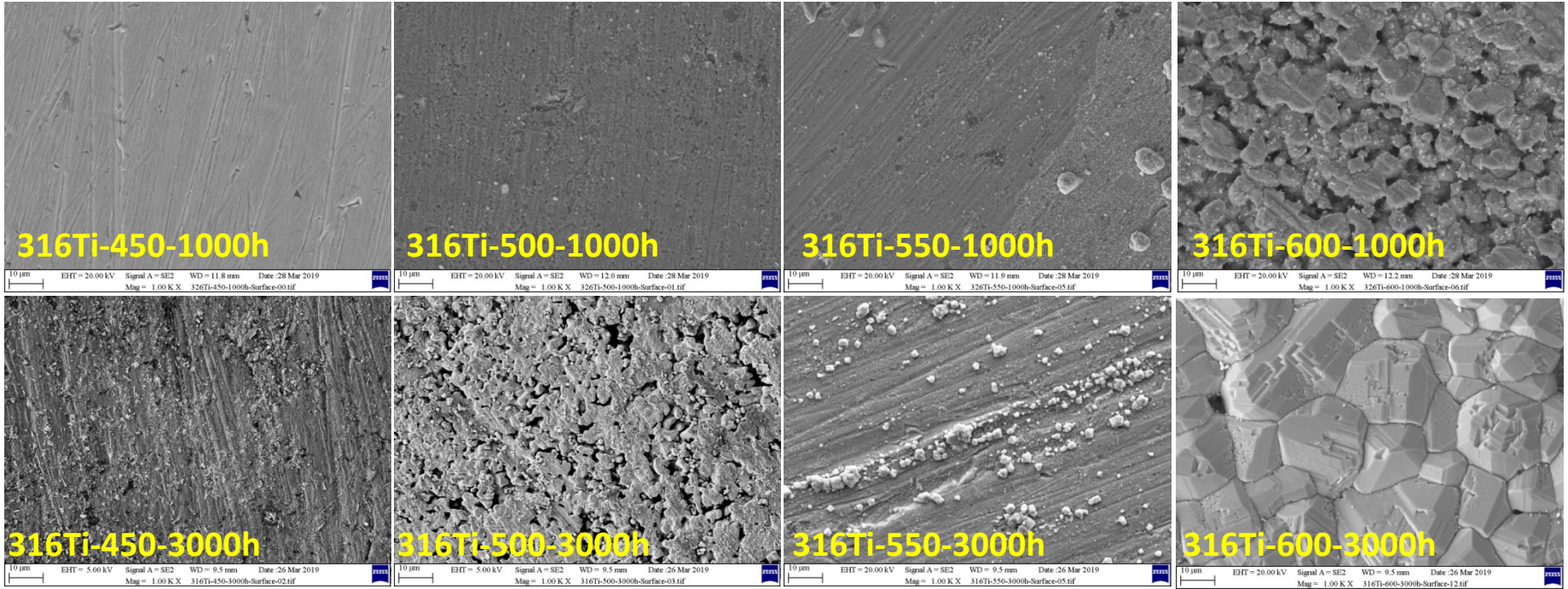
- After 1000h, the original texture of sample on surface can be observed, and at 600°C , an evident corrosion is clearly visible;
- After 3000h, at 450°C, part of the oxide layer is broken and peeled off; at 500 and 550°C,
- But at 600°C, SEM shows a recrystallized morphology.

CN-1515 -Surface Observation



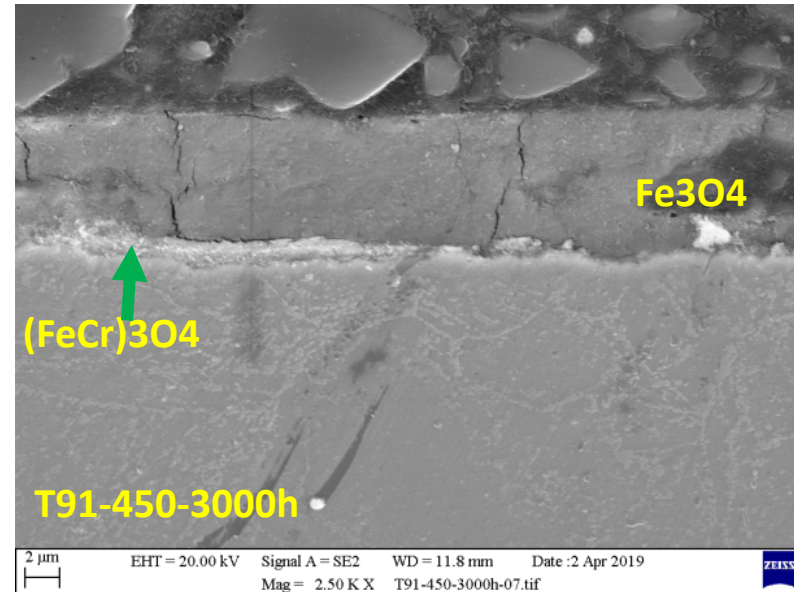
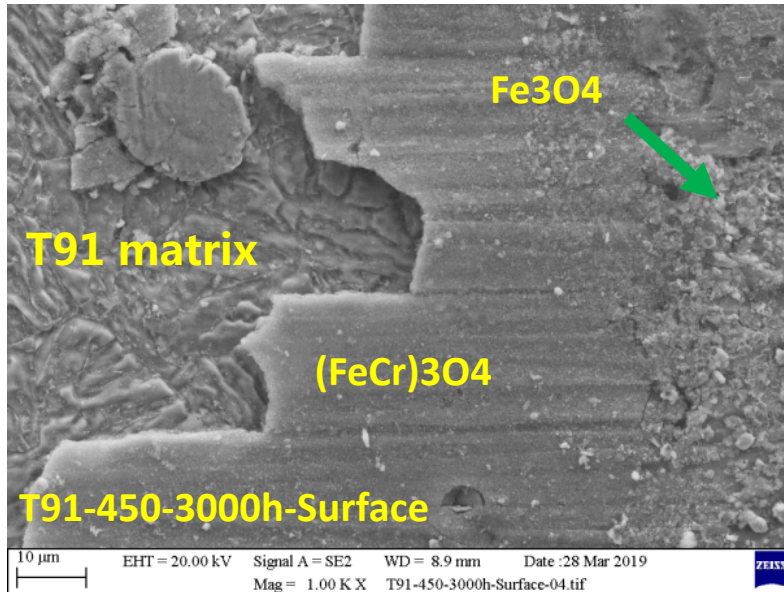
- At 450°C, the surface oxide layer is dense enough to protect matrix ;
- With the increase of temperature, corrosion cracks appear on the surface;
- After 1000h at 600°C, A severe dissolution corrosion occurs, and after 3000h, a recrystallized layer formed on the surface of CN-1515, same as T91.

316Ti-Surface Observation



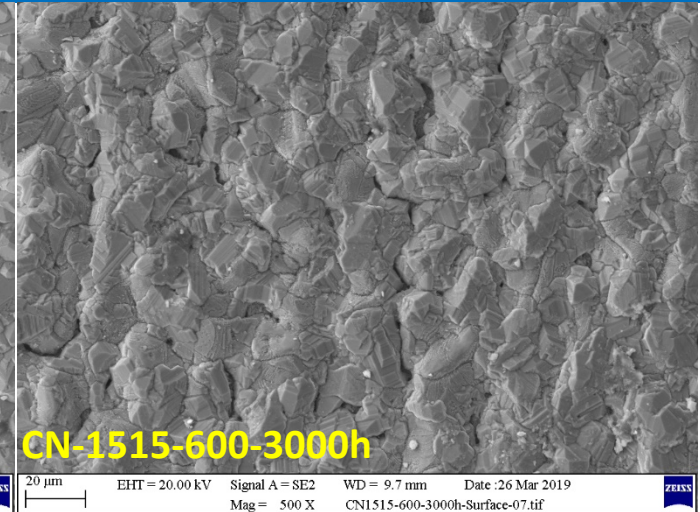
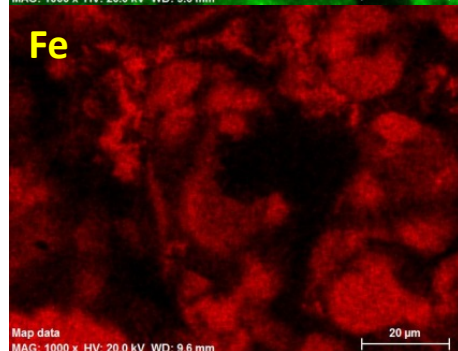
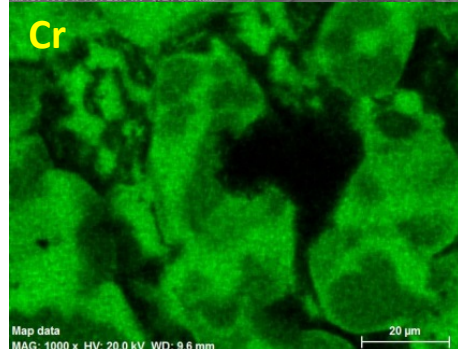
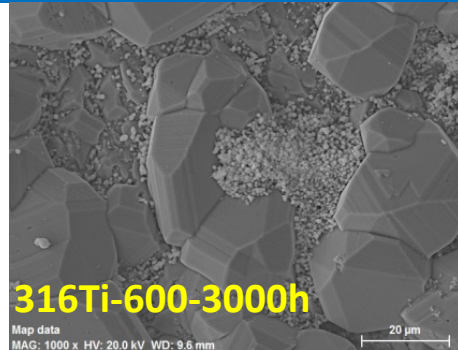
- As same as CN-1515, the 316Ti show a similar corrosion tendency in LBE at different temperatures;

Corrosion time effect



- With increment of corrosion time, the oxide layer is not stable.
- Cracks takes place in the out layers;
- The inner layer is also easy to be broken.

Severe mass transfer effect at high temperature



- At high temperature, LBE has a high solubility of Fe and Cr, and when reduce the experimental temperature, the over-saturated Fe and Cr precipitates on the surface of samples.

Summary

- At 450C°, an oxide film is formed on the surface of T91, CN-1515 and 316Ti, and the layer is not stable when it grows;
- At 500 and 550C°, a slight corrosion occurs on T91ferritic – martensitic steel ;
- For the austenitic steel (CN1515 and 316Ti), the Ni diffuse from corrosion layer to LBE, meanwhile, the Pb and Bi penetrate into materials matrix along the grain boundary;
- Because of Ni diffusion and dissolution, the original FCC structure is not stable, a FCC-BCC phase transformation occurs;
- At 600 C°, dissolution corrosion happened for all three materials;
- A severe mass transfer effect exists at high temperature as result of high solubility of Fe and Cr.

**Thanks for your
attention!**